



# Mn<sup>2+</sup>-activated calcium fluoride nanoprobe for time-resolved photoluminescence biosensing

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**ABSTRACT** Time-resolved (TR) photoluminescence (PL) technique has shown great promise in ultrasensitive biodetection and high-resolution bioimaging. Hitherto, almost all the TRPL bioprobes are based on the parity-forbidden  $f \rightarrow f$  transition of lanthanide ions. Herein, we report TRPL biosensing by taking advantage of the  $d \rightarrow d$  transition of transition metal (TM) Mn<sup>2+</sup> ion. We demonstrate that the Förster resonance energy transfer (FRET) signal can be distinguished from that of radiative reabsorption process through measuring the PL lifetime of Mn<sup>2+</sup>, thus establishing a reliable method for Mn<sup>2+</sup> in homogeneous TR-FRET biodetection. We also demonstrate the biotin receptor-targeted cancer cell imaging by utilizing biotinylated CaF<sub>2</sub>:Ce,Mn nanoprobe. Furthermore, we show in a proof-of-concept experiment the application of the long-lived PL of Mn<sup>2+</sup> for TRPL bioimaging through the burst shot with a cell phone. These findings provide a general approach for exploiting the long-lived PL of TM ions for TRPL biosensing, thereby opening up a new avenue for the exploration of novel and versatile applications of TM ions.

**Keywords:** manganese, time-resolved photoluminescence, energy transfer, biodetection, bioimaging

## INTRODUCTION

Time-resolved (TR) photoluminescence (PL) technique has shown great promise in ultrasensitive biodetection and high-resolution bioimaging, owing to its ability to eliminate the short-lived background noise from biological autofluorescence and scatter light (in the nanosecond range) by setting a temporal delay in signal acquisition [1–5]. Hitherto, almost all the TRPL bioprobes are based on lanthanide chelates or lanthanide-doped nanoparticles (NPs), by taking advantage of the

long-lived PL ( $\mu\text{s}$ – $\text{ms}$  range) of lanthanide ions originating from the parity-forbidden intra- $4f^N$  electronic transition [6–10]. In addition to lanthanide ions, the forbidden  $d \rightarrow d$  transition of transition metal (TM) ions such as Mn<sup>2+</sup> with  $d^5$  configuration results in a long PL lifetime ranging from microseconds to tens of milliseconds [11–15]. The long PL lifetime of Mn<sup>2+</sup> ions makes it a potential bioprobe in TRPL biosensing, which, however, has scarcely been reported [16].

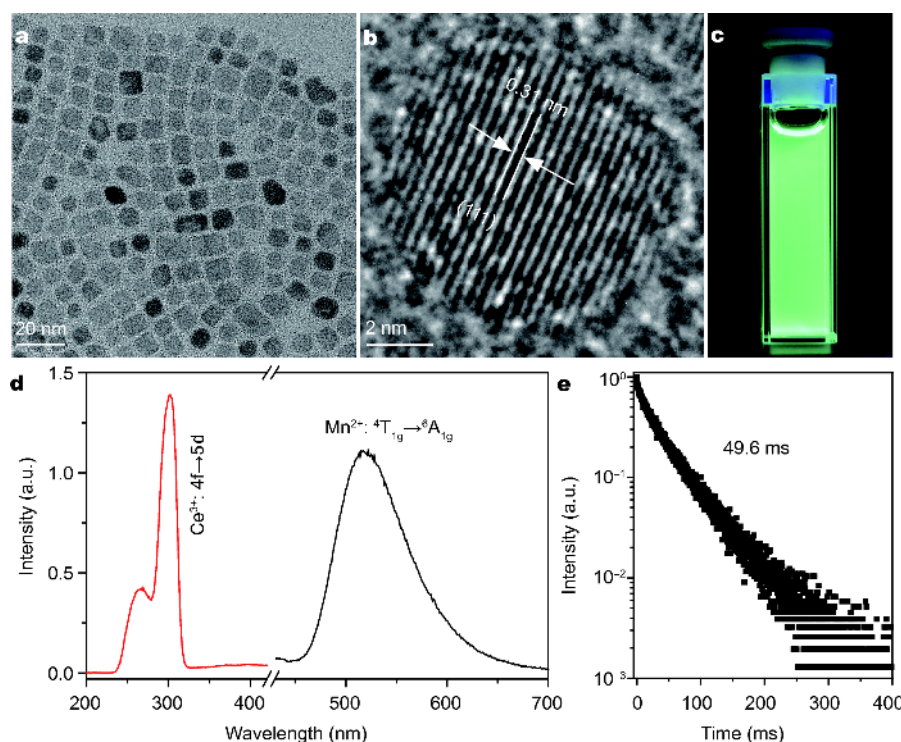
Mn<sup>2+</sup>-activated luminescent materials typically possess a broad emission band varying from green to deep red depending on the crystal field of Mn<sup>2+</sup> [17]. For example, the octahedrally coordinated Mn<sup>2+</sup> in CaF<sub>2</sub> lattice display green emission with a long PL lifetime up to tens of milliseconds [18–20]. Such long-lived and broad-band PL of Mn<sup>2+</sup> along with the excellent biocompatibility of CaF<sub>2</sub> is desirable for homogeneous TR Förster resonance energy transfer (FRET) biodetection [21–24]. The broad emission band of Mn<sup>2+</sup> enhances spectral overlap with the excitation band of the energy acceptor, resulting in an enhanced FRET efficiency [25–28]. Nonetheless, the reabsorption process, which is unavoidable in homogeneous luminescent bioassays, may interfere with the FRET signal and lead to incorrect detection results [29–31]. Therefore, it is of fundamental importance to gain deep insights into the energy transfer process and establish a reliable method for Mn<sup>2+</sup> in homogeneous TR-FRET biodetection.

