ALKALOIDS OF THE ROOTS OF THALICTRUM FOETIDUM

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From the roots of <u>Th. foetidum</u> L. we have previously isolated 0.3% of combined alkaloids, 0.1% of which consists of quaternary bases [1] and 0.2% of tertiary bases. The tertiary bases have been separated into phenolic and nonphenolic fractions, and two new alkaloids have been obtained. When the nonphenolic fraction was treated with ethanol, prismatic crystals of a base (I) with mp 141°-142° C (decomp.), $[\alpha]_D^{15}$ +69° (c 1.0; ethanol) deposited. The homogeneity of the base (I) was confirmed by TLC on a nonfixed layer of alumina in the chloroform—ethyl acetate (1:1) system.

UV spectrum in ethanol: λ_{max} 260, 348, m μ (log ε 4.58, 3.86); IR spectrum: 920, 1030, 1050, cm⁻¹ (CH₂O₂).

In the NMR spectrum of (I) taken on a JNM-100/100 MHz instrument in deuterochloroform (τ -scale) signals are observed for the protons of a N-methyl group at 7.80, for four methoxy groups at 6.60, 6.50, 6.39, and 6.24, and for a methylenedioxy group at 4.07 ppm. mol. wt. 648 (mass spectrometry).

The data given above and the elementary analysis enable us to establish for the base (I) the composition $C_{38}H_{36}O_8N_2$ and the developed formula $C_{32}H_{19}N(N-CH_3)(OCH_3)_4(CH_2O_2)(-O-)_2$. We have called the base thalfine.

The mother liquor after the isolation of thalfine was chromatographed on alumina. The benzene fractions (16-30) showed two spots, one of which corresponded to thalfine. Treatment with ethanol isolated thalfine, and from the mother liquor needle-like crystals of a perchlorate were obtained with mp 234°-235° C (decomp.), $[\alpha]_D^{21} + 135°$ [c 1.16; ethanol – water (2 : 1)]. The hydrochloride formed prisms with mp 223°-226° C (decomp., water). Decomposition of the perchlorate gave an amorphous base (II) with mp 117°-118° C, $[\alpha]_D^{16} + 115°$ (c 0.95, ethanol). UV spectrum in ethanol: λ_{max} 282 mµ (log ε 3.76). IR spectrum: 3600-3300 cm⁻¹ (hydroxy group), 1030, 920 cm⁻¹ (methylenedioxy group).

The NMR spectrum of the base (II) showed signals for protons of two N-methyl groups at 7.70 and 7.46 ppm, of four methoxy groups at 6.66, 6.57, 6.34, and 6.20 ppm, and of a methylenedioxy group at 4.20 ppm.

On the basis of the above data and its elementary analysis, substance (II) has the composition $C_{39}H_{44}O_8N_2$ and the developed formula $C_{32}H_{24}$ (N-CH₃)₂ (OCH₃)₄ (CH₂O₂) (OH) · (-O-). We have called the base II thalfinine.

REFERENCE

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THE STRUCTURE OF ERVINCININE

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Continuing our investigation of the alkaloids of the epigeal part of Vinca erecta Rgl. et Schmalh., we have isolated vincaridine and a new base-ervincinine [1, 2]. A comparison of the UV, IR, NMR, and mass spectra, and some of the chemical properties of vincaridine and picrinine has established the identity of these substances [3].

Ervincinine has the composition $C_{22}H_{26}O_4N_2$, mol. wt. 382 (by mass spectrometry), mp $247^{\circ}-248^{\circ}$ C (decomp.) from methanol), $[\alpha]_D^{22} - 80.5^{\circ}$ (c 0.39; chloroform). UV spectrum: λ_{max} (ethanol) 250, 330 mµ (log ε 4.08, 4.18), which is characteristic for α -methyleneindoline alkaloids. IR spectrum: 3265 cm⁻¹ (NH-), 1685 cm⁻¹ (ester carbonyl group conjugated with a double bond), and 840 cm⁻¹ (1, 2, 4-trisubstituted benzene ring).