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Multi-walled carbon nanotube oxidation dependent keratinocyte cytotoxicity and skin inflammation

Brian C. Palmer¹, Sarah J. Phelan-Dickenson¹ and Lisa A. DeLouise^{1,2,3,4*} 

Abstract

Background: The effects of carbon nanotubes on skin toxicity have not been extensively studied; however, our lab has previously shown that a carboxylated multi-walled carbon nanotube (MWCNT) exacerbates the 2, 4-dinitrofluorobenzene induced contact hypersensitivity response in mice. Here we examine the role of carboxylation in MWCNT skin toxicity.

Results: MWCNTs were analyzed by transmission electron microscopy, zetasizer, and x-ray photoelectron spectroscopy to fully characterize the physical properties. Two MWCNTs with different levels of surface carboxylation were chosen for further testing. The MWCNTs with a high level of carboxylation displayed increased cytotoxicity in a HaCaT keratinocyte cell line, compared to the MWCNTs with intermediate levels of carboxylation. However, neither functionalized MWCNT increased the level of in vitro reactive oxygen species suggesting an alternative mechanism of cytotoxicity. Each MWCNT was tested in the contact hypersensitivity model, and only the MWCNTs with greater than 20% surface carboxylation exacerbated the ear swelling responses. Analysis of the skin after MWCNT exposure reveals that the same MWCNTs with a high level of carboxylation increase epidermal thickness, mast cell and basophil degranulation, and lead to increases in polymorphonuclear cell recruitment when co-administered with 2, 4-dinitrofluorobenzene.

Conclusions: The data presented here suggest that acute, topical application of low doses of MWCNTs can induce keratinocyte cytotoxicity and exacerbation of allergic skin conditions in a carboxylation dependent manner.

Keywords: Nanoparticle, Carbon nanotube, Skin, Allergy, Dermatitis

Background

Carbon nanotubes (CNT) are layers of sp^2 hybridized carbon atoms formed into single-walled (SWCNT) or multi-walled (MWCNT) cylindrical structures that were first reported by Sumio Iijima [1, 2]. These high aspect ratio nanoparticles have a pore diameter < 100 nm and a length usually on the micron scale. Their composition imparts carbon nanotubes with unique physical properties, including both high tensile strength and electrical conductivity. The material science field has already exploited these properties to enhance polymers used in

vehicles and sports equipment; however, the expected levels of CNTs released from these products remains low [3]. Of greater concern are both the current occupational exposures from the manufacture of CNTs, and the potential exposures from future biomedical technologies that utilize CNTs.

Early assessments of CNT manufacturing and handling practices revealed that airborne CNT concentrations could be as high as $53 \mu\text{g}/\text{m}^3$ [4]. Research also suggested that in vivo respiratory exposures to CNTs led to acute lung inflammation and fibrosis [5–8]. These revelations prompted the National Institute for Occupational Safety and Health to recommend an 8 h time-weighted average of elemental carbon respirable mass exposure limit of $1 \mu\text{g}/\text{m}^3$ [9]. More recent surveys of CNT manufacturers reveal that most sites are below the new airborne exposure limit [10, 11]; however, less is known

* Correspondence: Lisa_DeLouise@urmc.rochester.edu

¹Department of Environmental Medicine, University of Rochester Medical Center, New York, USA

²Department of Biomedical Engineering, University of Rochester, Rochester, NY, USA

Full list of author information is available at the end of the article



about dermal exposures to CNTs. Based on the average surface area of an adult hand (414.4 cm^2) and the estimated deposition of CNTs during manufacturing processes of 0.2–6 mg per hand, we estimate potential dermal exposures to be anywhere from 0.5–14.5 $\mu\text{g}/\text{cm}^2$ [4, 12]. Due to the use of personal protective equipment, it is likely that only a small fraction of the deposited CNTs contact skin; therefore, examination of low dose dermal exposure of CNTs is more relevant. Even with increased use of personal protective equipment, there is still evidence for dermal exposure of CNTs to the hands and wrists of workers [11]. Aside from occupational exposures, new biomedical technologies are being developed that could increase skin exposure to CNTs. Specifically, CNTs are being developed as transdermal drug delivery devices [13–15], bio-sensing patches [16–19], and “smart” textiles [20–22]. While these technologies are in the early stages of development, additional dermal toxicological studies of CNTs are warranted.

There are a number of studies exploring the *in vitro* toxicity of CNTs on skin cells, including keratinocytes and fibroblasts. Investigations into pristine SWCNT toxicity suggested that these nanoparticles induce oxidative stress and decrease the viability of an immortalized keratinocyte cell line (HaCaT) [23, 24]. A separate study found that pristine MWCNT treatment decreased cell viability and altered protein expression in human epidermal keratinocytes (HEK) [25, 26]. Pristine, unmodified CNTs are hydrophobic; therefore, they are often functionalized to enhance stability in biologic media. An examination of carboxylated MWCNTs found that they were genotoxic and decreased cell viability in normal human dermal fibroblasts [27]. A more recent study also found carboxylated MWCNTs induce DNA damage in HaCaT cells [28]. While *in vitro* toxicology screening suggests a potential for CNTs to induce dose dependent cell death, *in vivo* studies are required to identify whether this translates to skin irritation.

Comparatively, there are few *in vivo* dermal toxicology studies on the effects of topically applied CNTs. Two separate studies examined topical applications of high doses ($>2\text{ mg}/\text{cm}^2$) of pristine SWCNT and pristine MWCNT in rabbit and guinea pig models to assess dermal irritation. The findings suggest that only one type of pristine MWCNT led to mild erythema that resolved after 24 h [29, 30]. Another lab examined topical applications of CNTs *in vivo* and found that SWCNTs increased skin thickness, neutrophil activation, and mast cell recruitment in the skin of mice; however, these findings are predominantly due to the high iron content of the unpurified SWCNTs [31]. Due to the increasing prevalence of allergic skin disorders such as atopic dermatitis and allergic contact dermatitis [32, 33], it is important to examine the toxicity of CNTs in both

healthy and disease models. Recently, our lab reported that a relatively low dose of 1 $\mu\text{g}/\text{ear}$ of carboxylated MWCNT could significantly augment the 2,4-dinitrofluorobenzene (DNFB) induced contact hypersensitivity (CHS) response in a hairless C57BL/6 mouse model, and this response was associated with an increase in mast cell degranulation [34].

Here we examine the properties of this specific carboxylated MWCNT that led to the increase in DNFB induced CHS ear swelling. The focus of this study is on how the level of surface carboxylation on the MWCNTs can alter not only the CHS response, but also the level of *in vitro* keratinocyte cytotoxicity. Furthermore, a more comprehensive examination of the skin histology, immune cell infiltration, and cytokine release after topical application of the carboxylated MWCNT and DNFB elucidates possible mechanisms of action. Importantly, we have identified a threshold level of carboxylation that induces proinflammatory effects.

Results

MWCNT physical characterization

The carboxylated MWCNTs previously used by our lab were reported to have a 30 nm outer pore diameter and a 5–20 μm length, suspended in water at a concentration of 1 mg/mL (Lot #1). To accompany Lot #1 in mechanistic studies, a second batch of carboxylated MWCNTs, with the same product code, were purchased from the same vendor (Lot #2). A pristine, unmodified MWCNT with the same dimensions as both carboxylated MWCNT samples will serve as a control. Both carboxylated nanoparticles were relatively well dispersed as evidenced by the low hydrodynamic diameter and polydispersity index values; however, as expected, the pristine MWCNT displayed poor dispersity in water. The dispersity in water is a direct result of the highly negative zeta potential of both carboxylated nanoparticles, which decreases particle agglomeration (Table 1). In addition, all MWCNTs were similar in overall size and shape as indicated by the representative TEM images (Fig. 1a–c) with a minor difference being the appearance of less defects and nanotube fragments in both the pristine MWCNTs and the MWCNTs of Lot #2. The presence of defects and fragments is likely a result of increased oxidation as it is well established that the oxidation process inherently induces fragments and surface defects into the CNTs [35]. To assess the level of oxidation/carboxylation, each nanoparticle was analyzed via x-ray photoelectron spectroscopy (XPS). In this technique, a high energy x-ray beam is used to eject core electrons from the sample surface. The kinetic energy of these electrons is measured to calculate the electron binding energy, which is quantitatively related to the elemental composition of the material. Analysis of the carbon 1s electrons (Fig. 1d–f) showed