Herein, we report the application of CaF<sub>2</sub>:Ce,Mn NPs for TRPL biodetection and bioimaging by utilizing the long-lived PL of Mn<sup>2+</sup>. We demonstrate that the PL lifetime of Mn<sup>2+</sup> can be exploited as a distinguishable FRET signal for Mn<sup>2+</sup> in TR-FRET biodetection based on the energy transfer between the NP donor and organic dye

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**Figure 1** (a) TEM and (b) HRTEM images of  $\text{CaF}_2$ :5%Ce,5%Mn NPs. (c) PL photograph of the NPs dispersed in cyclohexane under 304-nm UV lamp irradiation. (d) PL emission spectrum (black) of  $\text{CaF}_2$ :5%Ce,5%Mn NPs upon UV excitation at 304 nm, and their excitation spectrum (red) by monitoring the  $\text{Mn}^{2+}$  emission at 520 nm. (e) PL decay from  ${}^4\text{T}_{1g}$  by monitoring the  $\text{Mn}^{2+}$  emission at 520 nm.

acceptor in an avidin/biotin model system. We also demonstrate the biotin receptor-targeted cancer cell imaging by biotinylated  $\text{CaF}_2$ :Ce,Mn NPs as luminescent nanoprobes. Furthermore, we propose a convenient and versatile method, namely, the burst shot with a cell phone for TRPL bioimaging of zebrafish fed with the nanoprobes, thereby revealing the great potential of the forbidden  $d \rightarrow d$  transition of TM ions in TRPL biosensing.

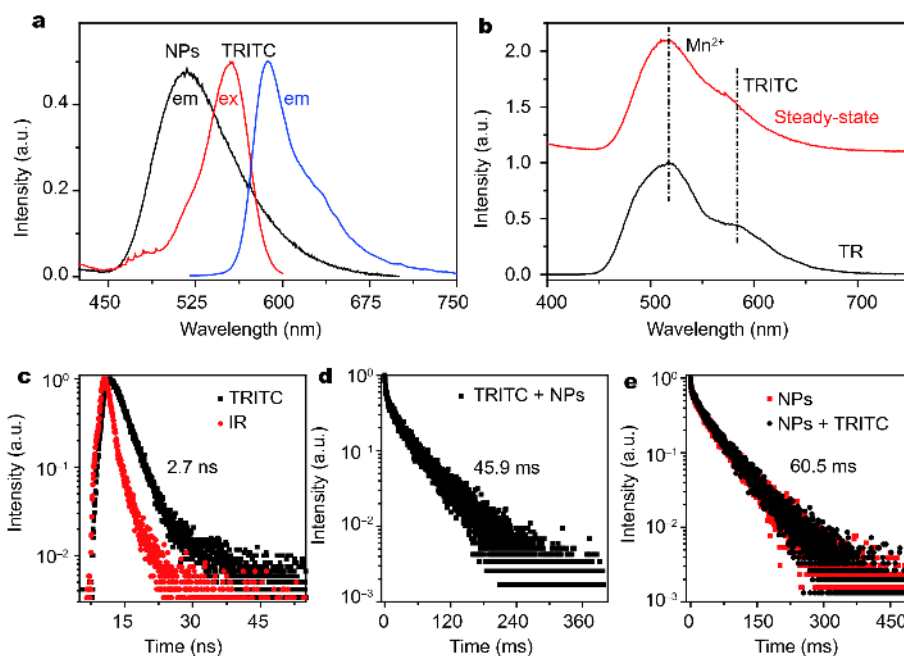
## RESULTS AND DISCUSSION

Monodisperse  $\text{CaF}_2$ :Ce,Mn NPs were synthesized through a solvothermal method in the presence of oleic acid as the surfactant and sodium ions as the codopant [32,33]. The as-synthesized NPs were hydrophobic and can be readily dispersed in nonpolar organic solvents such as cyclohexane. TEM image showed that the NPs were roughly cubic with an average length of  $6.0 \pm 1.1$  nm (Fig. 1a and Fig. S1a). High-resolution TEM image and XRD pattern confirmed the cubic phase and high crystallinity of the resulting NPs (Fig. 1b and Fig. S1b). Compositional analysis by EDS and ICP revealed the elements of Ca, F, Ce, Mn, and Na in the NPs (Fig. S1c and Table S1).

