



# Microbially-mediated indirect effects of silver nanoparticles on aquatic invertebrates

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## Abstract

Complex natural systems are affected by multiple anthropogenic stressors, and therefore indirect effects within food webs are increasingly investigated. In this context, dead organic matter (OM) or detritus provides a food source sustaining detrital food webs that recycle the retained energy through microbial decomposition and invertebrate consumption. In aquatic environments, poorly water-soluble contaminants, including nanoparticles (NPs), quickly adsorb onto OM potentially modifying OM-associated microbial communities. Since invertebrates often depend on microbial conditioning to enhance OM quality, adverse effects on OM-associated microbial communities could potentially affect invertebrate performances. Therefore, this study assessed the effect of environmentally relevant concentrations of the model emerging contaminant, silver nanoparticles (AgNPs), on OM-associated microorganisms and subsequent indirect effects on growth of the invertebrate *Asellus aquaticus*. At low concentrations (0.8 µg/L), AgNPs inhibited activity and altered metabolic diversity of the OM-associated microbial community. This was observed to coincide with a negative effect on the growth of *A. aquaticus* due to antimicrobial properties, as a decreased growth was observed when offered AgNP-contaminated OM. When *A. aquaticus* were offered sterile OM in the absence of AgNPs, invertebrate growth was observed to be strongly retarded, illustrating the importance of microorganisms in the diet of this aquatic invertebrate. This outcome thus hints that environmentally relevant concentrations of AgNPs can indirectly affect the growth of aquatic invertebrates by affecting OM-associated microbial communities, and hence that microorganisms are an essential link in understanding bottom-up directed effects of chemical stressors in food webs.

**Keywords** *Asellus aquaticus* · Food web · Freshwater biofilms · Decomposition and consumption · Silver nanoparticles · Ecosystem functioning

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## Introduction

Dead organic matter (OM) or detritus serves as a major food source fueling aquatic detritivorous food webs. The nutrients and energy stored in OM are made available through decomposition mediated by microbial activity and invertebrate consumption (Webster and Benfield 1986; Gessner et al. 2010). Additionally, OM-associated microbial communities can partly degrade OM through enzymatic activity (known as conditioning), and therewith form an essential trophic link by stimulating invertebrate feeding (Graça 2001; Danger et al. 2012). OM-associated microorganisms positively affect growth and reproduction of many invertebrates relative to unconditioned food sources (Graça et al. 1993a, b), a benefit that relies on the provisioning of essential nutrients such as nitrogen, some fatty acids, and vitamins (e.g. Vonk et al. 2016).

An increasing number of aquatic ecosystems are under continuous pressure from anthropogenic stressors (Goulson et al. 2014), fueling organismal research assessing direct lethal effects or molecular target pathways. Recently, efforts are increasingly directed towards assessing how contaminants can indirectly affect trophic interactions within food webs at sublethal, environmentally relevant concentrations (Holden et al. 2016; Schrama et al. 2017). Many contaminants that enter aquatic environments, including emerging compounds such as nanoparticles (NPs), are poorly soluble and therefore quickly adsorb to OM (Lu et al. 2014; Holsapple et al. 2005; Pradhan et al. 2016). Increasing evidence suggests that many such contaminants can have adverse effects on OM-associated microbial communities, with consequences for OM decomposition rates (e.g. Flores et al. 2014; Zubrod et al. 2011; Tlili et al. 2016; Du et al. 2018). Additionally, indirect effects on detritivorous food webs are possible. Pesticides, for instance, reduce OM consumption by several aquatic invertebrate species (e.g. Zubrod et al. 2015; Hunting et al. 2016), often driven by changes in the OM-associated microbial community (e.g. Feckler et al. 2016). However, while NPs were observed to negatively affect growth and diversity of OM-associated microbial communities (e.g. Tlili et al. 2016; Du et al. 2018; Pradhan et al. 2012; Batista et al. 2017a, b), bottom-up directed effects on detritivore growth remain unclear.

Many NPs have anti-microbial properties and the tendency to adsorb onto OM, and thus it can be hypothesized that NPs indirectly affect invertebrate growth by negatively affecting the OM-associated microbial communities (Holsapple et al. 2005; Pradhan et al. 2016; Fabrega et al. 2009; Du et al. 2018) or through the uptake of NPs as part of their food. We therefore assessed the effects of sub-lethal and environmentally relevant concentrations of silver nanoparticles (AgNPs) as a model emerging toxicant on (1) the activity and metabolic diversity of OM-associated microorganisms, and (2) the growth of the detritivore *Asellus aquaticus* in a well-controlled laboratory setting.

