ORIGINAL ARTICLE





A Fragmentation Study on Four Oligostilbenes by Electrospray Tandem Mass Spectrometry

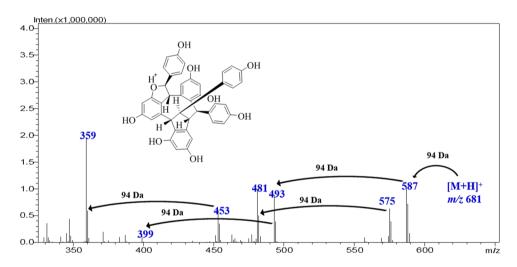
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Abstract

Oligostilbenes have attracted much interest due to their intricate structures and diverse bioactivities. In this study, two stilbene dimers, (–)-7,8-cis- ε -viniferin (1) and carasiphenol A (2), and two trimers, suffruticosol A (3) and suffruticosol C (4), were investigated by electrospray ionization ion-trap time-of-flight multistage mass spectrometry (ESI-IT-TOF-MSⁿ). Based on the MSⁿ study, the fragmentation pathways and diagnostic ions of four oligostilbenes in both positive and negative modes were proposed. The consecutive elimination of phenol (C_6H_6O) and resorcinol ($C_6H_6O_2$) moieties were the particular dissociation for oligostilbenes due to the presence of 1,2-diphenylethylene nucleus. The present MSⁿ fragmentation study will provide valuable information for the fast characterization of oligostilbenes from complicated natural mixtures.

Graphical Abstract



Keywords ESI-IT-TOF–MSⁿ · Fragmentation rules · Oligostilbenes

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1 Introduction

Natural stilbenes are an important group of polyphenols characterized by the presence of 1,2-diphenylethylene nucleus [1]. Naturally occurring stilbenes always have intricate structures with different numbers of stilbenes and polymeric types [2]. Stilbenes as the defensive chemicals of plants are revealed with diverse bioactivities including anti-tumor, anti-oxidant, anti-inflammatory,



anti-fungal, anti-diabetic and anti-Alzheimer's disease effects [3–7], whereas their natural distribution is quite limited, mainly in Cyperaceae, Dipterocarpaceae, Gnetaceae, Iridaceae, Leguminosae, Moraceae, Orchidaceae and Polygonaceae plants [2]. Thus, it is necessary to develop and establish a rapid and systematic method to profile stilbenes from natural resources.

Mass spectrometry (MS) with high sensitivity and resolution is one of the most efficient method in analyzing natural products [8–10]. Tandem MS techniques have advantages in ascertaining the relationship between precursor and product ions, by which the fragmentation rules and diagnostic ions of complicated compounds can be proposed [11, 12]. In this paper, we report the MSⁿ fragmentation rules of four oligostilbenes, (–)-7,8-cis-eviniferin (1), carasiphenol A (2), suffruticosol A (3) and suffruticosol C (4), by electrospray ionization ion-trap time-of-flight (ESI-IT-TOF) mass spectrometer to provide reference for their fast characterization from natural sources.

2 Results and Discussion

The first-stage MS of compounds 1–4 (Fig. 1) in both positive and negative ion modes were acquired in automatic pattern, by which their protonated ([M+H]⁺) and deprotonated ([M-H]⁻) molecule ions were readily detected. For compounds 1, 3 and 4, the [M+HCOO]⁻ ions in negative mode were also obtained due to the application of formic acid in the solvent [13]. The subsequent MSⁿ studies on compounds 1–4 in both positive and negative modes were performed, from which their fragmentation pathways were proposed (Figs. 2, 3, 4, 5, 6, 7, 8, 9). It should be noted that alternative ways of fragmentation that can reasonably interpret the product ions are also possible in addition to the proposed pathway. For example, the negative charge can be present at any hydroxy group rather than the position denoted.

2.1 MSⁿ Fragmentations of (–)-7,8-*cis-ε*-Viniferin (1) in Positive Mode

In the single-stage mass spectrum of (-)-7,8-cis- ε -viniferin (1), the [M+H]⁺ ion at m/z 455 (1A), [M-H]⁻ ion at m/z 453 (1b) and [M+HCOO]⁻ ion at

Fig. 1 Structures of compounds 1–4

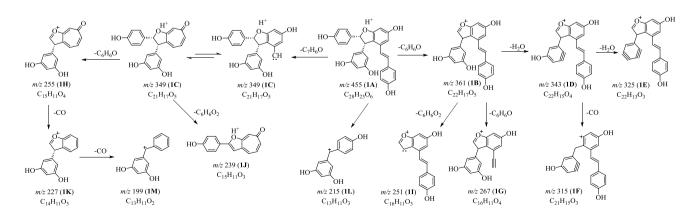


Fig. 2 Proposed fragmentation pathways of (-)-7,8-cis- ε -viniferin (1) in positive mode