To achieve efficient PL in  $\text{CaF}_2$ :Ce,Mn NPs, the nom-

inal dopant concentrations of  $\text{Ce}^{3+}$  and  $\text{Mn}^{2+}$  were optimized to be 5 mol% and 5 mol%, respectively (Fig. S2). Upon UV excitation at 304 nm, bright green PL of the colloidal NPs can be explicitly observed in cyclohexane solution (Fig. 1c). PL emission spectrum showed a broad emission band centered at 520 nm (Fig. 1d), which can be assigned to the spin-forbidden  ${}^4\text{T}_{1g}(\text{G}) \rightarrow {}^6\text{A}_{1g}(\text{S})$  transition of  $\text{Mn}^{2+}$  occupying an octahedral site ( $\text{O}_h$ ) in  $\text{CaF}_2$  lattice [18]. By monitoring the  $\text{Mn}^{2+}$  emission at 520 nm, two broad excitation bands centered at 260 nm and 304 nm were detected. The excitation band at 304 nm can be assigned to the  $4f \rightarrow 5d$  transition of  $\text{Ce}^{3+}$  in  $\text{C}_{4v}$  center, indicating an energy transfer from  $\text{Ce}^{3+}$  to  $\text{Mn}^{2+}$  [19]. The excitation band at 260 nm is associated with the  $\text{Ce}^{3+}$  transitions in clusters (Fig. S3) [20]. The PL decay from  ${}^4\text{T}_{1g}$  indicates an ultralong effective PL lifetime of 49.6 ms for the  $\text{Mn}^{2+}$  emission (Fig. 1e and Table S2), typical of the forbidden  $d \rightarrow d$  transition within  $\text{Mn}^{2+}$  ions [17].

For bioapplications, we rendered the hydrophobic  $\text{CaF}_2$ :Ce,Mn NPs to be hydrophilic and biocompatible by removing the oleate ligands from their surface through an acid-washing treatment [34], which was confirmed by Fourier transform infrared (FTIR) spectra and thermo-



**Figure 2** (a) PL emission spectrum of  $\text{CaF}_2\text{:}5\%\text{Ce},5\%\text{Mn}$  NPs (black); excitation (red) and emission (blue) spectra of TRITC. (b) Steady-state and TRPL (delay time=100  $\mu\text{s}$ , gate time=5 ms) emission spectra for the aqueous solution containing 50  $\mu\text{mol L}^{-1}$  of PAA-capped NPs and 10  $\text{nmol L}^{-1}$  of TRITC upon UV excitation at 304 nm. (c) PL decays of TRITC by monitoring its emission at 650 nm upon excitation with a 397-nm nanosecond pulsed laser. IR denotes the instrument response. (d) PL decay of the NPs-TRITC mixture by monitoring the TRITC emission at 650 nm under excitation at 304 nm. (e) PL decays of PAA-capped NPs (black) and NPs-TRITC mixture (red) by monitoring the  $\text{Mn}^{2+}$  emission at 520 nm under excitation at 304 nm.

gravimetric analysis (TGA) for NPs before and after acid-washing treatment (Figs S4, S5). As a result, these ligand-free NPs exhibited well water solubility. More importantly, we found that the integrated PL intensity of ligand-free NPs was enhanced by a factor of 3.1 relative to their oleate-capped counterparts, with their PL lifetime increasing from 49.6 ms to 59.5 ms (Fig. S6). Accordingly, the absolute PL quantum yield, defined as the ratio of the number of emitted photons to the number of absorbed photons, was determined to increase remarkably from  $15.3 \pm 0.8\%$  in oleate-capped NPs to  $37.0 \pm 2.3\%$  in ligand-free NPs, which is attributed to the increased absorption of the excitation light by  $\text{Ce}^{3+}$  followed by energy transfer to  $\text{Mn}^{2+}$  [35]. The long PL lifetime is probably due to the change of refractive index of the surrounding medium, which changed the radiative decay rate of  $\text{Mn}^{2+}$ , as well documented in  $\text{LaPO}_4\text{:Ce/Tb}$  NPs with nanocrystal-cavity model proposed by Meijerink *et al.* [36].

Due to the removal of surface ligands, positively charged  $\text{Ca}^{2+}$  ions were exposed on the surface of ligand-free NPs, endowing the NPs with a zeta potential of +48.2 mV at pH 6.9. As a consequence, these ligand-free NPs are allowed for direct conjugation with electronegative groups of hydrophilic and biocompatible molecules such

as biotin and poly(acrylic acid) (PAA) through the strong chelation of  $\text{Ca}^{2+}$  [37,38]. The conjugation of biotin and PAA to the NP surface was confirmed by FTIR spectra, TGA, dynamic light scattering and zeta potential measurements (Figs S4, S5, and S7). The PAA-capped NPs and biotinylated NPs preserved the intense PL of ligand-free NPs with PL lifetimes of 60.5 ms and 54.1 ms, respectively (Fig. S6 and Table S2).

To validate the feasibility of the long-lived and broadband PL of  $\text{Mn}^{2+}$  for homogeneous TR-FRET biodetection, we selected  $\text{CaF}_2\text{:Ce,Mn}$  NPs and tetramethylrhodamine B isothiocyanate (TRITC) as the energy donor and acceptor, respectively, in view of the large spectral overlap between the emission band of the NPs (450–700 nm) and the excitation band of TRITC (450–600 nm) (Fig. 2a). Fig. 2b compares the steady-state with TRPL emission spectra for the aqueous solution containing 50  $\mu\text{mol L}^{-1}$  of PAA-capped NPs and 10  $\text{nmol L}^{-1}$  of TRITC upon UV excitation at 304 nm, which show similar emission bands from 450 to 700 nm consisting of both  $\text{Mn}^{2+}$  and TRITC emissions. The steady-state PL emissions of  $\text{Mn}^{2+}$  and TRITC were attributed to direct excitation of the NPs and TRITC by the 304-nm UV light (Fig. S8). In the TRPL spectrum, the intrinsic short-lived PL (2.7 ns) of TRITC

