
Reactive Nitrogen and Oxygen Species: Role and Evidence of their Production in Humans

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Introduction

For a long time, it has been recognized that the effects of excessive production of reactive oxygen species leading to oxidative stress are damaging to cellular constituents with subsequent tissue necrosis. There is now evidence that reactive oxygen species can alter cell functions by acting as 'second messengers' [1] or by affecting proteins of the signal transduction pathway that leads to gene expression or to apoptosis [2–4]. But many questions remain about the role of reactive oxygen and nitrogen compounds: how and where do they act? Do they act by themselves or by way of secondary products like nitroso- or nitrated compounds, oxidized forms of thiol compounds such as GSSG (the oxidized form of glutathione), α,β -unsaturated aldehydes or carbonylated proteins? Do they alter specific proteins of the signal transduction pathway or intervene in Ca^{2+} metabolism? Do they act at the level of the nucleus by oxidation of thiol proteins involved in the activation of gene expression? Do specific reactive oxygen species (ROS) such as nitric oxide (NO) or peroxynitrite, have specific activity on selected biomolecules? And, above all else, what can be transposed from *in vitro* studies or laboratory models to *in vivo* situations?

Redox Balance and Oxidative Stress

As early as 1973, it was determined that $\pm 2\%$ of molecular oxygen (O_2) reduced by the mitochondrion forms superoxide anion (O_2^-) and its dismutation product hydrogen peroxide (H_2O_2) [5]. This observation, ignored for a long time, is now regaining importance with the development of molecular biology studies on signal transduction and apoptosis. A clear distinction has to be made between oxidative stress and redox modifications by 'physiological doses' of reactive nitrogen and oxygen species (RNOS). Most of the studies dealing with the ROS in signal transduction pathways were done with hydrogen peroxide at doses $\geq 10^{-4}$ M, which are concentrations unlikely to be reached *in vivo* except in (or in the neighborhood) of activated phagocytes.

The redox state (E_0) of a compound is defined by its capacity to capture (oxidant compound) or to release (reducing compound) one or more electrons. A normal biological milieu is characterized by a redox equilibrium between oxidants and reductors. Despite difficulties in the precise measurement of this redox equilibrium, cytosol seems to be a reducing milieu in physiological conditions, by the presence of

glutathione (GSH), NAD/NADH and NADP/NADPH, particularly in leukocytes [6]. In physiological conditions, enzymatic and non-enzymatic antioxidants balance the production of ROS, delaying oxidation in the cellular milieu and heavy alterations of redox status. But, this balance is altered by excessive activation of NADPH-oxidase in stimulated leukocytes. *In vivo*, an excessive production of RNOS is possible in inflammatory pathologies characterized by an intense activation of leukocytes, or in unusual situations such as massive irradiation. In these situations, when the antioxidant buffering capacity is overloaded or pathologically depressed, oxidative damage is observed and in these conditions an oxidative stress is possible.

Endogenous Origins of RNOS

RNOS are small size molecules, most often highly unstable particularly when they are in a radical form. They rapidly diffuse and react with neighboring molecules in an unspecific manner. The detection and identification of RNOS remains difficult, which has often shed some doubt on their exact site of production and on the importance of their production. Nonetheless, specific enzymes are responsible for their production and for their destruction, and most of these enzymes have been identified and characterized. They have been localized in many cell types, so that RNOS production appears to be a general *in vivo* phenomenon, tightly regulated and linked to normal cell metabolism. ROS production has been particularly studied in lymphocytes, monocytes, macrophages and neutrophils, these latter being the main producers of RNOS in stimulation conditions.

To summarize, it can be assumed that RNOS are produced mainly by seven pathways:

- 1) by NADPH-oxidase. This enzyme is a membrane-associated enzymatic complex which is responsible for the high oxygen consumption (the respiratory burst) that occurs during phagocytosis. NADPH is present in a resting form, but is quickly activated by soluble mediators and specific stimuli acting on membrane receptors. This enzyme uses NADPH as an electron donor to turn molecular oxygen to superoxide anion (O_2^-) by a monoelectronic reduction. H_2O_2 derives from superoxide anion. Hydroxyl radical ($\cdot OH$) can be formed from H_2O_2 in the presence of complexed iron ions (Fenton reaction), but the *in vivo* production of $\cdot OH$ by the Fenton reaction still remains unproved. H_2O_2 is also the substrate of the neutrophil enzyme myeloperoxidase (MPO), which (in presence of chloride anion) synthesizes the oxidant hypochlorous acid (HOCl). These ROS are released in the phagosomes and in the extracellular milieu, but also in the cytosol where they modify the redox status of the cell [7]
- 2) by the electron transport chain of mitochondria. The mitochondrial enzymes normally produce ROS as byproducts of the electron transport chain that normally transforms molecular oxygen into H_2O by a 4 electron reduction [5]. Leakages in the electron transport chain result in the monoelectronic reduction of molecular oxygen into superoxide anion, precursor of H_2O_2 . Two sites responsible for electron leakage have been identified, the NADH dehydrogenase and the intersection ubiquinone-cytochrome b. Hypoxia appears to severely increase the production of RNOS in mitochondria by an enhanced activity of monoelectron-