Fig. 3 Proposed fragmentation pathways of (-)-7,8-cis- ε -viniferin (1) in negative mode

Fig. 4 Proposed fragmentation pathways of carasiphenol A (2) in positive mode

m/z 499 (1a) were readily obtained, corresponding to the molecular formula of $C_{28}H_{22}O_6$. The subsequent MS² experiment on [M+H]⁺ (1A) gave rise to multiple product ions (1B-1L). The ion 1A lost a C_6H_6O or a C_7H_6O moiety to provide the ions of 1B (m/z 361) and 1C (m/z 349) [14]. Then, the subsequent elimination of two H_2O molecules from 1B generated ions at m/z 343 and m/z 325. The ion 1C was present in high abundance, which might be due to the rearrangement of 1C into tropone derivative. The ions 1G (m/z 267) and 1I (m/z 251) were proposed from 1B by the elimination of a C_6H_6O or a $C_6H_6O_2$ part.

The similar fragmentation was also observed for ion 1C, from which two product ions at m/z 255 (1H) and m/z 239 (1J) were obtained corresponding to neutral loss of a C_6H_6O or a $C_6H_6O_2$ part. When ion 1H was selected as the precursor ion to perform MS³ experiment, two ions 1K (m/z 227) and 1M (m/z 199) were formed by the successive loss of two CO molecules. Similarly, the ion 1F (m/z 315) was well explained by the loss of a molecule of CO from 1D [15–17]. In addition, a MS² ion at m/z 215 (1L) was also observed from 1A, but its fragmentation pathway was still unclear (Fig. 2).



Fig. 5 Proposed fragmentation pathways of carasiphenol A (2) in negative mode

Fig. 6 Proposed fragmentation pathways of suffruticosol A (3) in positive mode



Fig. 7 Proposed fragmentation pathways of suffruticosol A (3) in negative mode

 $\textbf{Fig. 8} \ \ \text{Proposed fragmentation pathways of suffruticosol } C \ (\textbf{4}) \ \text{in positive mode}$



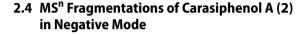
Fig. 9 Proposed fragmentation pathways of suffruticosol C (4) in negative mode

2.2 MSⁿ Fragmentations of (–)-7,8-*cis-ε*-Viniferin (1) in Negative Mode

Similar to that in positive mode, the $[M-H]^-$ ion **1b** gave rise to ions at m/z 359 (**1c**) and m/z 347 (**1d**) by the loss of a C_6H_6O or a C_7H_6O moiety. When ion **1c** was chosen to perform MS³ experiment, three product ions at m/z 289 (**1f**), m/z 265 (**1h**) and m/z 253 (**1i**) were obtained, corresponding to the elimination of $C_3H_2O_2$, C_6H_6O and C_7H_6O moieties [15–17]. The MS³ study on **1d** provided ions **1i** (m/z 253) and **1j** (m/z 225) in accordance with the departure of C_6H_6O and $C_7H_6O_2$ parts. The ions at m/z 273 (**1g**) and m/z 197 (**1k**) were tentatively deduced from **1f** and **1j** by the loss of an O or a CO moiety. Due to the presence of hydroxy groups in the structure, the elimination of H_2O from **1d** generated **1e** (Fig. 3).

2.3 MSⁿ Fragmentations of Carasiphenol A (2) in Positive Mode

The $[M+H]^+$ ion **2A** gave rise to fragments at m/z 385 (**2B**), m/z 335 (2D) and 323 (2E) by the elimination of C_2H_4O , C_6H_6O and C_7H_6O moieties [15–17]. The ion **2E** might undergo a rearrangement and further loss a C₂H₂O part to provide 2J (m/z 281). By the loss of a CH₂ part from 2B, the ion at m/z 371 (2C) was yielded, and further provided ion **2K** (m/z 277) by the loss of a C_6H_6O moiety. When the ion at m/z 309 (2G) was chosen for MS³ experiment, diverse ions at m/z 291 (2I), 267 (2M), 215 (2O) and 199 (2P) were obtained. The ion at m/z 269 (2L) was deduced from 2G by the elimination of C₃H₄ part. The subsequent MS³ study on **2L** generated ions at m/z 241 (**2N**), 175 (**2S**) and 147 (**2T**). With the elimination of H_2O , C_2H_2O or $C_9H_8O_2$ part from **2D**, three ions at m/z 317 (**2F**), 293 (**2H**) and 187 (**2Q**) were obtained. The ion at m/z 183 (2R) was tentatively deduced from **2H** by the loss of resorcinol moiety (Fig. 4).