arising from direct excitation (Fig. 2c) was eliminated by setting a delay time of 100  $\mu$ s. Thus, the long-lived PL of TRITC in the TRPL spectrum was ascribed to the energy transfer from the NPs to TRITC (Fig. S9), which caused the longer PL lifetime of TRITC from 2.7 ns to 45.9 ms due to the slow population of the TRITC excited state from the long-lived  $\text{Mn}^{2+}$  excited state (Fig. 2d) [25]. Because no specific binding between PAA-capped NPs and TRITC could bring them within effective FRET distance [39], we deduced that the NPs-to-TRITC energy transfer is governed by a radiative reabsorption process rather than a non-radiative FRET process. Such radiative energy transfer was further evidenced by the identical PL lifetime of  $\text{Mn}^{2+}$  in either NPs-TRITC mixture or pure PAA-capped NPs (Fig. 2e), since non-radiative FRET always results in a decrease in PL lifetime of energy donor by imposing additional relaxation channel on the donor [40]. These results unveil that the radiative energy transfer from long-lived energy donor to short-lived energy acceptor is able to lengthen the PL lifetime of the acceptor. Note that this is the first demonstration on PL lifetime lengthening of short-lived energy acceptor by long-lived donor through radiative reabsorption process, which is very important in homogeneous luminescent bioassays.

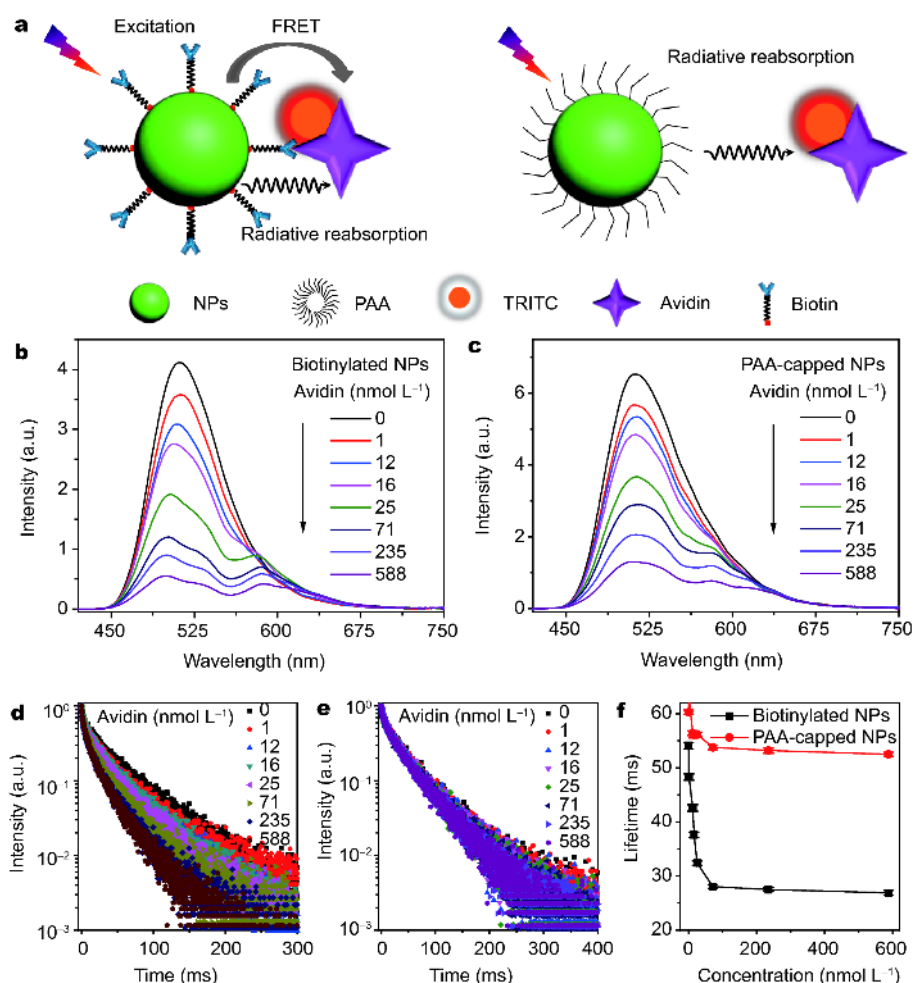
By employing biotinylated  $\text{CaF}_2\text{:Ce,Mn}$  NPs and TRITC-labelled avidin, we constructed the FRET pair in an avidin/biotin model system, in which the excitation energy was transferred from the NPs to nearby TRITC through either radiative reabsorption or FRET as a result of specific binding between avidin and biotin [38]. For non-binding control, we used PAA-capped NPs as the energy donor under otherwise identical conditions, whereby the radiative reabsorption process was the only energy transfer route (Fig. 3a). Because both the radiative reabsorption and FRET processes can lengthen the PL lifetime of TRITC (Fig. S10), the  $\text{Mn}^{2+}$  and TRITC emissions were detected in the TRPL spectra for either biotinylated or PAA-capped NPs incubated with TRITC-labelled avidin, yielding nearly identical emission patterns (Fig. 3b, c). As a result, the FRET signal was undistinguishable and submerged in the TRPL signal arising from radiative reabsorption, which makes the TRPL spectrum unreliable for homogeneous TR-FRET biodection.