- ic reduction of oxygen and NO synthase (NOS) activity. Mitochondria are equipped with a characteristic Cu-Mn²⁺ superoxide dismutase (SOD) that neutralizes superoxide anion, preventing its reaction with other ROS, especially NO, limiting the *in situ* production of peroxynitrite. This SOD is quickly inducible to assume mitochondrial defense, especially in hyperoxic and hypoxic conditions.
- 3) by the metabolism of arachidonic acid. Arachidonic acid metabolism is regulated by the family of lipoxygenase and prostaglandin H (PGH) synthase enzymes. The PGH synthases function as cyclooxygenase and hydroperoxidase, going through radical steps that are responsible for secondary reactions generating superoxide anion and other ROS. Like cyclooxygenases, lipoxygenases can produce superoxide anion as side products of their enzymatic activity.
 - 4) by cytochrome P450. The primary physiological role of the cytochrome P450 family is to detoxify lipid-soluble drugs and their metabolites. The enzymes of the cytochrome P450 family are ubiquitous. They use high-energy electrons transferred from NADPH to add hydroxyl groups to hydrophobic molecules associated with lipid bilayers. Both superoxide anion and H₂O₂ are produced during the catalytic cytochrome P450 activities. Members of this superfamily of enzymes are present in leukocytes and tissues (liver, lung) and may be physiologically important sources of ROS.
 - 5) by xanthine-oxidase. This cytosolic enzyme converts xanthine or hypoxanthine to uric acid, and is derived from xanthine dehydrogenase by the oxidation of thiol groups or limited proteolysis that occurs during hypoxia/reoxygenation. Xanthine oxidase uses molecular oxygen as electron acceptor and generates O₂⁻ and H₂O₂. Xanthine oxidase activity has been demonstrated in many tissues and cells including alveolar cells and leukocytes (especially macrophages and neutrophils). Its activity would be regulated by NO, possibly implicating the production of peroxynitrite.
 - 6) by NOS. This enzyme (in its constitutive or inducible isoforms) generates NO which appears to possess variable pro- or anti-inflammatory cellular effects. This enzyme is expressed in many cells and its inducible isoform is expressed in macrophages, neutrophils and endothelial cells. Unregulated NO production seems to have detrimental effects, which can be largely attributed to the production of peroxynitrite (ONOO⁻) formed by the reaction of NO with O₂⁻. ONOO⁻ is a highly toxic compound able to modify signal transduction pathway by oxidation or nitrosylation. Other effects of NO would be the generation of iron-nitrosyl complexes with FeS-enzymes (enzymes of mitochondria), the inactivation of antioxidant enzymes such as catalase, GSH-peroxidase, and SOD. NO would also induce apoptosis by increasing the activity of caspase-3 and by induction of the permeability transition in mitochondria.
 - 7) by several other sources such as the autoxidation of ascorbic acid, thiols, norepinephrine, and flavine coenzymes.

A Newcomer on the Set of RNOS: Peroxynitrite

In vivo Production

In vivo, ONOO⁻ is formed from the reaction of NO with O₂⁺. The first evidence of this synthesis is based on the observation that the presence of SOD prolonged the lifetime of NO[•] by removal of O₂⁺. The constant of the reaction of NO[•] with O₂⁺ is higher than the constant of the reaction of SOD with O₂⁺ [8], indicating that the production of ONOO⁻ *in vivo* is highly probable despite the presence of SOD, when NO and O₂⁺ are formed simultaneously, and particularly when NO is produced in large quantities.