The MS² experiment on [M–H]⁻ ion generated prolific fragments at m/z 385 (**2b**), 369 (**2c**), 343 (**2d**), 333 (**2e**), 321 (**2f**), 307 (**2g**) and 267 (**2k**). The following MS³ experiment on **2b** and **2c** gave rise to **2i** (m/z 291) and **2j** (m/z 275), respectively, corresponding to the neutral loss of a phenol moiety. The ions at m/z 301 (**2h**) and 223 (**2m**) were produced from the precursors **2d** and **2e**, by the elimination of C₂H₂O and C₆H₆O₂ [15–17]. When ion at m/z 307 (**2g**) was performed the MS³ study, three ions at m/z 265 (**2l**), m/z 213 (**2n**) and m/z 187 (**2o**) were formed. The ion **2p** (m/z 145) was affirmed from the precursor **2l** by the loss of a molecule of C₈H₆O (Fig. 5).

2.5 MSⁿ Fragmentations of Suffruticosol A (3) in Positive Mode

The MS² study on [M+H]⁺ ion gave rise to the fragments at m/z 587 (**3B**), 575 (**3C**), 493 (**3D**), 481 (**3E**) and 321 (**3J**). The production of ion **3E** (m/z 481) was verified as the successive elimination of a C_7H_6O and a C_6H_6O part from **3A**. The ion **3E** could further generate ion at m/z 387 (**3G**) and 371 (**3H**) by the neutral loss of a phenol (C_6H_6O) and a resorcinol ($C_6H_6O_2$) moiety [15–17]. With the elimination of a molecule of H_2O , the ion at m/z 369 (**3I**) was obtained from **3G**. Similarly, the fragment **3F** was produced from **3D** by the loss of a phenol (C_6H_6O) part (Fig. 6).

2.6 MSⁿ Fragmentations of Suffruticosol A (3) in Negative Mode

When the $[M-H]^-$ ion was chosen for MS² study, the product ions at m/z 585 (**3c**) and 573 (**3d**) were generated due to the loss of C_6H_6O and C_7H_6O moieties [15–17]. The following MS³ investigation on ion **3c** gave rise to fragments at m/z



543 (**3e**), 491 (**3f**), 479 (**3g**) and 475 (**3h**), which could be explained by the elimination of C_2H_2O , C_6H_6O , C_7H_6O and $C_6H_6O_2$. The ion **3j** (m/z 385) was deduced from **3g** by the neutral loss of a molecular of phenol (C_6H_6O) moiety. Two MS³ fragments at m/z 357 (**3k**) and 341 (**3l**) were obtained from **3i**, which were well in accordance with the departure of a phenol (C_6H_6O) and a resorcinol ($C_6H_6O_2$) parts (Fig. 7).

2.7 MSⁿ Fragmentations of Suffruticosol C (4) in Positive Mode

The [M+H]⁺ ion of **4A** gave rise to MS² fragments at m/z 587 (**4B**) and 575 (**4C**) due to the neutral loss of a molecule of C_6H_6O and C_7H_6O parts. The successively loss of two C_6H_6O moieties was further observed in the MS³ experiment on ion **4B**, and thus gave rise to the fragments at m/z 493 (**4D**) and 399 (**4G**). The ions **4E** (m/z 481) and **4H** (m/z 371) were generated from **4C** by the successive loss of a C_6H_6O and a $C_6H_6O_2$ moieties [15–17]. The ion at m/z 453 (**4F**) was deduced from **4E** by the elimination of a molecule of CO, and further gave rise to **4I** (m/z 359) and **4J** (m/z 265) which was well explained by the consecutive loss of two phenol (C_6H_6O) moieties. The ion at m/z 371 (**4H**) was generated from **4E** by the departure of a $C_6H_6O_2$ moiety (Fig. 8).

2.8 MSⁿ Fragmentations of Suffruticosol C (4) in Negative Mode

In the negative MS^2 experiment, the neutral loss of C_6H_6O from the $[M-H]^-$ ion (**4b**) gave rise to the fragment at m/z 585 (**4c**). When the ion **4c** was chosen to perform the MS^3 study, diverse ions at m/z 543, 491, 479, 451 and 447 were obtained. The ions **4e** and **4f** were well consistent with the elimination of C_6H_6O and C_7H_6O parts from the precursor **4c** [15–17]. However, the formation of ions **4d** (m/z 543), **4g** (m/z 451) and **4h** (m/z 447) was difficult to explain due to the complicated structure (Fig. 9).