We measured the PL lifetime of the energy donor to distinguish the FRET signal from that of the radiative reabsorption process. Fig. 3d shows the concentration-dependent PL lifetimes of  $\text{Mn}^{2+}$  in the mixture of biotinylated NPs and TRITC-labelled avidin. It was found that

the PL lifetime of  $\text{Mn}^{2+}$  in biotinylated NPs decreased from 54.1 to 26.4 ms as the avidin concentration increased from 0 to 588  $\text{nmol L}^{-1}$ , as a result of FRET from the NPs to TRITC. By contrast, in non-binding control, the PL lifetime of  $\text{Mn}^{2+}$  in PAA-capped NPs showed only a slight decrease at high avidin concentrations, attributed to non-specific binding between PAA-capped NPs and TRITC-labelled avidin, which can be avoided by blocking the residual active binding site of PAA-capped NPs for avidin through surface modification. This enables us to quantify the avidin concentration by measuring the PL lifetime of  $\text{Mn}^{2+}$  (Fig. 3f). The detection limit, defined as the concentration that corresponds to 3 times the standard deviation below the signal measured in the blank control, was determined to be 32  $\text{pmol L}^{-1}$ . This value is approximately one order of magnitude improvement relative to that in TR-FRET assays based on lanthanide-doped nanoprobe ever reported [33]. These results show that the PL lifetime of  $\text{Mn}^{2+}$  is sensitive to FRET and barely affected by the radiative reabsorption process, thus validating its reliability and advantages in TR-FRET bioassay.

Utilizing biotinylated  $\text{CaF}_2\text{:Ce,Mn}$  NPs, we also demonstrated biotin receptor-targeted cancer cell imaging. Biotin is a growth promoter at the cellular level, and biotin receptors are overexpressed in many cancer cells, including colon (Colo-26), lung (M109), renal (RD0995), ovarian (Ov2008) and cervical (HeLa) cancer cell lines [41,42]. We selected HeLa cells with biotin receptors overexpressed on the membrane as the target cancer cells and human normal liver (L-02) cells with low-expressed biotin receptors as the control. Owing to the high affinity between biotin and biotin receptors, biotinylated NPs can specifically target to HeLa cells, leading to bright green PL (green channel) of  $\text{Mn}^{2+}$  surrounding HeLa cells (Fig. 4a). By contrast, the green PL of  $\text{Mn}^{2+}$  was hardly observed on L-02 cells under otherwise identical conditions due to the lack of specific recognition between biotinylated NPs and L-02 cells (Fig. 4b). MTT assay on L-02 cells incubated with biotinylated NPs showed a cell viability larger than 95% even at a high NP concentration of 1  $\text{mg mL}^{-1}$  (Fig. S11), indicating that biotinylated  $\text{CaF}_2\text{:Ce,Mn}$  NPs are biocompatible and nontoxic to live cells. Our results show that  $\text{CaF}_2\text{:Ce,Mn}$  NPs modified with specific capture molecules, like biotin for biotin receptor, can be used as effective luminescent nano-bioprobes for targeted tumor imaging.

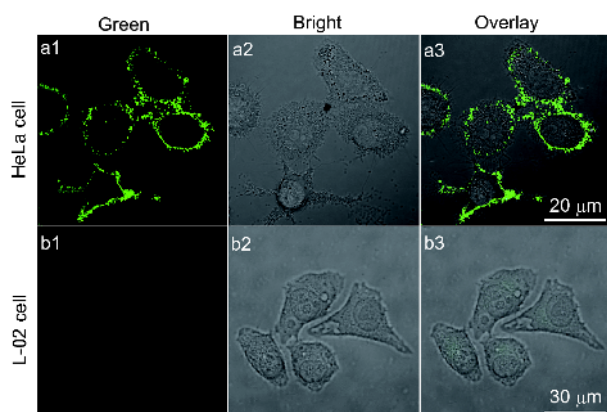
Furthermore, by means of the camera burst mode on a cell phone [43], we demonstrated in a proof-of-concept experiment the application of  $\text{CaF}_2\text{:Ce,Mn}$  nanoprobe



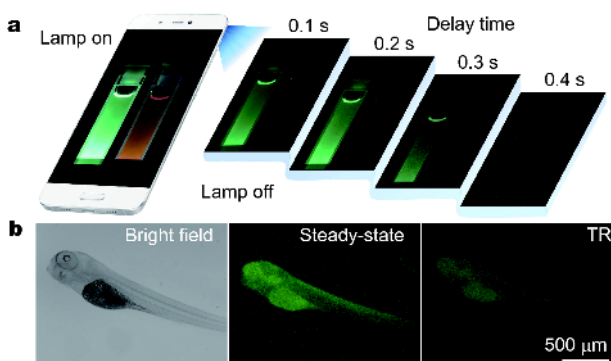
**Figure 3** (a) Schematic illustration of the energy transfer processes between CaF<sub>2</sub>:Ce,Mn NPs and TRITC in cases of specific binding (left) and non-specific binding (right). TRPL spectra of (b) biotinylated and (c) PAA-capped CaF<sub>2</sub>:Ce,Mn NPs incubated with TRITC-labeled avidin as a function of the avidin concentration. PL decays from (d) biotinylated and (e) PAA-capped CaF<sub>2</sub>:Ce,Mn NPs at different avidin concentrations by monitoring the Mn<sup>2+</sup> emission at 520 nm. (f) Effective PL lifetime of <sup>4</sup>T<sub>1g</sub> of Mn<sup>2+</sup> as a function of the avidin concentration, as obtained from (d, e). Each PL lifetime was measured independently for three times to yield the average value and standard deviation.