NOS is present in many cells that also possess the enzymatic equipment responsible for O₂⁺ synthesis. The existence of ONOO⁻ in biological systems was demonstrated for the first time by Beckman et al [9]. Since then, the production of ONOO⁻ has been identified in tissues, in cells, and even out of the cells. In macrophages and neutrophils, ONOO⁻ would be responsible for the major part of the luminol-dependent chemiluminescence observed when these cells are stimulated [10]. ONOO⁻ produced by the phagocytes appears as a major cytotoxic agent released during sepsis and used for the host defense against invading microorganisms, but also produced in excess in inflammation and in ischemia/reperfusion.

Concentrations of NO in the endothelium have been estimated to be between 100 to 400 nM, but when endothelial cells are stimulated (by endotoxins for example) or in pathological conditions of ischemia-reperfusion, concentrations of 0.5 μM NO have been measured on the surface of the endothelial cells [11]. By luminol-dependent chemiluminescence, an immediate oxidant production by stimulated aortic endothelial cells was observed that was efficiently inhibited by the NOS inhibitor, nitro-L-arginine methyl ester, and by SOD, implying dependence on the presence of both NO[•] and O₂⁺, and thus of ONOO⁻, for oxidant production. Vascular endothelium would also generate peroxynitrite in response to carbon monoxide [12], which is a byproduct of endothelial heme oxygenase activity, an enzyme induced by the exposure of the endothelial cells to free hemoglobin.

It also appears that NO is produced inside the mitochondria where it plays a role in the regulation of the respiration by its binding to cytochrome c oxidase [13]. As mitochondria normally form O₂⁺, albeit at a low level, the production of ONOO⁻ is possible, a production that could be enhanced during and after ischemia when the damaged mitochondrial respiratory chain uncompletely reduces O₂ and releases O₂⁺.

It is likely that, in stimulating and pathological conditions, ONOO⁻ could be formed in most of the cells which contain inducible or constitutive NOS and enzymes synthesizing O₂⁺ such as NADPH oxidase and xanthine oxidase, and that ONOO⁻ is particularly formed in mitochondria submitted to pathological conditions such as ischemia reperfusion. Endotoxins were reported to trigger the formation of ONOO⁻ in the lungs [14], data from which it could be extrapolated that the production of ONOO⁻ is highly probable in sepsis.

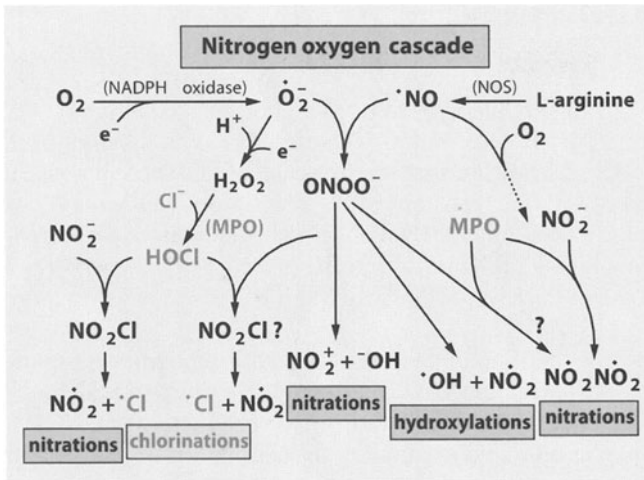


Fig. 1. Schematic overview of the mechanisms of production of peroxynitrite (ONOO⁻) and its reaction with other reactive oxygen species.

NOS: NO synthase; MPO: neutrophil myeloperoxidase; O₂⁻: superoxide anion; ·NO₂: nitryl radical; NO₂⁻: nitrite ion; NO₂⁺: nitronium ion; NO₂Cl: nitryl chloride; ·Cl: chlorine radical; HOCl: hypochlorous acid; ·OH: hydroxyl radical; ·OH: hydroxyl anion

Interactions of Peroxynitrite with Myeloperoxidase in Phagocytes

In neutrophils, a net of interactions is possible between the ROS and the RNOS (Fig. 1). The enzymatic equipment of neutrophils allows them to synthesize NO[·] by their NOS, O₂⁻ by their NADPH-oxidase and HOCl by their myeloperoxidase using H₂O₂ derived from the important flux of O₂⁻ produced by the stimulated neutrophils. They also produced ONOO⁻ from the diffusion-controlled reaction between NO and O₂⁻, and this ONOO⁻ can be reduced to NO₂⁻. HOCl may react with this latter molecule to form nitryl chloride (NO₂Cl) which can undergo heterolysis to nitryl cation (NO₂⁺) responsible for nitration. There is also a possibility of ·NO₂ production (and thus of nitration) by the activity of myeloperoxidase using NO₂⁻ or reacting directly with peroxynitrite [15, 16]. In 1996, a nitration of bacteria ingested by cytokine-stimulated neutrophils was observed [17], confirming that these cells produce ONOO⁻ as an oxidant in host defense. However, in situations of excessive activation and degranulation, ONOO⁻ and myeloperoxidase can be released out of the cells, damaging the neighboring cells and tissues. It was demonstrated that human phagocytes and lymphocytes produce ONOO⁻ when they are challenged with lipopolysaccharide [18].