3 Experimental

3.1 Apparatus and Analytical Conditions

All of the MSⁿ experiments were performed on the LCMS-IT-TOF mass spectrometer (Shimadzu, Kyoto, Japan). Accurate masses were calibrated using sodium trifluoroacetate (CF₃CO₂Na) clusters. MS experiments were performed in automatic pattern, and MSⁿ experiments were achieved in direct mode. The MS parameters are in accordance with the previous report [18].

3.2 Chemicals and Samples

Acetonitrile (CH₃CN) of HPLC grade was purchased from Merck Co., Ltd., Germany, and formic acid was bought from Aladdin Chemistry Co., Ltd., China. Deionized water was purified using a MingCheTM-D 24UV Merck Millipore system (Merck Millipore, Shanghai, China). Compounds **1–4** were isolated from the seeds of *Paeonia lactiflora* Pall. in our previous investigation. Samples were diluted in MeOH at the concentration of 0.5 mg/mL.

4 Conclusion

The ESI multi-stage mass spectra (MSⁿ) of four oligostilbenes were studied for the first time by LCMS-IT-TOF, by which their fragmentation pathways were deduced. The consecutive elimination of phenol (C₆H₆O) and resorcinol (C₆H₆O₂) moieties from the precursor ions was the particular dissociation due to the presence of 1,2-diphenylethylene nucleus in the structure. Interestingly, the elimination of a C₇H₆O moiety was always detected due to the fracture of the double bond in 1,2-diphenylethylene nucleus, and this fragmentation pathway might be impelled by the rearrangement of the free radical into a stable conjugated system (e.g. tropone). Based on the fragmentation rules deduced above, (-)-7,8-cis- ε -viniferin (1), carasiphenol A (2), suffruticosol A (3) and suffruticosol C (4) could be well differentiated by their respective ion pars of 455-215, 429-267, 681-321 and 681-359 in positive mode, and 453–359, 427–307, 679–451 and 679–447 in negative mode. The present MSⁿ fragmentation study will provide valuable information for the fast characterization of oligostilbenes from complicated natural mixtures.

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Compliance with Ethical Standards

Conflict of interest These authors have no conflict of interest to declare.

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References

- C. Riviere, A.D. Pawlus, J.M. Merillon, Nat. Prod. Rep. 29, 1317–1333 (2012)
- T. Shen, X.N. Wang, H.X. Lou, Nat. Prod. Rep. 26, 916–935 (2009)
- P. Langcake, C.A. Cornford, R.J. Pryce, Phytochemistry 18, 1025–1027 (1979)
- J.A. Sirerol, M.L. Rodríguez, S. Mena, M.A. Asensi, J.M. Estrela, A.L. Ortega, Oxid. Med. Cell. Longev. 2016, 1–15 (2016)
- 5. X.F. Wang, C.S. Yao, J. Asian Nat. Prod. Res. 18, 376–407 (2016)
- T. Szkudelski, K. Szkudelska, Biochim. Biophys. Acta 1852, 1145–1154 (2015)
- T. Ahmed, S. Javed, A. Tariq, D. Šamec, S. Tejada, S.F. Nabavi, N. Braidy, S.M. Nabavi, Mol. Neurobiol. 54, 2622–2635 (2017)
- G.Z. Xin, J.L. Zhou, L.W. Qi, P. Li, Comb. Chem. High Throughput Screen. 14, 93–103 (2011)
- L. Hoffer, J.P. Renaud, D. Horvath, Comb. Chem. High Throughput Screen. 14, 500–520 (2011)

- M. Zhou, H. Luo, Z. Li, F. Wu, C. Huang, Z. Ding, R. Li, Comb. Chem. High Throughput Screen. 15, 306–315 (2012)
- R. Li, Z.J. Wu, F. Zhang, L.S. Ding, Rapid Commun. Mass Spectrom. 20, 157–170 (2006)
- K. Ablajan, A. Tuoheti, Rapid Commun. Mass Spectrom. 27, 451–460 (2013)
- W. Niu, X.H. Zhu, K. Yu, L. Li, Y. Sun, C. Li, J. Mass Spectrom. 47, 370–380 (2012)
- L.M. Szewczuk, S.H. Lee, I.A. Blair, T.M. Penning, J. Nat. Prod. 68, 36–42 (2005)
- N. Mulinacci, M. Innocenti, A.R. Santamaria, G.L. Marca, G. Pasqua, Rapid Commun. Mass Spectrom. 24, 2065–2073 (2010)
- R. Moss, Q.Y. Mao, D. Taylor, C. Saucier, Rapid Commun. Mass Spectrom. 27, 1815–1827 (2013)
- J.B. Jean-Denis, R. Pezet, R. Tabacchi, J. Chromatogr. A 1112, 263–268 (2006)
- 18. C.A. Geng, J.J. Chen, Nat. Prod. Bioprospect. 6, 297-303 (2016)

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