for TRPL bioimaging. The fast burst mode on HUAWEI P10 cell phone is able to continuously capture 10 photos in 1 s with an exposure time of 0.1 s for each photo. Fig. 5a compares the PL photographs for the aqueous solution of ligand-free NPs and TRITC, respectively. The NPs and TRITC solution displayed bright green and orange PL, respectively, under 304-nm UV lamp irradiation. When the lamp was off, the TRITC PL vanished immediately in the photos captured by the camera due to the short-lived PL (~2.7 ns) of TRITC. By contrast, the green PL of Mn<sup>2+</sup> can last for ~400 ms and remained explicitly visualized in the first four photos after the lamp was off. This suggests that the cell phone can be explored as a convenient and efficient detector to capture the long-lived PL of Mn<sup>2+</sup> for TRPL bioimaging. To demonstrate this concept, we car-

ried out steady-state and TRPL imaging of 5-day-old zebrafish after feeding it with ligand-free CaF<sub>2</sub>:Ce,Mn NPs through burst shot with the cell phone. In the steady-state PL image when the lamp was on, the green PL was observed in both the zebrafish and the background, whereas in the TRPL image when the lamp was off, the green PL was observed exclusively in the zebrafish (Fig. 5b), confirming that the PL signal in the zebrafish originated from the digested CaF<sub>2</sub>:Ce,Mn NPs. These results verify that the TRPL technique based on the long-lived PL of Mn<sup>2+</sup> can effectively suppress the short-lived background noise and offer improved imaging sensitivity relative to that of steady-state PL. Along with the large longitudinal magnetic relaxivity of Mn<sup>2+</sup> (Fig. S12), these CaF<sub>2</sub>:Ce,Mn NPs may function as TRPL/MRI dual-mode



**Figure 4** Confocal laser scanning microscopy images of (a1–a3) HeLa cells and (b1–b3) L-02 cells after incubation with biotinylated  $\text{CaF}_2:\text{Ce},\text{Mn}$  NPs ( $0.5 \text{ mg mL}^{-1}$ ) at  $37^\circ\text{C}$  for 2 h. Intense green PL of  $\text{Mn}^{2+}$  ( $\lambda_{\text{em}}=500\text{--}560 \text{ nm}$ ,  $\lambda_{\text{ex}}=408 \text{ nm}$ ) was observed exclusively in HeLa cells. Panel 1 and 2 show the green PL images and bright-field images, respectively. Panel 3 is the overlay images of panels 1 and 2.



**Figure 5** (a) Schematic representation of the camera burst mode on a cell phone. The image in the cell phone represents the steady-state PL photo for the aqueous solution of ligand-free  $\text{CaF}_2:\text{Ce},\text{Mn}$  NPs (left) and TRITC (right) under 304-nm UV lamp irradiation. The images outside the cell phone denote the corresponding TRPL photos of the NPs and TRITC, captured by the cell phone in sequence with a time interval of 0.1 s. (b) Microscopic bright field, steady-state and TRPL images of the 5-day-old zebrafish fed with ligand-free  $\text{CaF}_2:\text{Ce},\text{Mn}$  NPs ( $0.5 \text{ mg mL}^{-1}$ ).

nano-bioprobes. It is worthy of mentioning that, these  $\text{CaF}_2:\text{Ce},\text{Mn}$  NPs are more suitable for *in-vitro* biodetection and bioimaging than *in-vivo* applications due to the shallow tissue penetration depth of visible light.

## CONCLUSIONS

In summary, we have demonstrated the advantages of  $\text{CaF}_2:\text{Ce},\text{Mn}$  NPs as TRPL nano-bioprobes for sensitive biodetection and high-resolution bioimaging based on the long-lived PL of  $\text{Mn}^{2+}$ . Our mechanistic investigation on the energy transfer processes revealed that the FRET

signal can be distinguished from that of reabsorption process by measuring the PL lifetime of  $\text{Mn}^{2+}$  instead of the TRPL spectrum, thus establishing a reliable tool for homogeneous TR-FRET bioassay. The proposed burst shot with a cell phone can effectively capture the long-lived PL of  $\text{Mn}^{2+}$ , providing a convenient and versatile approach for TRPL bioimaging without the need of a sophisticated instrument. These findings offer new routes to the development of ultrasensitive TRPL biosensing by exploiting the forbidden  $d \rightarrow d$  transitions of TM ions, thereby opening up a new avenue for clinical applications, such as *in-vitro* detection and targeted cancer imaging.