Role of RNOS

ROS are powerful antimicrobial agents. Their production by stimulated phagocytes allows the destruction of microorganisms into the phagolysosome, but ROS are also damaging agents for the neighboring cells and tissues when they are produced

in excess [19]. They are especially damaging for the unsaturated lipids of membranes, of which the permeability is altered by lipoperoxidation. Lipoperoxidation produces stable lipid-derived aldehydes (such as hydroxynonenal) able to diffuse and react with amino acids and that seem to be responsible for immunosuppression. Nucleic acids are altered by ROS (reaction with thymine, guanine and deoxyribose), leading to DNA breaks and mutations. ROS are also responsible for depolymerization of hyaluronic acid, for protein alterations (tryptophan, histidine and cysteine are frequent targets) and for zymogen activation. HO[•] and lipid radicals are highly reactive by their radical nature, but HOCl and to a lesser extent H₂O₂ are also dangerous oxidant species by their longer lifetime, what allows them to reach cellular parts distant from their site of production.

The small quantities of NO biosynthesized by the constitutive isoforms of NOS have important physiological roles in cardiovascular, nervous, immune, and inflammatory systems. There is little or no evidence of cytotoxic effects of physiological amounts of NO, but in special cases, neuronal NOS can generate sufficient amounts of NO to damage target cells in the central nervous system. Inducible NOS can generate large quantities of NO leading to profound and life-threatening hypotension, as observed in sepsis, or associated with widespread cytotoxicity (cytostatic or cytotoxic actions) not only for invading microorganisms [20], but also for host cells. The mechanisms of this cytotoxicity remain a hotly debated topic. NO may directly interact with protein-bound iron and iron-sulfur causing their destruction or inactivation. It also reacts with oxygen to form N₂O₃ and to release NO₂⁻, an oxidant species that can be used by myeloperoxidase or react with HOCl to produce nitrating (and chlorinating) species. But the most popular mechanism for NO cytotoxicity implicates ONOO⁻, derived from NO and O₂^{•-}.

Despite its instability, the lifetime of ONOO⁻ seems sufficient to permit reaction at long distance, crossing through the cell membranes and reaching the DNA [21]. With a model of microbeads covered with macrophages, it has been demonstrated that NO rapidly reacts with oxygen to form N₂O₃ and to release NO₂⁻, and with O₂^{•-} to form ONOO⁻, which could reach distance of ± 30 μm before being decomposed [22]. Albeit short, this distance allows ONOO⁻ to be cytotoxic for the producer cells themselves and for the neighboring cells. Numerous *in vivo* reactions of ONOO⁻ have been reported, with enzymes and biomolecules, but the exact mechanisms of these reactions, just as their consequences, most often remain incompletely understood. ONOO⁻ can react with SOD yielding a stable yellow protein adduct (identified as nitrotyrosine) and releasing copper [23].

As ONOO⁻ attacks heme molecules, its reaction with the heme enzymes is thus expected. Recent data confirmed that ONOO⁻ reacts with the enzymes of the mitochondrial respiratory chain, with cyclo-oxygenase, with peroxidases and with NOS. Peroxynitrite also oxidizes tetrahydrobiopterin, a cofactor of NOS [24]; it thus plays a role in the regulation of NO[•] production. The reaction of ONOO⁻ with amino acids and proteins generates spontaneous chemiluminescence, that appears to be due especially to the attack of tryptophan residues [25]. We demonstrated that the reaction of ONOO⁻ with tryptophan was accompanied by a light emission and by the generation of free radical species [26]. We also demonstrated that ONOO⁻ reacted in a similar manner (with light emission and generation of free radical species) with other heterocyclic molecules such as histamine and serotonin, a tryptophan-re-

lated compound. Tyrosine, free or bound in proteins, appears to be a main target for ONOO⁻, by a reaction yielding nitrotyrosine and catalyzed by SOD [27]. Direct evidence of radical reaction of ONOO⁻ with tyrosine in plasma was furnished by electron spin resonance (ESR) detection of peroxynitrite-induced-tyrosine centered protein radicals [28]. However, the reality of tyrosine nitration at physiological pH remains debated [29].

