

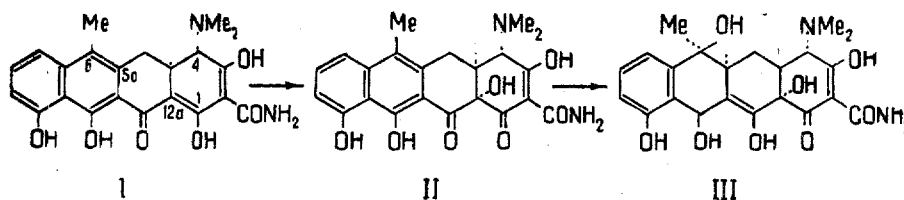
INVESTIGATIONS IN THE FIELD OF TETRACYCLINES

XLIII. Partial Synthesis of Anhydrotetracycline

A. I. Gurevich, M. G. Karapetyan, and M. N. Kolosov

Khimiya Prirodnikh Soedinenii, Vol. 2, No. 2, pp. 141-142, 1966

Anhydrotetracyclines, which are the first products of the acid degradation of the tetracycline antibiotics, are of great interest as direct biosynthetic representatives of the natural tetracyclines [1] and as intermediates for the complete synthesis of the antibiotics of this group [2]. In view of this, we have undertaken the partial synthesis of 5a, 6-anhydrotetracycline (II), starting from the known [3] 12a-deoxy-5a, 6-anhydrotetracycline (I).



The 12a-hydroxylation of the deoxyanhydro compound (I) was effected by catalytic oxidation with oxygen at 20° C in the presence of platinum (from PtO₂, prepared by a published method [4]), MnCl₂ or CeCl₃ (cf. [5, 6]). The best results were obtained with a platinum catalyst (350 mg/mmole) in anhydrous tetrahydrofuran (70 ml/mole) containing triethylamine (15 mmole/mmole). The hydroxylations lasted 6-8 hr. After purification of the reaction product by countercurrent distribution [120 transfers in the system ethyl acetate-heptane (4:1), citrate-phosphate buffer with pH 8.0] a substance was isolated with mp (micro) 225°-228° C (decomp., from benzene) [α]₅₇₈²⁰ + 50 ± 5°, [α]₅₄₆²⁰ - 5 ± 5° (c 0.3; dimethylformamide); on descending chromatography on Whatman No. 2 paper impregnated with Trilon B, R_f 0.87 [in the acetic acid-butanol-water (1:4:5) system] and R_f 0.56 [in the butanol-4% aqueous ammonia (2:3) system]; in 0.01 N methanolic HCl, λ_{\max} : 223, 273, 298 inflexion, 310 (inflexion), 324 (inflexion), 427 m μ (log ϵ 4.49, 4.74, 3.90, 3.72, 3.37, 3.94); ν_{\max}^{KBr} 1510, 1573, 1626, 1648, 3320, 3380 cm⁻¹.

A direct comparison of this substance with the anhydrotetracycline obtained by the acid dehydration of natural tetracycline [7] showed their complete identity. Thus, in contrast to the 12a-hydroxylation of 12a-deoxytetracycline described previously [5], the conversion of (I) into (II) is not accompanied by the epimerization of the asymmetric center at C₄, but leads to the "natural" configuration of ring A.

Since anhydrotetracycline (II) has recently been rehydrated to tetracycline [8], the conversion that we have effected also forms a partial synthesis of the main antibiotic of this group, tetracycline (III).

REFERENCES

1. J. R. D. McCormick, et al., J. Am. Chem. Soc., 84, 3023, 1962.
2. A. I. Scott and C. T. Bedford, J. Am. Chem. Soc., 84, 2271, 1962.
3. A. Green and J. H. Boothe, J. Am. Chem. Soc., 82, 3950, 1960; R. K. Blackwood, G. G. Rennhard, and C. R. Stephens, J. Am. Chem. Soc., 5194, 1960.
4. V. L. Frampton, J. D. Edwards, and H. R. Henze, J. Am. Chem. Soc., 73, 4432, 1951.
5. H. Muxfeldt, G. Buhr, and R. Bangert, Angew. Chem., 74, 213, 1962.
6. L. H. Conover, et al., J. Am. Chem. Soc., 84, 3222, 1962; (Brit. pat. no. 947 601; German Fed. Rep. pat. no. 1 163 318; Hungarian pat. no. 149 704.)
7. C. R. Stephens, et al., J. Am. Chem. Soc., 76, 3568, 1954.
8. M. Schrach von Wittenau, J. Org. Chem., 29, 2746, 1964; Belgian pat no., 631 118, (C. A., 61, 1815a).

28 October 1965

Institute of the Chemistry of Natural Compounds, AS USSR