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- 1 Siitari H, Hemmilä I, Soini E, *et al.* Detection of hepatitis B surface antigen using time-resolved fluoroimmunoassay. *Nature*, 1983, 301: 258–260
- 2 Connally RE, Piper JA. Time-gated luminescence microscopy. *Ann New York Acad Sci*, 2008, 1130: 106–116
- 3 Jin D, Piper JA, Leif RC, *et al.* Time-gated flow cytometry: an ultra-high selectivity method to recover ultra-rare-event  $\mu$ -targets in high-background biosamples. *J Biomed Opt*, 2009, 14: 024023
- 4 Zheng W, Tu D, Huang P, *et al.* Time-resolved luminescent biosensing based on inorganic lanthanide-doped nanopropes. *Chem Commun*, 2015, 51: 4129–4143
- 5 Song B, Ye Z, Yang Y, *et al.* Background-free *in-vivo* imaging of vitamin C using time-gateable responsive probe. *Sci Rep*, 2015, 5: 14194
- 6 Bünzli JCG, Piguet C. Taking advantage of luminescent lanthanide ions. *Chem Soc Rev*, 2005, 34: 1048–1077
- 7 Yuan J, Wang G. Lanthanide-based luminescence probes and time-resolved luminescence bioassays. *TRAC Trends Anal Chem*, 2006, 25: 490–500
- 8 Bouzigues C, Gacoin T, Alexandrou A. Biological applications of rare-earth based nanoparticles. *ACS Nano*, 2011, 5: 8488–8505
- 9 Huang P, Tu D, Zheng W, *et al.* Inorganic lanthanide nanopropes for background-free luminescent bioassays. *Sci China Mater*, 2015, 58: 156–177
- 10 Zheng W, Zhou S, Xu J, *et al.* Ultrasensitive luminescent *in vitro* detection for tumor markers based on inorganic lanthanide nanopropes. *Adv Sci*, 2016, 3: 1600197
- 11 Klick CC, Schulman JH. On the luminescence of divalent manganese in solids. *J Opt Soc Am*, 1952, 42: 910–916
- 12 He Y, Wang HF, Yan XP. Exploring Mn-doped ZnS quantum dots for the room-temperature phosphorescence detection of enoxacin in biological fluids. *Anal Chem*, 2008, 80: 3832–3837
- 13 Zhao Q, Huang C, Li F. Phosphorescent heavy-metal complexes for bioimaging. *Chem Soc Rev*, 2011, 40: 2508–2524
- 14 Deng X, Dai Y, Liu J, *et al.* Multifunctional hollow  $\text{CaF}_2:\text{Yb}^{3+}/\text{Er}^{3+}/\text{Mn}^{2+}$ -poly(2-Aminoethyl methacrylate) microspheres for Pt(IV) pro-drug delivery and tri-modal imaging. *Biomaterials*, 2015, 50: 154–163
- 15 Liu X, Wang Y, Li X, *et al.* Binary temporal upconversion codes of  $\text{Mn}^{2+}$ -activated nanoparticles for multilevel anti-counterfeiting. *Nat Commun*, 2017, 8: 899

- 16 Zhu D, Chen Y, Jiang L, *et al.* Manganese-doped ZnSe quantum dots as a probe for time-resolved fluorescence detection of 5-fluorouracil. *Anal Chem*, 2011, 83: 9076–9081
- 17 Blasse G, Grabmaier BC. *Luminescent Materials*. Berlin: Springer-Verlag, 1994
- 18 Alcalá R, Alonso PJ, Lalinde G, *et al.* Manganese centers in low temperature X-irradiated CaF<sub>2</sub>:Mn. *Phys Stat Sol b*, 1980, 98: 315–322
- 19 McKeever SWS, Brown MD, Abbundi RJ, *et al.* Characterization of optically active sites in CaF<sub>2</sub>:Ce,Mn from optical spectra. *J Appl Phys*, 1986, 60: 2505–2510
- 20 G ÜC. On the Ce–Mn clustering in CaF<sub>2</sub> in which the Ce<sup>3+</sup>→Mn<sup>2+</sup> energy transfer occurs. *J Phys-Condens Matter*, 2003, 15: 3821–3830
- 21 Lundin K, Blomberg K, Nordström T, *et al.* Development of a time-resolved fluorescence resonance energy transfer assay (cell t-fret) for protein detection on intact cells. *Anal Biochem*, 2001, 299: 92–97
- 22 Wang L, Tan W. Multicolor FRET silica nanoparticles by single wavelength excitation. *Nano Lett*, 2006, 6: 84–88
- 23 Shen J, Sun LD, Zhu JD, *et al.* Biocompatible bright YVO<sub>4</sub>:Eu nanoparticles as versatile optical bioprobes. *Adv Funct Mater*, 2010, 20: 3708–3714
- 24 Li Z, Zhang Y, Huang L, *et al.* Nanoscale “fluorescent stone”: luminescent calcium fluoride nanoparticles as theranostic platforms. *Theranostics*, 2016, 6: 2380–2393
- 25 Kaiser U, Sabir N, Carrillo-Carrion C, *et al.* Förster resonance energy transfer mediated enhancement of the fluorescence lifetime of organic fluorophores to the millisecond range by coupling to Mn-doped CdS/ZnS quantum dots. *Nanotechnology*, 2016, 27: 055101
- 26 Qiao J, Xia Z, Zhang Z, *et al.* Near UV-pumped yellow-emitting Sr<sub>9</sub>MgLi(PO<sub>4</sub>)<sub>7</sub>:Eu<sup>2+</sup> phosphor for white-light LEDs. *Sci China Mater*, 2018, 61: 985–992
- 27 Yu M, Qu Y, Pan K, *et al.* Enhanced photoelectric conversion efficiency of dye-sensitized solar cells by the synergetic effect of NaYF<sub>4</sub>:Er<sup>3+</sup>/Yb<sup>3+</sup> and g-C<sub>3</sub>N<sub>4</sub>. *Sci China Mater*, 2017, 60: 228–238
- 28 Wang D, Wang R, Liu L, *et al.* Down-shifting luminescence of water soluble NaYF<sub>4</sub>:Eu<sup>3+</sup>@Ag core-shell nanocrystals for fluorescence turn-on detection of glucose. *Sci China Mater*, 2017, 60: 68–74
- 29 Kochubey VI, Konyukhova YG, Zabenkov IV, *et al.* Accounting for scattering and reabsorption in the analysis of luminescence spectra of nanoparticles. *Quantum Electron*, 2011, 41: 335–339
- 30 Anni M, Alemanno E, Creti A, *et al.* Interplay between amplified spontaneous emission, Förster resonant energy transfer, and self-absorption in hybrid poly(9,9-dioctylfluorene)-CdSe/ZnS nanocrystal thin films. *J Phys Chem A*, 2010, 114: 2086–2090
- 31 Achatz DE, Meier RJ, Fischer LH, *et al.* Luminescent sensing of oxygen using a quenchable probe and upconverting nanoparticles. *Angew Chem Int Ed*, 2011, 50: 260–263
- 32 Wang G, Peng Q, Li Y. Upconversion luminescence of monodisperse CaF<sub>2</sub>:Yb<sup>3+</sup>/Er<sup>3+</sup> nanocrystals. *J Am Chem Soc*, 2009, 131: 14200–14201
- 33 Zheng W, Zhou S, Chen Z, *et al.* Sub-10 nm lanthanide-doped CaF<sub>2</sub> nanoprobe for time-resolved luminescent biodetection. *Angew Chem Int Ed*, 2013, 52: 6671–6676
- 34 Bogdan N, Vetrone F, Ozin GA, *et al.* Synthesis of ligand-free colloiddally stable water dispersible brightly luminescent lanthanide-doped upconverting nanoparticles. *Nano Lett*, 2011, 11: 835–840
- 35 Chen X, Jin L, Sun T, *et al.* Energy migration upconversion in Ce (III)-doped heterogeneous core-shell-shell nanoparticles. *Small*, 2017, 13: 1701479
- 36 Senden T, Rabouw FT, Meijerink A. Photonic effects on the radiative decay rate and luminescence quantum yield of doped nanocrystals. *ACS Nano*, 2015, 9: 1801–1808
- 37 Huang P, Zheng W, Zhou S, *et al.* Lanthanide-doped LiLuF<sub>4</sub> upconversion nanoprobe for the detection of disease biomarkers. *Angew Chem Int Ed*, 2014, 53: 1252–1257
- 38 Zhou S, Zheng W, Chen Z, *et al.* Dissolution-enhanced luminescent bioassay based on inorganic lanthanide nanoparticles. *Angew Chem Int Ed*, 2014, 108: 12498–12502
- 39 Su Q, Feng W, Yang D, *et al.* Resonance energy transfer in upconversion nanoplateforms for selective biodetection. *Acc Chem Res*, 2017, 50: 32–40
- 40 Tu D, Liu L, Ju Q, *et al.* Time-resolved FRET biosensor based on amine-functionalized lanthanide-doped NaYF<sub>4</sub> nanocrystals. *Angew Chem Int Ed*, 2011, 50: 6306–6310
- 41 Russell-Jones G, McTavish K, McEwan J, *et al.* Vitamin-mediated targeting as a potential mechanism to increase drug uptake by tumours. *J InOrg Biochem*, 2004, 98: 1625–1633
- 42 Chen S, Zhao X, Chen J, *et al.* Mechanism-based tumor-targeting drug delivery system. Validation of efficient vitamin receptor-mediated endocytosis and drug release. *Bioconjugate Chem*, 2010, 21: 979–987
- 43 [https://en.wikipedia.org/wiki/Burst\\_mode\\_\(photography\)](https://en.wikipedia.org/wiki/Burst_mode_(photography)), Wikipedia