Many other compounds are targets for ONOO⁻: glycosaminoglycans and particularly chondroitin sulfates which are essential constituents of basement membranes [30], pterines and heme molecules such as cytochrome P450 [31]. With thiol compounds, ONOO⁻ generates oxidation products, with intermediate thiyl radicals that have been detected by ESR [32]. Base modifications and strand breakages of nucleic acids by ONOO⁻ have also been described [33]. By still unclear mechanisms, ONOO⁻ seems to attack double bonds in lipids and is implicated in the peroxidation of lipoproteins (at the level of lipids and apoproteins) [34]. Until now the products of lipid oxidation by ONOO⁻ are not exactly identified, and their potential noxious effects remain unknown [35].

Except for the normal phagocytosis functions of monocytes/macrophages and neutrophils, the *in vivo* formation of ONOO⁻/ONOOH is thus undesirable. The normal cellular concentration of SOD seems sufficient to assume the complete disappearance of O₂⁻ on condition that NO is produced at nanomolar levels, in the concentration range needed for signaling functions. But, the NO concentrations produced by stimulated phagocytes are much higher, reaching the μM range, so that the *in situ* generation of ONOO⁻/ONOOH is certain.

RNOS in Signal Transduction and Apoptosis

Signal transduction is a network of phenomena resulting in the activation of gene expression when cytosolic proteins known as nuclear transduction factors have entered into the nucleus and have bound to specific DNA loci. Nuclear factors induce the expression of cell proteins that intervene in host defense as in immunity reaction and inflammation, but also to the triggering of programmed cell death known as apoptosis (often presented as cell suicide) [36]. The mechanisms by which the cell machinery is oriented to the expression of selected genes or apoptosis remain largely unexplained. Signal transduction and apoptosis are processes of homeostasis. To work correctly, they need to be tightly regulated. Alterations of the redox equilibrium in the cell medium are directly implicated in the mechanisms of signal transduction and the thiol (-SH) functions appear to play a main role. They are present in a lot of compounds that intervene in the signal transduction pathway: cell receptors, caspases, nuclear factors and even nuclear proteins that regulate the binding of nuclear factors to their corresponding DNA elements. These -SH functions are easily oxidable and are ideal targets for the RNOS.

Role of ROS

Evidence has accumulated that ROS can intervene as intracellular messengers in the regulation of signal transduction and apoptosis [1–4]. For this function, ROS have to fulfill special conditions: small size, easy diffusion capacity, ubiquity, rapid synthesis and destruction. ROS fulfill these conditions, but they are also highly toxic by their oxidative capacity, so that only low concentrations of ROS could be considered as physiological cellular messengers. A normal role of ROS would be the modification of the redox balance by the consumption of reductor molecules (GSH, NADH, NADPH) or the oxidative modification of protein structures. The oxidative alteration of proteins could allow them to interact with receptors, to modify essential enzymatic activities or to bind to specific targets on DNA and trigger the signal transduction and gene expression [37]. In leukocytes, alterations of the redox state have been demonstrated to modify signal transduction.

In 1991, Bauerle and coworkers underlined the role of oxidant compounds (H_2O_2 or ionization radiation) on the activation of the nuclear transduction factors of the nuclear factor-kappa B (NF- κ B)/Rel family [1]. Since these pioneer observations, the effects of ROS have been particularly studied on the member of this family commonly referred to as NF- κ B, which is the heterodimer p50: p65 (Rel A) [38]. NF- κ B is a unique factor by its rapidity of activation and its mechanism of regulation. It plays a major role in situations where a rapid gene induction is necessary for cell life (immunity response and inflammation) [39]. It is present in all cell types where it is present in an inactive form in the cytosol, bound to its specific inhibitor, the I κ B [38]. NF- κ B induces a large number of genes, especially the genes implicated in the immune and inflammatory responses. In most cell lines, it protects against apoptosis. NF- κ B is activated by a host of stimuli (pro-inflammatory cytokines, bacteria, lipopolysaccharide [LPS], viruses...). Its activation starts with the phosphorylation of I κ B followed by ubiquitination and leading to rapid degradation of I κ B by the proteasome. NF- κ B is released and quickly translocated into the nucleus. Regulation of NF- κ B activation is controlled mainly by the level of I κ B phosphorylation and ubiquitination, but also by the activity of the proteasome [38, 40].