Table 1 Carboxylated MWCNT physical characterization

MWCNT	Hydrodynamic Diameter (nm)	Polydispersity Index	Zeta Potential (mV)	C-C (sp^2/sp^3)	C-OH (%)	C=O (%)	HO-C=O (%)
High-COOH (Lot #1)	284.3	0.459	-42.1	0.26	13.88	7.59	19.7
Low-COOH (Lot #2)	178.5	0.268	-54.0	1.47	4.16	8.73	9.42
Pristine	> LOD ^a	0.865	6.79	2.06	2.66	4.66	0

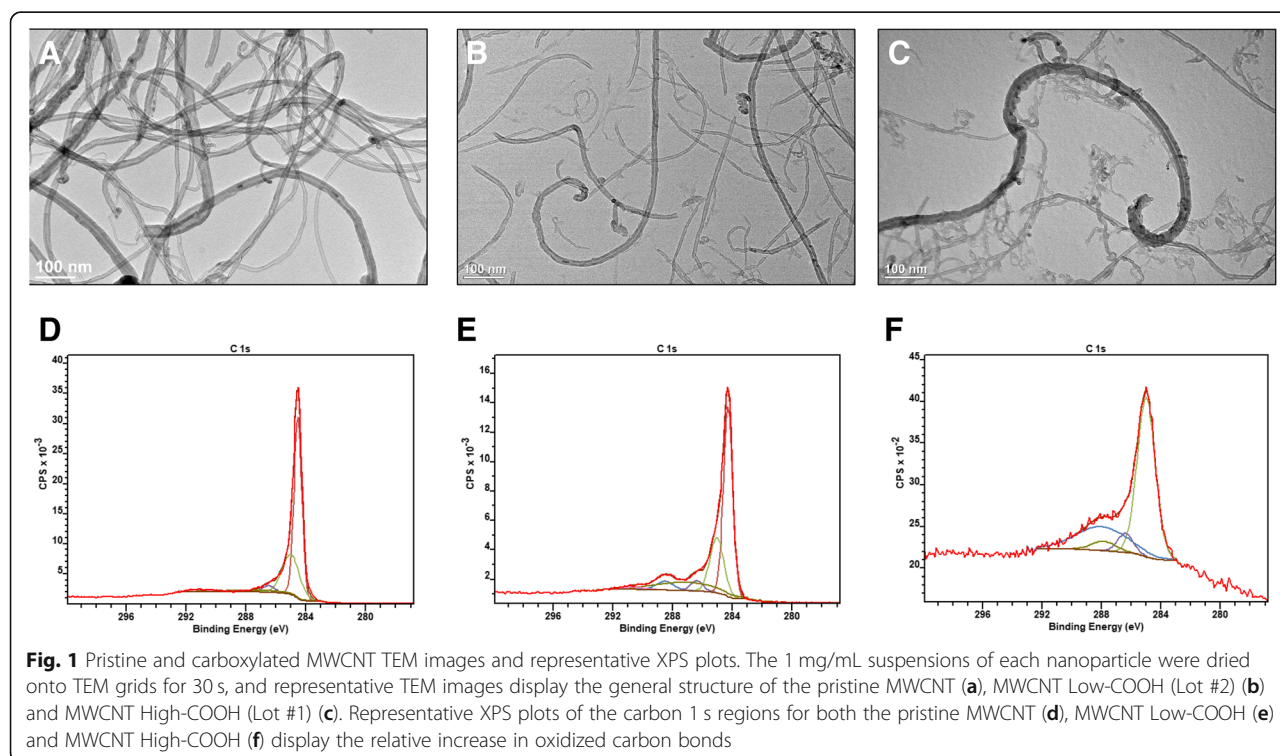
The nanoparticle suspensions of pristine, Lot #1, and Lot #2 MWCNTs were diluted in water and analyzed in a Malvern zetasizer. The hydrodynamic diameter, polydispersity index, and zeta potential are presented in the table. Also, suspensions of each nanoparticle were dried onto silicon wafers and analyzed via XPS, and information on the carbon region are presented in the table. The sp^2/sp^3 carbon bond ratio is the percentage of sp^2 carbon bonds divided by the sp^3 carbon bonds on the surface and indicates the level of surface defects. The C-OH (alcohol), C=O (carbonyl), and HO-C=O (carboxyl) functional groups are all represented as a percent of total surface carbon bonds. ^a The hydrodynamic diameter for the pristine MWCNTs was above the machines limit of detection

unexpected differences between the different MWCNTs. First, the carbon-carbon (sp^2 and sp^3) bonds made up 90.83% of the pristine MWCNT, 77.69% of Lot #2, and 58.83% of Lot #1 MWCNTs. The remaining carbon atoms in each MWCNT lot were bound to oxygen in alcohol, carbonyl, or carboxylic acid functional groups. Predictably, the pristine had only minimal signs of oxidation from the acid washing process; however, the MWCNTs from Lot #1 had a two-fold higher level of carboxylation than Lot #2, which henceforth will be termed MWCNT High-COOH and MWCNT Low-COOH, respectively. Unmodified CNTs are comprised of sp^2 hybridized carbon bonds, and surface defects lead to a higher level of sp^3 hybridized carbon bonds. As expected, the MWCNT High-COOH nanoparticles have a much lower sp^2/sp^3 ratio compared to either the pristine or Low-COOH MWCNTs (Table 1). MWCNTs are often made with transition metal catalysts, which can induce toxicity as well

[31]. These nanoparticles were acid washed and carboxylated which removed any trace of transition metals as detected by the XPS elemental survey (Additional file 1: Table S1).

MWCNT oxidation dependent exacerbation of DNFB induced CHS

Interestingly, we have shown that the MWCNT High-COOH particles induce greater keratinocyte cell death in vitro, when compared to either pristine MWCNT or MWCNT Low-COOH particles (Additional file 1: Figure S1A). However, pristine MWCNT produce far more reactive oxygen species (ROS) than either oxidized MWCNT treatment, in cultured keratinocytes, indicating that the oxidized MWCNT induced cell death is largely ROS independent (Additional file 1: Figure S1B). The in vitro data is suggestive of potential MWCNT induced dermal irritation, and to examine the



effect of these MWCNTs on normal and inflamed skin, our lab uses hairless C57BL/6 mice treated with either vehicle or a DNFB sensitizer. As we have previously reported, topically applied MWCNT High-COOH particles exacerbate the DNFB induced CHS ear swelling response 24 h after co-challenge with 0.2% DNFB and 1 $\mu\text{g}/\text{ear}$ MWCNT High-COOH [34]. This response appears to be synergistic as the MWCNT High-COOH particles have no effect on ear swelling without the sensitizer (Fig. 2). The MWCNT High-COOH particles have a high level of surface defects and particle fragments as indicated by TEM. To assess whether these small fragments are responsible for the increased swelling response, the MWCNT High-COOH particles were filtered through a 0.45 μm filter and the filtrate was used to co-challenge with 0.2% DNFB. The 0.45 μm filtrate was unable to augment the ear swelling response; therefore, the larger MWCNTs are likely the source of increase ear swelling in the CHS model (Fig. 2). Lastly, the MWCNT Low-COOH and pristine MWCNT particles were also tested at a concentration of 1 $\mu\text{g}/\text{ear}$; however, these particles had no effect on ear swelling, when co-treated with either vehicle or 0.2% DNFB (Fig. 2). The data suggests that only MWCNT with a high level of surface carboxylation will exacerbate DNFB induced CHS ear swelling.

Characterization of MWCNT High-COOH Induced Skin Inflammation in a CHS Model

Only the MWCNT High-COOH particles exacerbated DNFB induced CHS ear swelling, and this nanoparticle also reduced the cell viability of cultured keratinocytes more than its pristine or Low-COOH counterpart. To further assess the possible in vivo mechanisms of action of this augmented allergic skin response, a 5 cm^2 area of mouse dorsal skin was challenged with either vehicle, 1 $\mu\text{g}/\text{cm}^2$ of MWCNT High-COOH particles alone, 0.2% DNFB alone, or MWCNT High-COOH at 1 $\mu\text{g}/\text{cm}^2$ plus 0.2% DNFB. After 24 h, the mice were euthanized and the skin histology, immune cell infiltrates, and cytokine expression were examined.