## Materials and methods

### Study system

Individuals of *A. aquaticus* were collected from ditches in the south-west of Netherlands (52°09'51.4"N 4°27'55.3"E) and sustained in a glass aquarium of 40 × 25 × 33 cm filled with a mixture of distilled water (dH<sub>2</sub>O) and ditch water (2:1 v/v; aquarium water) and a layer (1–2 cm) of quartz sand under continuous aeration.

AgNPs with a nominal particle size of 15 nm were purchased from Nanostructured and Amorphous Materials (Houston, USA). Characterizations of the particle

morphology of the AgNPs were performed using transmission electron microscopy (TEM) (JEOL 1010, JEOL Ltd., Japan) and dynamic light scattering (DLS) on a zetasizer Nano-ZS instrument (Malvern, Instruments Ltd., UK). Their physico-chemical characteristics of AgNPs during the early stages of the experiment are summarized in Table S1 and S2. AgNP's have a complex mode of toxic action, in which both AgNP's themselves and Ag ions in aquatic environments contribute to toxicity, often depending on various environmental conditions (Pradhan et al. 2011; Völker et al. 2013; Tlili et al. 2016; Zhai et al. 2016, 2017). Since this is difficult to disentangle within complex systems (e.g. natural OM) and AgNPs were merely used to distort OM-associated microbial communities, contributions of AgNPs and ions were not assessed here.

DECOTABs were used as a surrogate OM to allow manipulation of AgNP concentrations (Kampfraath et al. 2012; Van der Lee 2018). In short, DECOTABs were prepared from 60 g/L powdered organic hay serving as particulate OM (POM) and 20 g/L purified agar (Sigma-Aldrich). An AgNP stock solution (1 g Ag/L) was sonicated (38 ± 10 kHz) for 8 min at 4 °C in a water bath to ensure a homogeneous dispersion of the particles (Zhai et al. 2016). Agar was dissolved in dH<sub>2</sub>O and heated up to 100 °C. After cooling to below 50 °C, POM was mixed together with AgNP with concentrations ranging 0.0, 0.2, 0.4, 0.8, 1.6, 3.2 and 6.4 µg/L. Subsequently, the mixture was poured into the mold. Solidified DECOTABs were stored at –20 °C. To measure the actual AgNP concentrations in the DECOTABs and the released Ag<sub>(total)</sub> (AgNP and any other speciation of Ag) from DECOTABs into the aquatic medium, DECOTABs were resuspended and digested in *aqua regia* (HNO<sub>3</sub>: HCl = 1:3) at room temperature overnight followed by the evaporation of the acid at 70 °C and then resuspended in 3 mL of 5% *aqua regia*. Also, water samples were acidified using 5% *aqua regia* before measurements. The Ag<sub>(total)</sub> concentrations in the AgNP-contaminated DECOTABs, as well as the Ag<sub>(total)</sub> released from the AgNP-contaminated DECOTABs into the aquatic medium, were determined 7 days after test initiation using Graphite Furnace Atomic Absorption Spectroscopy (GF-AAS; Perkin Elmer 1100B, The Netherlands). Actual Ag<sub>(total)</sub> concentrations within the AgNP-contaminated DECOTABs were determined in triplicate for the several concentrations (Table S3), and the percent of Ag<sub>(total)</sub> released from the AgNP-contaminated DECOTABs into the aquatic test medium was max. 11% and increased with increasing AgNP concentration in the DECOTABs (Table S3).

### Experimental design

To assess microbial metabolic diversity and activity, OM-associated microorganisms were grown on DECOTABs

containing a range of AgNP concentrations (5 DECOTABs per concentration) as described above. DECOTABs were incubated in Petri-dishes filled with 40 mL aquarium water for 21 days. Afterwards the DECOTABs were vortexed for 30 s in 5 mL dH<sub>2</sub>O to separate the developed biofilms from the DECOTABs, and the microbial metabolic diversity of the biofilms was measured using Ecoplates (Biolog, Hayward, CA, USA, Garland and Mills 1991). Ecoplates are microplates that contain 31 ecologically relevant carbon substrates, and the utilizations of the substrates reflects the metabolic potential of the tested microbial community. Although the actual functioning of the biofilm inherent microorganisms cannot be directly related to substrate utilization, the differences in the metabolic profiles can indicate the distinct effect of stressors (Hunting et al. 2015, 2017; Echavarri-Bravo et al. 2015). Therefore, the biofilms were diluted 30 times with dH<sub>2</sub>O (triplicate per replicate) and each well of the Ecoplates was inoculated with 50 µL of the dilution at 20 °C for approximately 46 h. After the incubation, optical density was measured at 600 nm using a BioTek microplate reader. Color development of substrates greater than 0.25 were included for assay evaluation as proposed by Garland and Mills (1991). The biofilms' Electron Transport System Activity (ETSA) as a proxy for bacterial activity was measured by determining spectrophotometrically the degradation of 2-(p-iodophenyl)-3-(p-nitrophenyl)-5-phenyl tetrazolium chloride to formazan at a wavelength of 490 nm according to Hunting et al. (2010).