Other nuclear transcription factors and several cascades of protein kinases are sensitive to ROS, and the common target for ROS appears to be a regulation domain which is activated by phosphorylation. In leukocytes, physiological concentrations of H_2O_2 are produced in response to cytokines, and H_2O_2 thus acts as second messenger [41]. The mechanisms of this activation by ROS remain unexplained. ROS would intervene by oxidation of thiol functions and glutathione would be a main target for this oxidation. The redox-sensitive transcription factors hypoxia inducible factor-1 α (HIF-1 α) and NF- κ B are tightly regulated by the GSH-GSSG balance in lungs [42]. H_2O_2 would act by favoring the phosphorylation of tyrosine kinases or by oxidation (at the level of a cystein residue) of the protein tyrosine phosphatases (PTPs) which regulate signal transduction by dephosphorylation of the protein kinases [39]. Products of lipid oxidation (oxidized low density lipoprotein [LDL], oxidized forms of cholesterol and lipid aldehydes like 4-hydroxynonenal) also act on the regulation of gene expression for adhesion molecules, heat shock proteins, cytokines, growth factors, etc. [43]. Whatever their mechanism of action, the role of ROS is confirmed by the use of antioxidant compounds (N-acetyl-cysteine) and enzymes

(catalase), which inhibit signal transduction, mainly by inhibition of tyrosine phosphorylation [44]. But many problems remain unsolved, and the role of ROS in transduction signal activation is still contested. A recent study with endothelial cells demonstrated that H₂O₂ induced the DNA binding activity of NF-κB; however, this binding was transcriptionally ‘silent’ as it could not activate intercellular adhesion molecule (ICAM-1) expression [45]. Problems also remain concerning the role of ROS/RNOS in apoptosis. After an oxidative stress induced by H₂O₂, the cell can be destroyed by apoptosis or necrosis, and this orientation would depend on the concentration of H₂O₂: at doses ≥ 1 mM (oxidative stress), the cells are moving toward necrosis, and at micromolar concentrations, the cells move toward apoptosis. This difference of evolution would be dependent on the ATP level [46].

Peroxynitrite: a Key Point in Signal Transduction Alterations?

In the last years, an increasing number of studies have implicated ONOO⁻ in the regulation of signal transduction, by damaging receptors or by intervening at other intracellular sites of the signal transduction [47, 48]. By nitration of essential tyrosine residues, ONOO⁻ can induce the dysfunction of enzymes implicated in signal transduction and apoptosis, such as tyrosine kinases (Fig. 2). Nitrated proteins would thus be not only the inactive ‘footprint’ of ONOO⁻ but also active contributors in disease processes. Tyrosine kinases must be phosphorylated on their active tyrosine site to participate to the signal transduction, and they must further be dephosphorylated to stop the signal. The enzymes that are implicated in the dephosphorylation are also tyrosine-bearing enzymes. Nitration of the tyrosine residues may alter the enzyme functions by hindering the phosphorylation of tyrosine kinases and by inactivating the dephosphorylating enzymes. Peroxynitrite may thus act

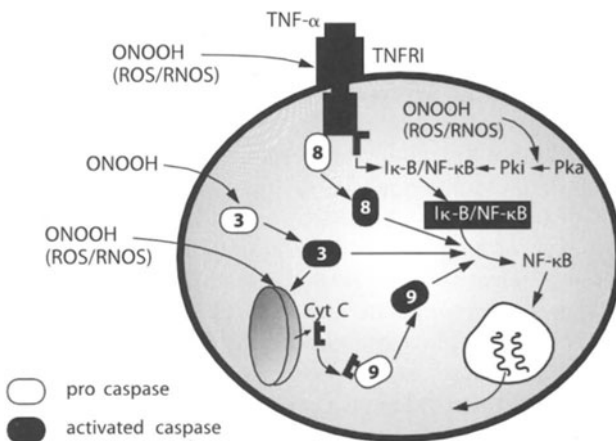


Fig. 2. Simplified view of signal transduction processes, and possible points of intervention of peroxynitrite (and reactive nitrogen-oxygen species: ROS/RNOS). *Pk*: protein kinase (*i*: inactive; *a*: active); *TNFR*: tumor necrosis factor receptor; *NF-κB*: nuclear factor κB; *IκB*: inhibitor of NF-κB; *cyt c*: cytochrome C; *N*: nucleus