First, histology was examined on H&E stained skin and the average viable epidermal thickness was measured. It is noteworthy that even the MWCNT High-COOH treated skin had a significantly thicker epidermis, compared to control skin. Skin irritants are known to elicit epidermal hyperproliferation and increased skin thickness [36]. DNFB is both a sensitizer and a skin irritant [37], and as expected, DNFB caused the greatest increase in epidermal thickness in both the DNFB group and the co-treatment of both DNFB and the MWCNT High-COOH (Fig. 3e). It is also noteworthy, that both DNFB treated skin sections displayed increased cellular infiltration in the dermis. A

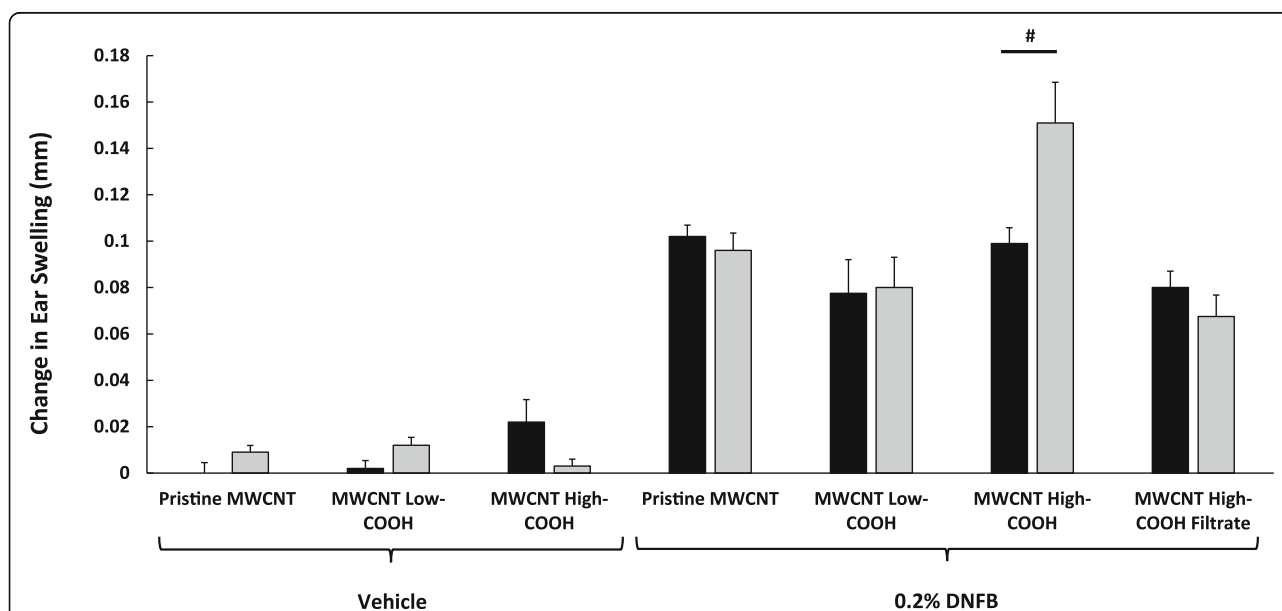
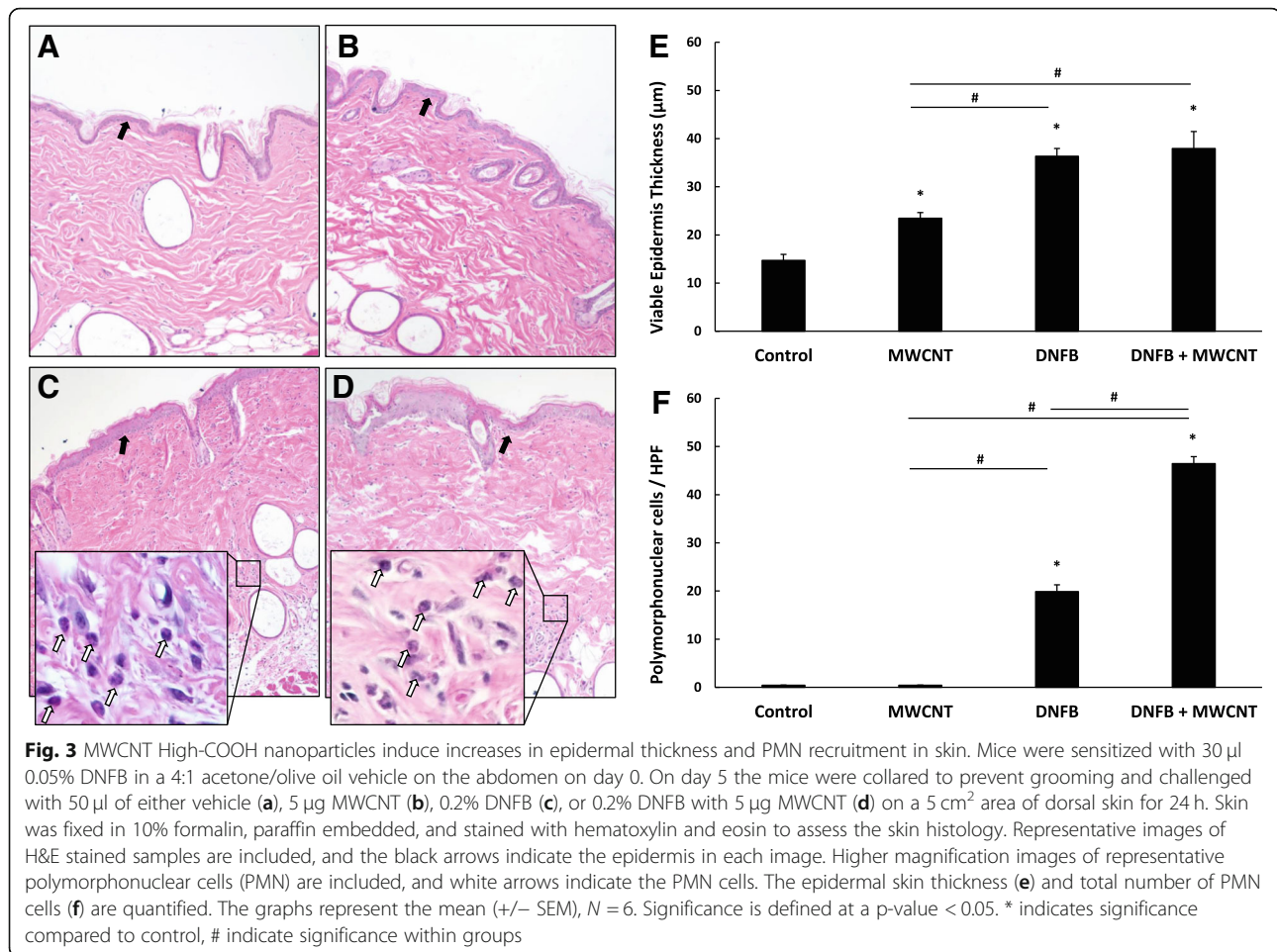


Fig. 2 Effects of MWCNT on the DNFB induced CHS ear swelling responses after 24 h. Mice were sensitized with 30 μl of 0.05% DNFB in a 4:1 acetone/olive oil vehicle on the lower dorsum on day 0. On day 5, the mice were challenged with either vehicle or 0.2% DNFB. The right ears (black bars) are controls that were not treated with MWCNT. The left ears (gray bars) were treated with either pristine MWCNT, MWCNT Low-COOH, MWCNT High-COOH, or MWCNT High-COOH 0.45 μm filtrate. The DNFB increased ear swelling significantly, compared to all vehicle treated mice. Only ears treated with DNFB and MWCNT High-COOH displayed significantly increased swelling responses, compared to DNFB only. Graphs represent changes in mean ear swelling after 24 h (\pm SEM), $N = 5$. A Student's T-test was used to analyze all data presented here, since comparisons were between corresponding right and left ears. Significance is defined at a p -value < 0.05 , and # indicates significance between groups



large portion of these cellular infiltrates were polymorphonuclear cells (PMN), and the DNFB and MWCNT High-COOH co-treatment had significantly more PMNs compared to DNFB only treated skin (Fig. 3f).

We previously reported that the exacerbation of CHS induced ear swelling caused by the MWCNT High-COOH was associated with increased mast cell degranulation [34]. Here we examined the levels of both mast cell and basophil degranulation by counting the number of degranulating and intact cells in toluidine blue stained skin sections. The total number of mast cell or basophils per field was unchanged (Fig. 4e); however, the percentage of degranulating mast cells or basophils was significantly increased by MWCNT High-COOH treatment (Fig. 4f). DNFB also increased the percent of degranulating cells, when compared to control. The MWCNT and DNFB co-treatment significantly increased degranulation when compared to DNFB treatment only. Interestingly, the increased percent of degranulation observed after the MWCNT treatment does not translate to an increase in skin swelling as shown in Fig. 2, which indicates that mast cell or basophil degranulation alone was not sufficient to induce skin swelling in this model.