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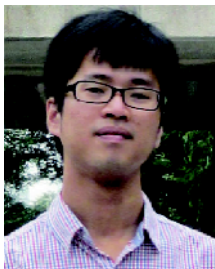
**Author contributions** Wei J, Zheng W and Chen X conceived the project, wrote the paper and were primarily responsible for the experiments. Shang X and Li R carried out PL measurements. Huang P and Gong Z synthesized and characterized the NCs. Liu Y, Zhou S and Chen Z contributed to the TRPL biodetection and bioimaging. All authors contributed to the general discussion and revision of the manuscript.

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**Supplementary information** Experimental details and supporting data are available in the online version of the paper.



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## 基于 $\text{Mn}^{2+}$ 激活氟化钙纳米荧光探针的时间分辨荧光生物分析

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**摘要** 时间分辨荧光探测技术在超灵敏生物检测和高分辨生物成像领域具有广泛的应用前景. 目前报道的时间分辨荧光生物探针大都是利用稀土离子 $4f^n$ 电子组态间的禁戒跃迁发光. 本文报道了基于过渡金属 $\text{Mn}^{2+}$ 离子 $d \rightarrow d$ 禁戒跃迁发光的时间分辨荧光生物分析. 我们证明通过测试 $\text{Mn}^{2+}$ 的荧光寿命变化可以将荧光共振能量传递与辐射再吸收信号区分开来, 从而为 $\text{Mn}^{2+}$ 发光在时间分辨荧光共振能量传递均相生物检测的应用提供了一种可靠的分析方法. 利用生物素化的 $\text{CaF}_2:\text{Ce}, \text{Mn}$ 纳米荧光探针, 我们还实现了对生物素受体过表达癌细胞的靶向荧光成像. 通过概念性验证并利用手机连拍功能, 我们证明了 $\text{Mn}^{2+}$ 的长寿命发光可用于时间分辨荧光生物成像. 这些研究结果为过渡金属长寿命发光在时间分辨荧光生物分析领域的应用提供了普适方法, 也为过渡金属离子的新型、多功能用途开辟了新的方向.