as a deregulating agent in apoptosis, blocking or increasing the transduction of the signal, according to the place where it is produced and the type of enzymes it attacks [49, 50]. Peroxynitrite also intervenes at other sites in the apoptosis pathway. It was recently reported that ONOO^- was an activator of procaspases into caspases, the cysteine aspartases which are executioners of apoptosis triggered by activation of cell surface receptors. ONOO^- was demonstrated to activate procaspase 3 and to act on mitochondria; from mitochondria, cytochrome c is released which activates procaspases 3 and 9 [51, 52]. Peroxynitrite could also affect transcription factors by altering their cysteine residues and by oxidation of their redox sensitive iron-sulfur clusters and zinc-finger moieties, destroying the ability of proteins to bind to DNA or to recognize the correct DNA sequences [47].

Evidence for Reactive Oxygen and Nitrogen Species in Humans

In human pathologies characterized by an acute inflammatory reaction such as sepsis and acute lung injury (ALI), to demonstrate a production of ROS is not an easy task and direct demonstrations are not easy to obtain. But, indirect proof is accumulating that ROS and RNOS are produced in blood and in lungs, mainly by the activity of leukocytes. Proof of an oxidative activity *in vivo* has been obtained by measuring the alterations of antioxidant status of blood (decrease in vitamin E concentrations, decrease in total antioxidant potential of plasma, fall in the activities of antioxidant enzymes) [53–54]. Broncho-alveolar fluids that are obtained from lung lavage during ALI and/or sepsis have been extensively studied [19, 55]. Oxidized anti-proteinases were found as early as 1986 [19], and granulocytic enzymes were measured as markers of phagocyte activation [56]. These studies accumulated data proving that an oxidative activity, largely linked to phagocyte activation, occurred in the lung during acute inflammation pathologies (Fig. 3).

Protein Nitrations: a Marker of *in vivo* Oxidative Stress?

For a few years, new methods have been proposed to identify the consequences of oxidative stress in biological fluids. These new techniques propose to measure 'nitrated proteins' as a marker of *in vivo* production of peroxynitrite, and these nitrated proteins are mainly considered to be 3-nitrotyrosine-associated proteins [57, 58]. They could be measured by immunological techniques (immunohistochemistry or ELISA with mono- or polyclonal antibodies, and Western blot analysis) or by quantitative analytical methods such as high performance liquid chromatography (HPLC) or gas liquid chromatography coupled to mass spectrometry (GCMS).

Many results have accumulated in humans reporting the presence of 3-nitrotyrosine, free or bound to proteins, in various pathologies, often associated with acute inflammation and septic diseases characterized by neutrophil activation and infiltration (Table 1). Nitrotyrosine was detected in extra- and intracellular proteins in human whole blood containing PMA-activated neutrophils [59]. Extensive nitration of protein tyrosines in human atherosclerosis was detected by immunohistochemistry [60]. In Western blots of crude homogenates of atherosclerotic aorta, the antiser-