In the H&E stained skin, immune cell infiltration was observed in the dermis of both the DNFB treated samples (Fig. 3f). To more fully characterize the cellular infiltrates, the skin was analyzed by flow cytometry. The total percentage of antigen presenting cells (APCs) was determined by major histocompatibility complex II (MHCII) staining. There is a significantly higher percentage of MHCII stained cells in both DNFB treated samples when compared to control or MWCNT High-COOH treated skin, but the DNFB and MWCNT High-COOH co-treatment induced the highest percentage of skin APCs (Fig. 5a). The percentage of granulocytes (CD11b+, F4/80-, and high side scatter) in the skin displayed a slight, nonsignificant increase due to DNFB alone; however, significantly more granulocytes were present in skin co-treated with both DNFB and MWCNT High-COOH (Fig. 5b). These granulocytes are most likely neutrophils, since toluidine blue histology suggested that the total number of mast cells and basophils remained unchanged, and DNFB is a T_H1 skewing sensitizer that is unlikely to recruit high levels of eosinophils [38, 39]. While the percentage of macrophages (MHCII+, F4/80+, CD11c-) appeared to increase due to MWCNT High-COOH treatment, the increases are nonsignificant (Fig. 5c). Lastly, the

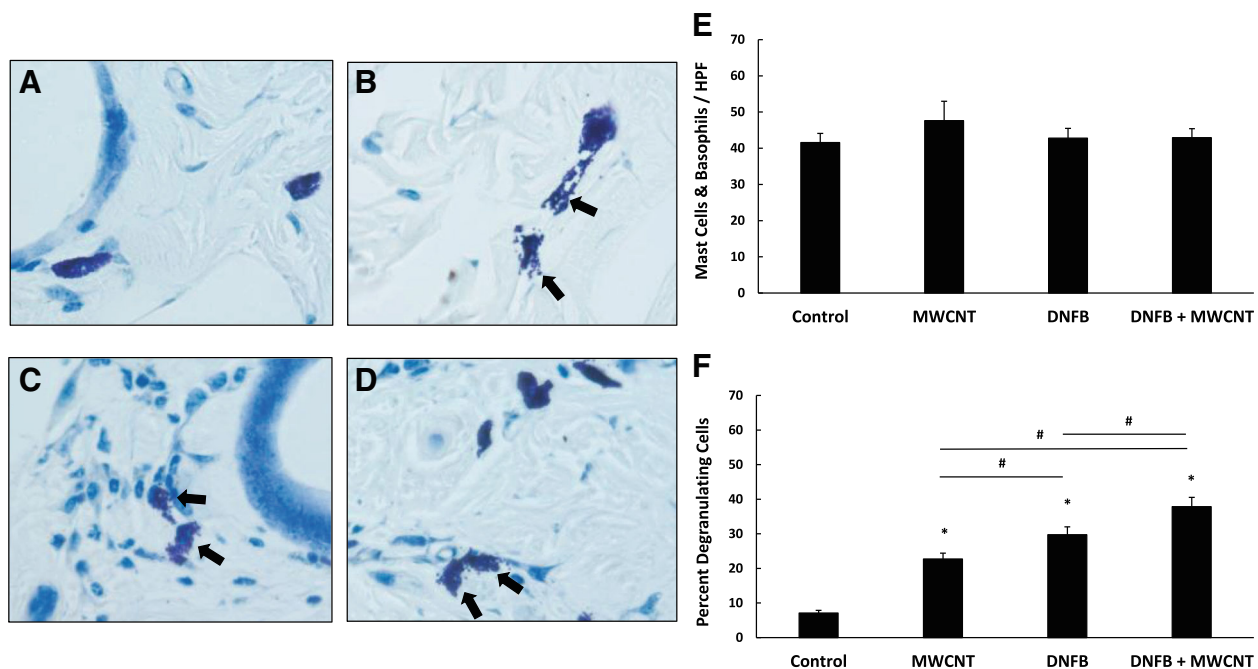


Fig. 4 MWCNT High-COOH nanoparticles induce mast cell/ basophil degranulation in skin. Mice were sensitized with 30 μ l 0.05% DNFB in a 4:1 acetone/olive oil vehicle on the abdomen on day 0. On day 5 the mice were collared to prevent grooming and challenged with 50 μ l of either vehicle (a), 5 μ g MWCNT (b), 0.2% DNFB (c), or 0.2% DNFB with 5 μ g MWCNT (d) on a 5 cm² area of dorsal skin for 24 h. Skin was fixed in 10% formalin, paraffin embedded, and stained with toluidine blue to assess the number of degranulating mast cells/basophils. Representative images of toluidine blue stained samples are included, and black arrows indicate degranulating mast cells. The total number of mast cells/ basophils (e) and the percentage of degranulating mast cells/ basophils (f) are quantified. The graphs represent the mean (+/- SEM), N = 6. Significance is defined at a p-value < 0.05. * indicates significance compared to control, # indicate significance within groups

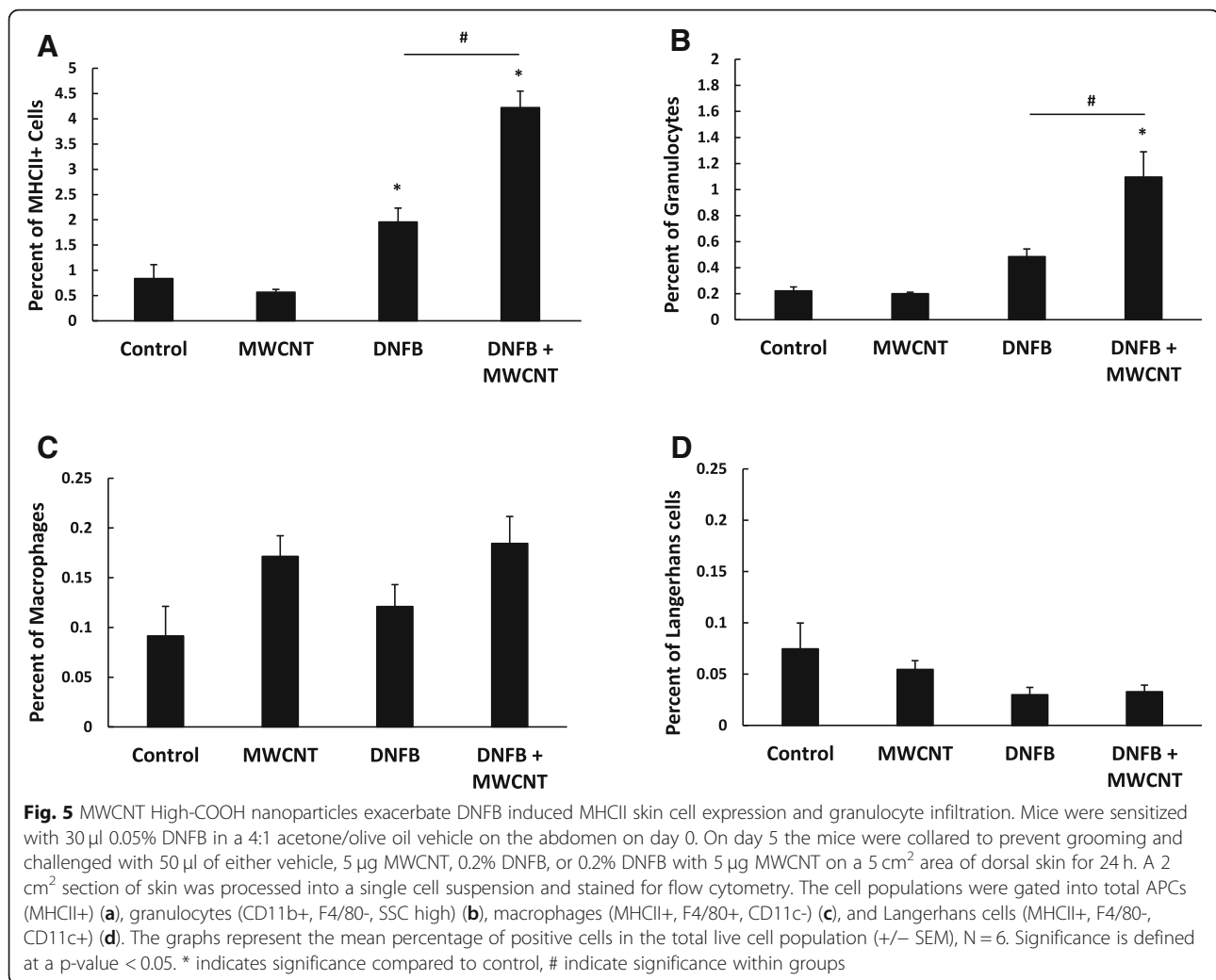
percentage of Langerhans cells (MHCII+, F4/80+, CD11c+) in the skin was relatively low, and was slightly reduced due to DNFB treatment (Fig. 5d) which is consistent with their trafficking out of skin to the local lymph node. It should be noted, that the total percentage of granulocytes, macrophages, or Langerhans cells did not equal the total MHCII positive skin cell percentage. The remaining cells are most likely B cells or keratinocytes, as keratinocytes are reported to express MHCII under inflammatory conditions [40].