To assess whether DECOTAB-associated microorganisms provide a nutritional source for the freshwater isopod *A. aquaticus*, DECOTABs without AgNPs were prepared for a second incubation. This procedure involved an additional set of DECOTABs, which served as a negative control. To this end, DECOTABs were kept sterile by placing them for 1 h in 70% ethanol followed by three times washing with dH<sub>2</sub>O and resting in dH<sub>2</sub>O for another 10 min. Finally, one sterilized or conditioned DECOTAB was offered as food to 5 *A. aquaticus* (average 0.17 cm in length) in Petri-dishes filled with 40 mL aquarium water for 28 days. Both treatments were run in duplicate. Sterile DECOTABs were refreshed every other day, while conditioned DECOTABs were refreshed every 5 days. Every seven days, individual *A. aquaticus* were imaged using eScope (v. 1.1.7.17) and body length measured using ImageJ (v. 1.51j8).

The third series of incubations assessed direct and indirect (food quality related) effects of AgNPs on the microbial metabolic diversity and activity, as well as the growth of the freshwater isopod *A. aquaticus* as described above. DECOTABs containing different AgNP concentrations (see above) were offered as food to 5 *A. aquaticus* (average 0.17 cm in length) in Petri-dishes filled with 40 mL aquarium water for 35 days. DECOTABs were refreshed every 5 days. Due to a constraint in the number of isopod offspring, the experiment

was split up into two sequential runs using independent juvenile *A. aquaticus* individuals, in which each treatment was run in duplicates. This finally resulted in 4 independent replicates per treatment. Microbial metabolic diversity and activity were assessed after 21 days as described above. Every 7 days, individuals of *A. aquaticus* were imaged using eScope (v. 1.1.7.17) and body length was measured using ImageJ (v. 1.51j8).

## Statistical analysis

Differences in Ecoplate (Biolog) substrate utilization between different AgNP treatments were analyzed using a Gower-based cluster analysis and a one-way analysis of similarities (ANOSIM) using PAST 3.0 to determine the bacterial functional composition (Hammer et al. 2001; Villéger et al. 2008). Isopod growth was assessed by measuring the mean increase in growth of the 5 pseudo-replicates within 1 replicate Petri-dish. Microbial activity and isopod growth were assessed in relation to increasing concentrations of AgNPs by non-linear and linear least square regression, respectively.

## Results

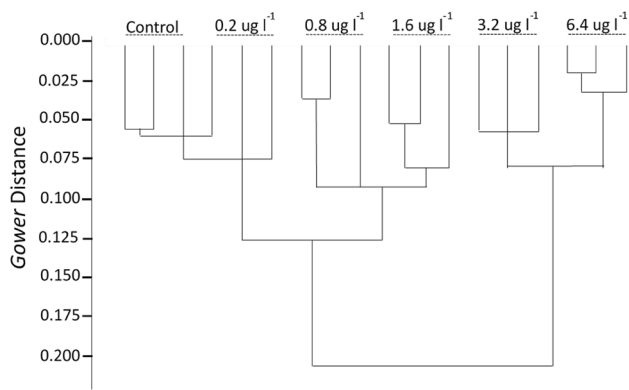
### Metabolic diversity and activity of OM-associated microorganisms

The metabolic diversity of the microbial biofilms that developed on the DECOTABs were assessed by substrate utilizations under different AgNP treatments (Fig. 1). The microbial communities in the control were in general able to metabolize 14 of the 31 carbon sources, and the utilization of most of the substrates decreased with the increasing concentrations of the AgNPs. Cluster analysis showed a clear clustering of AgNP treatments; microbial metabolic potential differed significantly (one-way ANOSIM:  $R=0.919$ ,  $p<0.05$ ) (Fig. 1). Furthermore, metabolic diversity of the DECOTAB-associated biofilm is expressed as average well colour development (AWCD; Figure S1, Supplementary Data), indicating a decreased metabolic diversity with increasing AgNP concentration.