To determine signaling events that prompted the difference in immune cell infiltration in the skin, we analyzed cytokine and chemokine protein expression in the skin 24 h after treatment. DNFB is highly inflammatory and led to significant increases in most cytokines analyzed. Only DNFB had significant effects on Interleukin 6 (IL-6), keratinocyte chemoattractant (KC), macrophage inflammatory protein 2 (MIP-2), and tumor necrosis factor alpha (TNF α) (Fig. 6). IL-6 and TNF α are both inflammatory cytokines produced by a number of cells in the skin, including keratinocytes [41]. KC and MIP-2 are both neutrophil chemotactic cytokines produced by macrophages and keratinocytes [42, 43]. DNFB and MWCNT High-COOH co-treatment significantly enhanced the expression of monocyte chemoattractant protein 1 (MCP-1) and interferon gamma inducible protein 10 (IP-10); however, this is

a synergistic effect, since MWCNT High-COOH treatment alone had no effect on these cytokines. MCP-1 is produced by monocytes, macrophages, or dendritic cells and can induce degranulation of mast cells or basophils [44]. IP-10 is produced in response to IFN γ , and is chemotactic for T cells [45]. It is noteworthy that skin treated with MWCNT High-COOH alone trended toward an increase of both macrophage inflammatory protein 1 alpha (MIP-1 α) and interleukin 1 beta (IL-1 β) by 2–6 fold, compared to control. MIP-1 α is predominantly produced by macrophages, and it is involved in the activation of neutrophils [46]. IL-1 β is part of the inflammasome, and it is released from a number of cells, including keratinocytes [41]. Overall, the MWCNT treatment affected cytokines involved in both neutrophil and T cell recruitment.

Determining a Threshold of MWCNT Carboxylation for Induction of Skin Toxicity

To confirm that the observations above correlate with the level of carboxylation and to determine the level of carboxylation that induces skin toxicity, the pristine MWCNTs characterized in Fig. 1 were acid oxidized in-house for either 1.5, 3, 6, or 12 h. The pristine MWCNTs are neutrally charged and highly agglomerated, while the oxidized MWCNTs are all negatively charged and well dispersed in



water (Table 2). All oxidized MWCNTs, except for those oxidized for 1.5 h, have a sp^2/sp^3 ratio below 1, suggesting a high degree of surface defects. All oxidized MWCNTs have detectable carboxylic acid groups, and the 1.5, 3, 6, and 12 h oxidized MWCNTs have 9.66, 12.96, 30.61, and 26.7% carboxylated surface carbons, respectively (Table 2, Additional file 1: Figure S2). It appears as though 25–30% surface carboxylation is the limit achieved by this acid oxidation, and TEM images indicate that by 12 h significant nanotube fragmentation began to occur (Additional file 1: Figure S3).

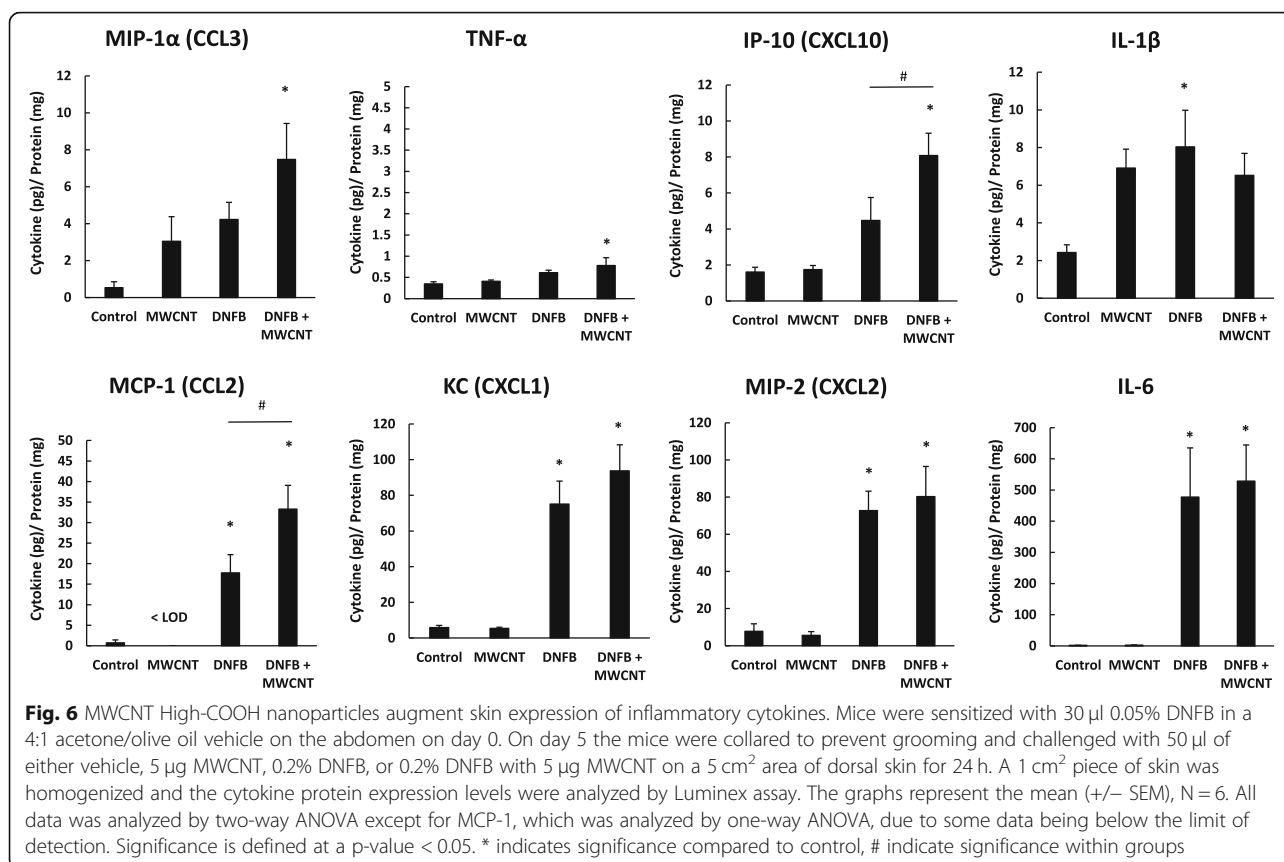
To examine the effect of carboxylation level in vivo we co-treated the ears of sensitized mice with DNFB on the right ear (black bars) and DNFB with 1 μ g/ear of the oxidized MWCNTs on the left ear (gray bars) (Fig. 7). Both MWCNT oxidized for 1.5 and 3 h had no effect on the DNFB induced ear swelling; however, both the 6 and 12 h oxidized MWCNT significantly increased in the ear swelling response, compared to ears treated with only DNFB (Fig. 7). These data confirm that MWCNT with

> 20% surface carboxylation exacerbate DNFB induced CHS responses compared to less oxidized MWCNT.

Discussion

Acute exposures to low doses of highly oxidized MWCNTs lead to epidermal irritation characterized by increased keratinocyte proliferation, cytokine release, and mast cell or basophil degranulation. While the MWCNTs alone are not sufficient to induce skin swelling, they may exacerbate allergen related skin inflammation, primarily through the enhanced recruitment of neutrophils. This trend of carboxylation dependent effects was also observed in cell culture, as HaCaT keratinocytes displayed increased cytotoxicity when exposed to MWCNTs with high levels of carboxylation; however, the increased cytotoxicity was unrelated to ROS generation. To our knowledge, our lab is the first to report this carboxylation dependent MWCNT dermal toxicity, in vivo.