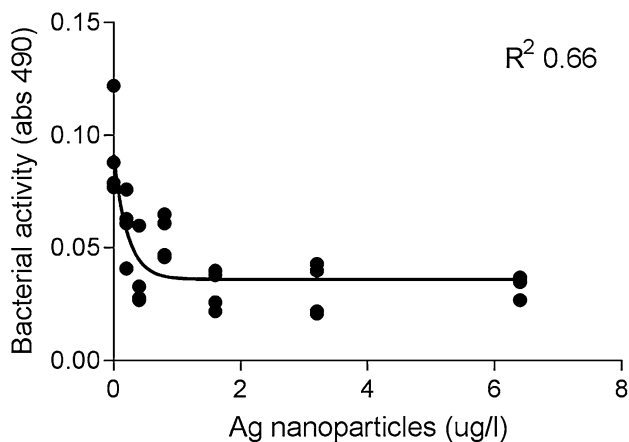
The effect of AgNPs on the activity of microorganisms within biofilms developed on the DECOTABs is provided in Fig. 2. Microbial activity decreased exponentially with increasing concentrations of AgNPs.

### Growth of *A. aquaticus*

Juveniles fed on sterilized DECOTABs exhibited a significantly smaller increase in body length compared to the juveniles fed on conditioned DECOTABs ( $F_{(1,8)}=51.6$ ,  $P<0.0001$ ;



**Fig. 1** Cluster analysis of the metabolic diversity of bacterial communities under increasing AgNP concentrations. Clustering according to the Ecoplate (Biology) substrate utilization patterns by OM-associated microbial communities (one-way ANOSIM, Gower-based similarity,  $n=3$ ,  $R^2=0.919$ ,  $p<0.05$ )



**Fig. 2** Microbial activity of the DECOTAB-associated biofilm with increasing AgNP concentration (measured at abs. 490 nm). Fitted line represents an exponential decay function; goodness of fit (adjusted  $R^2$ ) is shown in the figure

**Fig. 3** Growth of *A. aquaticus* feeding on sterile (dotted line) and conditioned (solid line) DECOTABs, in which both treatments differ significantly in their slopes ( $t$ -test,  $p<0.01$ ). Goodness of fits and significances are presented for both treatments

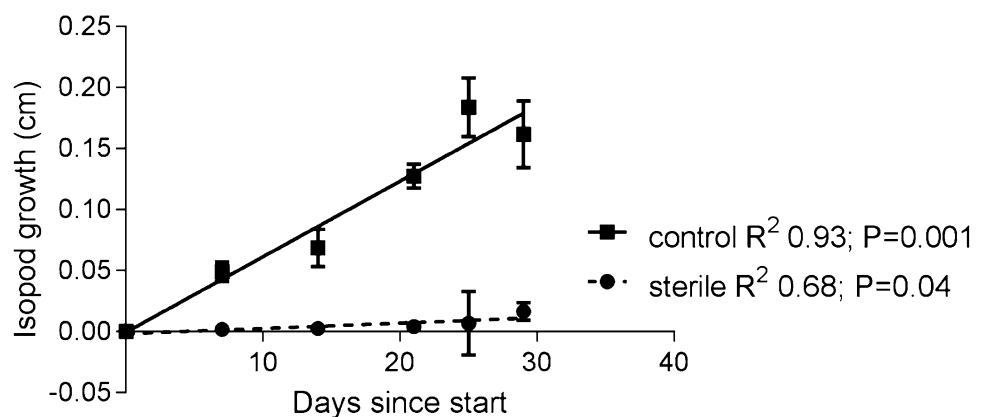
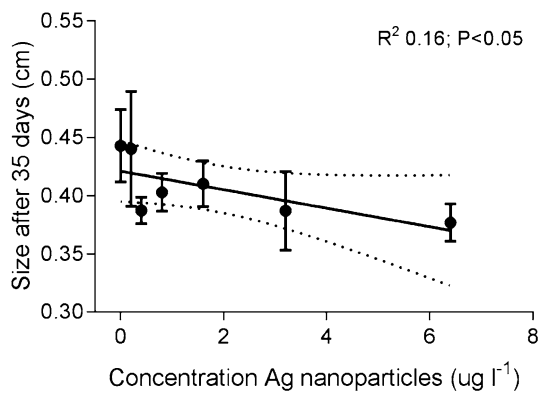


Fig. 3), with almost no growth over the entire test period in the animals feeding on the sterilized DECOTABs (Fig. 3). In general, weekly measurements of the juveniles revealed an exponential growth during the 28-day period (Figure S2, Supplementary Data), but the addition of AgNPs had an overall negative effect on *A. aquaticus* growth over the course of the experiment (Fig. 4), where nominal concentrations  $\geq 0.8$   $\mu\text{g/L}$  were significantly different to the control ( $p<0.05$ ).