CHS is a biphasic, delayed-type immunity that requires interactions and signaling from a number of different



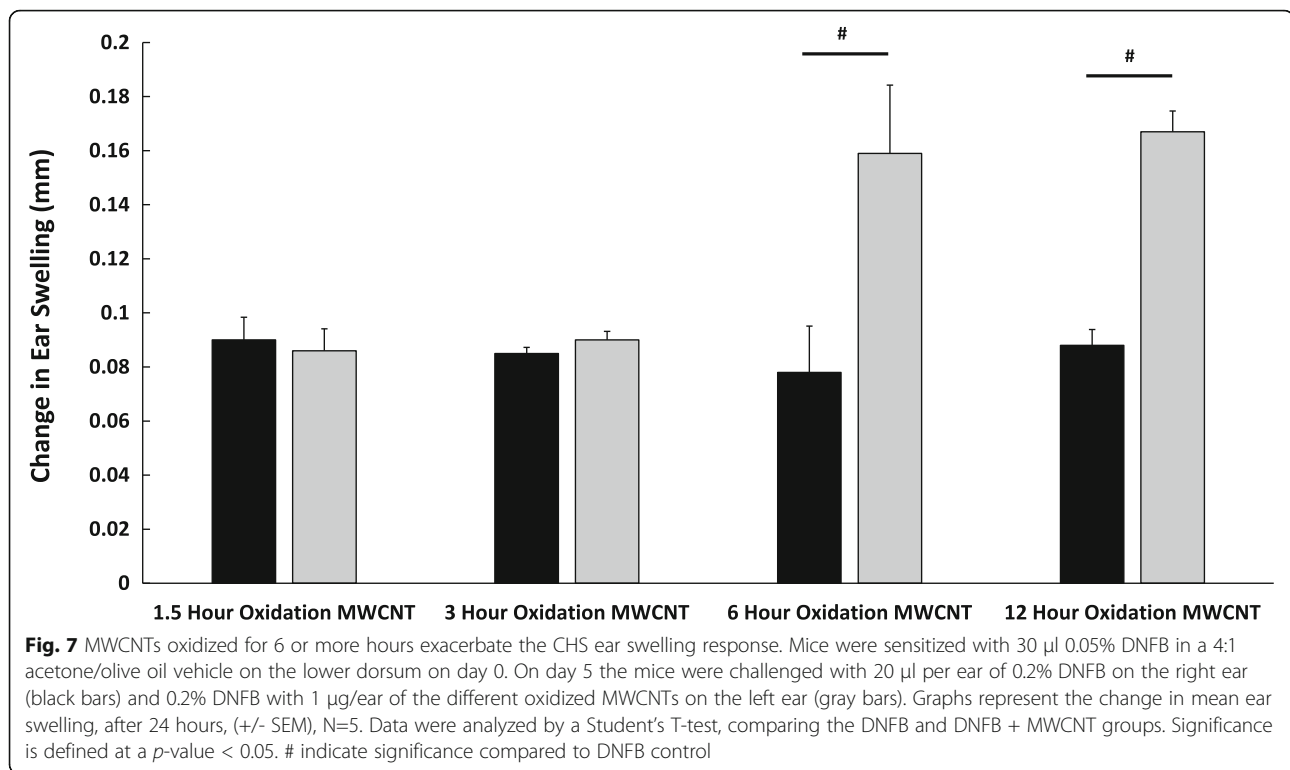
immune cells. First, the sensitization phase produces memory T cells that recognize a specific antigen [47]. In the subsequent challenge phase, the allergen promotes early, innate immune events that depend on keratinocyte inflammatory cytokine release [48], mast cell activation [49], and neutrophil recruitment [50]. Following these early events, antigen specific cytotoxic CD8⁺ and effector CD4⁺ T cells infiltrate skin and lead to skin swelling [51]. Due to the large size of the MWCNTs, it is unlikely the nanoparticles directly interact with cells in the dermis (e.g. mast cells or T cells). Therefore, we focused on the early innate immune responses of CHS, mediated by keratinocytes and APCs at the skin surface.

The mechanism of MWCNT exacerbation of DNFB induced CHS appears to be mediated through epidermal irritation. Both DNFB and MWCNT increased the epidermal thickness in the mouse skin, which is often a response to barrier disruption. The *in vitro* data displaying increased HaCaT cytotoxicity with increasing surface carboxylation level is also suggestive of epidermal irritation. It is plausible that either the carboxylic acid groups have a special reactivity with epidermal keratinocytes, or the oxidation enhances the dispersity of the tubes to allow for enhanced interaction with viable epidermal cells. It is noteworthy that the MWCNT oxidized for 3 h display more total oxidation than the MWCNT oxidized for 6 or 12 h, but this oxidation is limited to mostly

Table 2 In-house Oxidized MWCNT physical characterization

MWCNT	Hydrodynamic Diameter (nm)	Polydispersity Index	Zeta Potential (mV)	C-C (sp ² /sp ³)	C-OH (%)	C=O (%)	HO-C=O (%)
1.5 Hour Oxidation	373.0	0.461	-66.3	1.54	28.78	3.79	9.66
3 Hour Oxidation	332.3	0.459	-68.8	0.79	26.91	3.95	12.96
6 Hour Oxidation	208.8	0.363	-48.7	0.57	6.09	0.2	30.61
12 Hour Oxidation	178.9	0.354	-61.9	0.72	5.33	0.01	26.7

The nanoparticle suspensions of 1.5 h oxidized, 3 h oxidized, 6 h oxidized, and 12 h oxidized MWCNTs were diluted in water and analyzed in a Malvern zetasizer. The hydrodynamic diameter, polydispersity index, and zeta potential are presented in the table. Also, suspensions of each nanoparticle were dried onto silicon wafers and analyzed via XPS, and information on the carbon region are presented in the table. The sp²/sp³ carbon bond ratio is the percentage of sp² carbon bonds divided by the sp³ carbon bonds on the surface and indicates the level of surface defects. The C-OH (alcohol), C=O (carbonyl), and HO-C=O (carboxyl) functional groups are all represented as a percent of total surface carbon bonds



alcohol functional groups. Compared to the other oxidized MWCNTs, the MWCNT oxidized for 3 h have approximately half the amount of carboxylic acid surface functionalization, and these nanoparticles do not affect the *in vivo* CHS responses. In the future, the effect of surface functionalization density on MWCNT toxicity should be examined for other functional groups as well.

This study also identified mast cell or basophil degranulation as a key event in the MWCNT exacerbation of DNFB induced CHS. The MWCNT treatment increased degranulation in both the control and DNFB treated mice. While we were unable to identify a direct signal that led to mast cell and basophil degranulation, it is likely that the inflammatory skin environment produced by MWCNT induced epidermal irritation is the primary cause. MCP-1 is known to induce mast cell degranulation in some tissues [44], and MCP-1 was elevated by MWCNT in the DNFB treated mice. However, MCP-1 was not detected in animals exposed to only MWCNT; therefore, MCP-1 expression alone does not explain the MWCNT induced mast cell and basophil degranulation. MIP-1 α expression was upregulated in skin due to MWCNT exposure, and this cytokine has been described as a co-stimulatory signal for mast cell degranulation in the conjunctiva, meaning it is required for degranulation, but not sufficient to induce degranulation alone [52]. It is possible that the primary degranulation signal is complement activation and signaling through

IgM [53]; however, this mechanism would require specific antigen recognition. Alternatively, skin irritation, which results in keratinocyte necrosis, could release damage associated molecular patterns. Some of these proteins have been identified to induce mast cell degranulation [54]. Mast cell degranulation and activation leads to the release of inflammatory cytokines, histamine, prostaglandins, and leukotrienes. These mediators cause vasodilation and enhance immune cell extravasation through endothelial cells [55], which has implications for immune cell infiltration in CHS responses.

There is evidence that suggests mast cell activation is a prerequisite for neutrophil infiltration into tissues [56, 57]. In this study, we identified an increase in granulocyte skin infiltration after DNFB exposure, and MWCNT exacerbated this DNFB induced effect. However, mast cell degranulation is not sufficient to induced neutrophil recruitment alone, since granulocyte infiltration was unchanged by MWCNT treatment in control skin. DNFB exposure led to increased KC and MIP-2 cytokine expression, both of which are potent neutrophil chemotactic cytokines necessary for CHS [42, 43]. More recently, MIP-1 α and MCP-1 have been identified as factors that recruit and activate neutrophils by enhancing extravasation and increasing the expression of lipid mediators [46]. These two cytokines were upregulated by MWCNT skin exposure, and along with mast cell degranulation could explain the increased granulocyte

infiltration in skin. Neutrophils enhance skin inflammation through cytokine and protease release; however [58, 59], they also have a role in T cell recruitment to enhance CHS responses. Neutrophils, along with other APCs, are known to express IP-10, a chemotactic cytokine for the recruitment of T cells [45], and IP-10 expression follows a similar pattern as neutrophil infiltration in our model. Studies suggest that level of neutrophil infiltration is correlated with the level of cytotoxic T cell infiltration in the later phases of CHS [60]. We conclude that both direct effects of enhanced neutrophil infiltration, and indirect effects of subsequent cytotoxic T cell recruitment play a role in the MWCNT exacerbation of DNFB induced CHS skin swelling.