## Discussion

This study used AgNPs as a model emerging toxicant to illustrate indirect effects on the growth of an aquatic invertebrate via adverse effects on the metabolic activities of OM-associated microbial communities. The metabolic activity of the OM-associated microorganisms was negatively affected by AgNP at low and environmentally relevant concentrations with a decline in microbial activity and metabolic diversity with increasing AgNP concentration. Ag ions and AgNPs are known to generate reactive oxygen species causing damage to mitochondrial respiratory function, hyperoxidation of lipids, and proteins (Morones et al. 2005; Pal et al. 2007; Sillen et al. 2015; Dakal et al. 2016), and the observed reduction in metabolic activity likely points to a disruption in electron transport chains in both cell membranes or extracellular matrix (Trevors 1984; Hunting et al. 2015). OM-associated AgNP contamination thus constrains metabolic diversity and retards microbial respiratory activity, which itself may already retard microbial-mediated OM degradation (Tlili et al. 2016).

Microorganisms are considered an important diet of macro-invertebrates (Graça et al. 1993ab; Chung and Suberkropp 2009), supporting their growth (Findlay and Tenore 1982; Findlay et al. 1984). This observation is also supported during the present study documenting only low growth rates of *A. aquaticus* when fed with sterile relative to conditioned



**Fig. 4** Isopod size after 35 days of juvenile *A. aquaticus* individuals. Solid fit represents a linear regression, dotted lines represent 95% confidence intervals. *x*-axis represents nominal AgNP concentrations within DECOTABs

DECOTABs. Traditionally, the potential beneficial role of microorganism is attributed to their ability to decompose structural polysaccharides in OM such as leaf litter thereby reducing its toughness as well as by increasing nutritious values by concentrating essential nutrients, amino and fatty acids (Graça 2001; Danger et al. 2012). Since agar cannot be degraded by most organisms, the degradation of polysaccharides unlikely played a role, illustrating the importance of microorganisms for life history strategy (i.e., growth) of *A. aquaticus*. These observations point towards an even higher relevance of OM-associated microbes (here presumably mainly bacteria) for some leaf shredding invertebrates than previously anticipated (Chung and Suberkropp 2009).

Here we observed indirect effects of NPs in a simplified detrital food chain considering interactions between two trophic levels. Even at low concentrations, adverse effects of AgNPs on microorganisms were observed to coincide with a reduced growth of *A. aquaticus*. Likewise, adverse sub-lethal effects of OM-contamination with pesticides affected OM consumption for several aquatic invertebrate species (e.g. Flores et al. 2014; Zubrod et al. 2015; Hunting et al. 2016; Feckler et al. 2016). As commonly observed with metal NPs (Pradhan et al. 2012), we found a 1–11% release of silver ions from the OM towards the surrounding water, depending on the concentrations of AgNPs in the DECOTABs (Table S3). It is likely that waterborne metals adsorb and accumulate in leaves and OM, as observed for CuNP (Pradhan et al. 2012), and therewith fuel the dietary toxicity towards invertebrates. Overall, the waterborne concentrations were low and typically do not elicit direct effects (Baptista et al. 2015; Mckee et al. 2016), however, we cannot rule out the possibility of waterborne Ag ions contributing to the impairments in *A. aquaticus* growth. Likewise, ingestion of OM containing AgNP may have contributed to growth

impairment. Despite this, results presented here suggest that OM-associated AgNPs also can have microbially-mediated effects on the performances of aquatic invertebrates. Since contamination of food items and its inherent trophic interactions seem more subtle and sensitive than test organisms typically used to assess toxicity, our results indicate that contamination of OM with AgNPs is a relevant exposure route to consider when assessing realistic effects of NPs in the aquatic environment.

This study shows that AgNPs can inhibit the activity and alter metabolic diversity of OM-associated microbial communities and that this can coincide with a negative effect on the growth of *A. aquaticus*. Impaired growth rates may have wider implications for aquatic food webs as less biomass will be available for predatory invertebrates and fish (e.g. Rask and Hiisivuori 1985; Krisp and Maier 2005). While direct toxicity is typically observed at high AgNP concentrations (Zhao and Wang 2011; Topuz and Van Gestel 2015; Baptista et al. 2015; Mckee et al. 2016), we observed low AgNP-concentration effects on microbial communities to translate to invertebrate growth effects. This calls for further investigations and consideration of trophic interactions assessing hazards and risks of NPs for the environment. Since effects observed on a realistic exposure pathway were caused by an AgNP concentration in the ng/L range, the observed trophic cascade poses concerns provided the effective AgNP concentrations fall within existing estimates of current environmental concentrations (Gottschalk et al. 2013).

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