The use of mouse models for dermal nanotoxicology are useful since they offer insight into immune cell activation that keratinocyte cell culture alone cannot provide. However, mouse skin differs from human skin in both thickness and hair follicle density [61]. The thickness of the epidermis in humans can be anywhere from 2-15x thicker than the mouse, depending on the area. Therefore, mouse skin exposures to nanoparticles are more representative of worst-case exposures to barrier disrupted human skin. Highly oxidized MWCNT should be examined in reconstructed human epidermal cultures to identify if they induce the same level of skin irritation. However, the data presented here may be relevant for a large population of individuals with skin inflammatory conditions [32, 33]. People with inflammatory skin conditions have barrier disrupted skin that makes nanoparticle exposure to the viable epidermis more likely [62].

The finding that highly carboxylated MWCNT exacerbate allergic skin responses is in line with data published suggesting that MWCNT exposure to the lung exacerbates allergic airway inflammation. Early reports indicate that inhaled or instilled MWCNT exacerbate ovalbumin induced airway inflammation [63, 64]. More recently, the effects of MWCNT on allergic airway inflammation seem to be caused by inflammasome activation and innate cytokine production, like IL-33 [65–67]. However, these reports on MWCNT exacerbating allergic airway inflammation examined pristine, unmodified MWCNT. Studies on the lung inflammation of oxidized MWCNT suggest that functionalization decreases bioactivity, compared to unmodified MWCNT [68, 69]. While the level of surface carboxylation was not mentioned in either study, it is likely that oxidized MWCNT toxicity is both exposure route and tissue specific.

While the *in vitro* cytotoxicity data and *in vivo* CHS ear swelling data presented here are well correlated, the *in vitro* ROS data displays an opposite trend. The ROS data would suggest that the pristine tubes would exert a high level of ROS dependent toxicity in skin; however these nanoparticles had no effect on the CHS response.

It is important to remember the differences in dosimetry between *in vitro* and *in vivo* nanoparticle exposures that can make interpretation of *in vitro* data difficult. This could explain why the *in vitro* data on the toxicity of functionalized MWCNT in the literature displays less consensus. Early studies suggested that increasing the surface functionalization density of SWCNT decreased the cytotoxicity in an *in vitro* human dermal fibroblast model [70]. Examination of carboxylated MWCNT in a THP-1 monocyte cell line found that the functionalization decreased inflammasome activation [71]. More recent examination of carboxylated MWCNT in a keratinocyte cell line found that the MWCNT induced genotoxic DNA damage, regardless of functionalization [28]. Lastly, there are reports of MWCNT induced macrophage cell line cytotoxicity increasing with higher levels of oxidation [72]. The lack of consensus in these reports is likely due to differences in particle characteristics, doses, and cell lines. Altogether, this is evidence of the need for more consistent *in vitro* toxicity models and increased nanoparticle physical characterization.

Conclusions

This study has identified a MWCNT carboxylation dependent keratinocyte toxicity that is not mediated by ROS generation. Low, physiologically relevant exposures to the highly oxidized MWCNT also lead to epidermal irritation, cytokine release, and mast cell degranulation in a mouse model. These MWCNT skin effects exacerbate DNFB induced CHS reactions by increasing neutrophil skin recruitment. Altogether, this data suggests that more dermal toxicity studies of oxidized MWCNT exposures should be conducted before widespread commercial use. Alternatively, more biocompatible alternative functional groups should be explored, to reduce toxicity.

Methods

Carbon Nanotube Oxidation and Characterization

The MWCNT (High COOH) and MWCNT (Low COOH) were two different nanoparticle lots purchased from Nanolab (Cat# PD30L5–20COOH) and they were suspended in water at a 1 mg/mL concentration. The MWCNT (Pristine) particles were also purchased from Nanolab (Cat# PD30L5–20), but these were in powder form. All of the carbon nanotubes used in this study have a pore diameter of 30 nm and a length ranging from 5 to 20 μm .

To oxidize the carbon nanotubes, 100 mg of dry MWCNT (Pristine) were placed into a round bottom flask containing 100 mL of a 4:1 mixture of concentrated sulfuric acid and 70% nitric acid. The mixture was refluxed, while stirring at 80° C for 3, 6, or 12 h. To stop the oxidation process, the acid mixture was diluted with water, and the carbon nanotubes were centrifuged at

3000 × g for 20 min. The carbon nanotubes were washed with water two additional times. Finally, the tubes were suspended in 50 mL of water, sonicated for 15 min with a 300 W probe sonicator at a 20% duty cycle with 40% power, and neutralized with dilute sodium hydroxide. For the 0 h sample, 100 µg of dry MWCNT (Pristine) was simply placed in 50 mL of water and sonicated. Since the oxidation process results in the loss of some carbon material, the sample concentrations were confirmed by drying a known volume of the samples in a 50° C oven overnight in pre-weighed crucibles.

To assess the relative size, dispersity, and charge of the MWCNT; the nanoparticles were diluted 200x in water and analyzed in the Malvern Zetasizer Nano ZS instrument to determine the hydrodynamic diameter and surface charge. The shape and structure of the MWCNT was analyzed via transmission electron microscopy (TEM). Lastly, the chemical composition of the MWCNT surface was assessed with x-ray photoelectron spectroscopy (XPS, Kratos Axis Ultra DLD). For XPS analysis, the samples were dried onto a 1 cm × 1 cm silicon wafer (p-type, 0.001 ohm-cm) in a 50° C oven overnight. The complete elemental survey was completed with a 160 eV pass energy and represents the average of 5 sweeps, and the peaks were assigned by an automated elemental electron binding energy library on CasaXPS version 2.3.18 (Casa Software, Ltd., Teignmouth, UK). The dried MWCNT film was sufficiently thick that silicon was not detected in the survey scan. The carbon 1 s region was analyzed with a 20 eV pass energy and represents the average of 5 sweeps. The relative percentages of the different carbon bonds were estimated by an automated Gaussian/Lorentzian curve fitting after setting the following carbon electron binding energy peaks based on literature values: C-C sp² bond (284.3–284.6 eV), C-C sp³ bond (285–286 eV), C-OH alcohol (286.4–286.7 eV), C=O carbonyl (287.1–288.1), HO-C=O carboxylic acid (288–289.4 eV), or pi-stacking electrons (290.5–291.5 eV) [73].

Cell culture

The HaCaT cells used in this study are a spontaneously transformed, immortal keratinocyte cell line [74]. These adherent cells were grown in Dulbecco's Modified Eagle Medium (DMEM) (Gibco Cat# 11965–092) supplemented with 10% fetal bovine serum (Gibco Cat# 10082–147) and 1% penicillin/streptomycin (Gibco Cat# 15140–122). The cells were incubated in 5% carbon dioxide at 37° C. Before nanoparticle treatment, the cells were trypsinized with 0.25% Trypsin-EDTA (Gibco Cat# 25200–056) and seeded into two 96 well plates at a density of 35,000 cells per well. The HaCaT cells were grown for 24 h, or until they were 70% confluent.

For the reactive oxygen species (ROS) assay, the cells were cultured in a black walled 96 well plate, washed with phosphate buffered saline (PBS), and incubated with 100 µL of 10 µM DCF-DA dye (Invitrogen Cat. D399) dissolved in PBS for 30 min prior to MWCNT treatment. After loading the cells with DCF-DA, the PBS was removed, and the wells were treated with 100 µL of culture media containing 0–25 µg/mL of the MWCNT (High COOH), MWCNT (Low COOH), MWCNT (Pristine), MWCNT (3 Hour Oxidation), MWCNT (6 Hour Oxidation), or MWCNT (12 Hour Oxidation). For a positive control, 100 µL of culture media containing 100 µM of hydrogen peroxide was added to induce ROS in the HaCaTs. To measure the DCF fluorescence after 24 h, the 96 well plate fluorescence was measured in a multimode plate reader (Perkin Elmer) at an excitation wavelength of 490 nm and an emission wavelength of 525 nm. The data were then normalized to the control.

For the cell viability assay, the cells cultured in the white walled 96 well plate were directly treated with 100 µL of culture media containing 0–25 µg/mL of the MWCNT (High COOH), MWCNT (Low COOH), MWCNT (Pristine), MWCNT (3 Hour Oxidation), MWCNT (6 Hour Oxidation), or MWCNT (12 Hour Oxidation). After 24 h, 100 µL of CellTiter-Glo (Promega Cat# G7571) reagent was added to each well, without washing. After 15 min, the luminescence of each well was measured with a Turner Biosystems Modulus microplate reader. The data were then normalized to the control, which represent the relative number of metabolically active cells remaining in culture by measuring the ATP content of each well.

Mouse contact hypersensitivity model

A hairless C57BL/6 mouse model was used for all in vivo studies. This mouse is fully immunocompetent, but possesses a gene mutation that causes alopecia after the first follicular maturation, post birth. This model is preferable, since topical treatments would necessitate hair removal in other mouse models, and shaving or depilatory agents can cause unwanted skin barrier defects. The mice examined were all male and between 5 and 6 months old, since we have identified age, but not sex-based differences in the DNFB induced CHS response of these mice [34]. The mice were housed individually in standard cages to prevent interactions that would damage the skin or lead to cross contamination of topical agents. The mice were kept on a 12 h light/dark cycle and had access to food and water ad libitum.

To measure the level of skin swelling in the contact hypersensitivity response, mice were sensitized to 30 µL 0.05% DNFB in a 4:1 acetone (Sigma Aldrich Cat# AGCN20-25 M) and olive oil (Wegman's Brand Pure Olive Oil) mixture on the lower dorsum on day 0. This

sensitization dose is lower than most doses reported in the literature; however, we have previously reported that the mouse model used in this study is more sensitive to DNFB induced skin irritation [34]. On day 5, the ear thickness of each mouse was measured with digital calipers before treatment with 20 μ L 0.2% DNFB in a 4:1 acetone and olive oil mixture on the right ear (10 μ L on each side of the ear). The left ear was treated with the same concentration of DNFB, but the MWCNT treatments were also added to this ear at a concentration of 1 μ g/ear. On day 6, the ear thickness was measured again by digital calipers, and the difference between the day 5 and day 6 measures represents the increase in ear swelling induced by the allergen and nanoparticles.

To determine the inflammatory effect of the DNFB and the MWCNT (High COOH), the CHS protocol was altered to challenge the lower dorsum. Briefly, the abdomen of each mouse was sensitized with 30 μ L 0.05% DNFB in a 4:1 acetone and olive oil mixture on day 0. On day 5, the mice were placed in Elizabethan collars to prevent grooming, and the mice were pre-treated with 50 μ L of a 10:1 acetone and water mixture with or without 5 μ g of MWCNT (High COOH) over a 5 cm^2 area of the lower dorsum. After 15 min, the mice were challenged with either 50 μ L of vehicle or 50 μ L 0.2% DNFB in a 4:1 acetone and olive oil mixture over the same 5 cm^2 area of the lower dorsum. On day 6, the mice were euthanized via carbon dioxide asphyxiation, and the skin was harvested for analysis.

Histology

After euthanasia, the treated dorsal skin was immediately cut into 1 $\text{cm} \times 1 \text{cm}$ squares and fixed in 10% formalin (Electron Microscopy Sciences Cat# 15740–04) for 48 h. The fixed skin was then embedded in paraffin blocks and 5 μ m transverse sections were cut with a microtome and placed on glass slides. Two sets of slides were stained with either hematoxylin/eosin (Electron Microscopy Sciences Cat# 26030–20/26051–21) for general histological examination, or 0.1% toluidine blue (MP biomedical Cat# 152649) for examination of mast cells and basophils. The stained images were obtained using a Nikon Eclipse E800 bright field microscope. The epidermal thickness was measured using the free source ImageJ software [75]. The mast cell and basophil counts were performed by counting both the total intact and degranulating cells over ten 20x fields of view. The PMN counts were performed by counting the number of cells over ten 40x fields of view.

Flow cytometry

Following euthanasia, 2 cm^2 of treated dorsal skin was excised and placed into 1 mL of 1% fetal bovine serum

in Hank's Balances Salt Solution (HBSS). An additional 1 mL of 0.1 U Collagenase/ 0.8 U Dispase (Sigma Aldrich Cat# 11097113001) was added to the skin samples along with 0.3 M calcium chloride, and the samples were incubated at 37° C for 30 min. After 30 min, the 0.5 M EDTA was added to the reaction to halt enzymatic activity. Each section of skin was homogenized and filtered through a 100 μ m filter to yield a single cell suspension. The cell suspension (1×10^6 cells) was washed with PBS and then blocked with 100 μ L of a 1:10 dilution of anti-mouse CD16/32 Fc blocking reagent (eBioscience Cat# 14–0161-82) for 15 min at 4° C. Again, the samples were washed with PBS, and 100 μ L of a 1:200 dilution of each antibody was added to the samples: CD11b (Biolegend Cat# 101243), CD11c (Biolegend Cat# 117331), F4/80 (Invitrogen Cat# 4323732), MHCII (eBioscience Cat# 47–5321-82), and fixable viability dye (Invitrogen Cat # 4339520). The samples were incubated in antibody for 30 min at 4° C, and then washed with PBS. Each sample was fixed by incubating in 250 μ L of a 2.5% formalin solution for 20 min at 4° C, before washing in PBS and suspending the samples in 300 μ L of PBS. Appropriate compensation controls and fluorescence minus one samples were prepared alongside the treated samples. The stained cell suspensions were run through an 18-color LSRII flow cytometer (BD Biosciences) at a 35 μ L/minute flow rate, for 5 min each. The data was analyzed using FlowJo version 8.8.7 (FlowJo LLC, Ashland, OR). (Example gating scheme in Additional file 1: Figure S4) All flow cytometry data presented are percentages of live cell populations in the skin sections.

Cytokine/chemokine analysis

A 1 $\text{cm} \times 1 \text{cm}$ piece of each treated skin sample was placed in 400 μ L of tissue protein extraction reagent (Thermoscientific Cat# 78510) with a 1:1000 dilution of HALT protease inhibitor cocktail (Thermoscientific Cat# 78,410) added. The skin samples were homogenized and incubated for 15 min at 4° C. The samples were then centrifuged for 5 min at 12,000 RPM to pellet the tissue debris. Protein concentrations were determined using a BCA protein assay (Pierce Cat# 23225). The following cytokines and chemokines were analyzed via a bead based ELISA kit (Millipore Cat# MYCYTOMAG-70 K): MCP-1, TNF α , IP-10, MIP-1 α , MIP-2, KC, IL-6, IL-1 β . The samples were all analyzed using the Bio-Plex 200 system (Bio-Rad), and at least 50 beads were analyzed per sample.

Statistics

All statistical analyses were performed with JMP Pro version 13.2.1 (SAS Institute Inc., Cary, NC). A two-way analysis of variance with post-hoc Tukey tests were used when appropriate, unless otherwise specified in the

figure legends. *P*-values < 0.05 were considered significant. All data are presented as means ± standard error of the mean (SEM).

Additional file

Additional file 1: Table S1. Carboxylated MWCNT XPS elemental survey. **Figure S1.** Pristine MWCNT display less HaCaT cytotoxicity and more ROS generation than oxidized MWCNT. **Figure S2.** In-house oxidized MWCNT representative XPS plots. **Figure S3.** In-house Oxidized MWCNT TEM images. **Figure S4.** Example flow cytometry gating strategy. (DOCX 2170 kb)

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Availability of data and materials

Data referenced in this article will be made available from the corresponding author upon request.

Authors' contributions

BCP and LAD designed experiments, analyzed data, and wrote the manuscript. SJP helped with data collection. All authors read and approved the final manuscript.

Ethics approval and consent to participate

All of the in vivo experiments were approved by the University Committee on Animal Resources (UCAR#2010–024/100360) at the University of Rochester Medical Center. All methods were performed in accordance to the relevant regulations.

Consent for publication

Not applicable.

Competing interests

The authors declare that they have no competing interests.

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Author details

¹Department of Environmental Medicine, University of Rochester Medical Center, New York, USA. ²Department of Biomedical Engineering, University of Rochester, Rochester, NY, USA. ³Department of Dermatology, University of Rochester Medical Center, Rochester, NY, USA. ⁴University of Rochester Medical Center, School of Medicine and Dentistry, 601 Elmwood Avenue, Box 697, Rochester, NY 14642, USA.

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