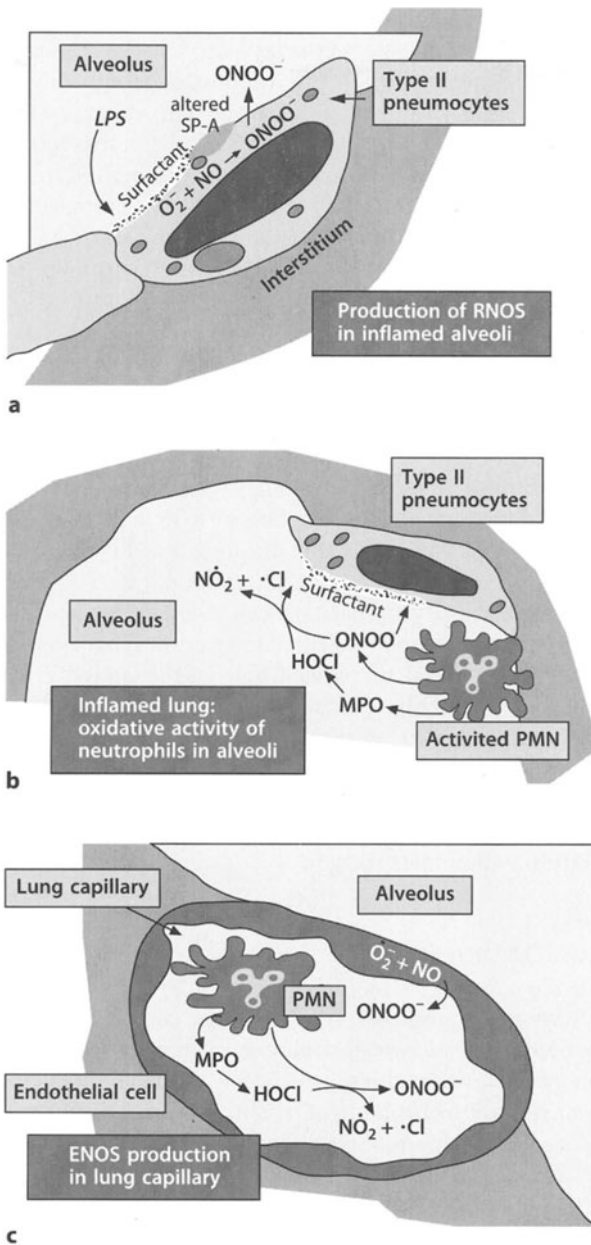


Fig. 3. Main points of peroxynitrite production in the lungs during inflammatory processes (cell activation by endotoxin, by cytokines ...), highlighting the role of mitochondria in hypoxia consecutive to respiratory distress. $NO\cdot$ is produced by activated lung cells and by stimulated phagocytes. Superoxide anion ($O_2^{\cdot-}$) is produced by xanthine oxidase in hypoxic cells and by stimulated phagocytes

um to nitrotyrosine reacted with a wide range of protein bands (approximately 180 to 30 kd) [61].

In the respiratory tract, the reality of local synthesis of NO by constitutive and inducible NOS is no longer contested. NOS is present in airway and alveolar epithelial cells, alveolar macrophages, neutrophils, mast cells, vascular endothelial and smooth muscle cells. Most of these cells are also equipped with NADPH oxidase responsible

Table 1. Summary of the studies reporting the detection of nitrated proteins in humans

| Pathology and/or site of detection | Type of study | Reference [number] |
|--|-----------------|----------------------------------|
| Atherosclerosis (coronary arteries) | human | Beckman JS, et al 1994 [60] |
| Acute lung injury | human | Haddad IY, et al 1994 [64] |
| Rheumatoid arthritis (synovial fluid, blood) | human | Kaur H & Halliwell B 1994 [69] |
| Acute lung injury | human | Kooy NW, et al 1995 [66] |
| Myocardial inflammation | human | Kooy NW, et al 1997 [71] |
| Atherosclerosis (intima) | human | Leeuwenburgh C, et al 1997 [72] |
| Obliterative bronchiolitis | human | McDermott CD, et al 1997 [67] |
| Cystic fibrosis | human | van der Vliet A, et al 1997 [73] |
| Idiopathic pulmonary fibrosis | human | Saleh D, et al 1997 [74] |
| Bronchopulmonary dysplasia | human (infants) | Banks BA, et al 1998 [75] |
| Perennial nasal allergy | human | Sato MN, et al 1998 [76] |
| Asthma | human | Saleh D, et al 1998 [77] |
| ARDS (with NO administration) | human | Lamb NJ, et al 1999 [78] |
| Lung allotransplantation | human | Andrade, et al 2000 [68] |
| Bronchopneumonia/ARDS | human | Mathy-Hartert, et al 2000 [56] |

for O_2^- production, and with xanthine dehydrogenase, a ubiquitous cytosolic enzyme of which the activity of dehydrogenase can be shifted to an activity of oxidase generating O_2^- in pathological circumstances such as ischemic situations. All the conditions are thus met to permit the simultaneous production of NO and O_2^- , particularly in cases of infection and inflammation. In these pathological conditions, the production of NO is increased as a host defense against pathogens [20, 62] and the respiratory burst of phagocytes releases oxidant species. The probability is thus high of an *in situ* production of ONOO⁻ that will be responsible for oxidative alterations. Nitration is one of these oxidative alterations. As neutrophil infiltration and activation is a common event in lung inflammation, nitration could be further enhanced by myeloperoxidase released by neutrophils. Quantitation of nitrotyrosine levels was performed with polyclonal antibodies in lung sections from patients with acute lung injury and hyperoxic injury, using quantitative immunofluorescence [14, 63–66]. Nitrotyrosine has been found in inflammatory cells, airway epithelium and vascular epithelium of human lung allotransplants [67]. Nitrotyrosine and other NO-derived products (nitrites and nitrates) were found in epithelial lining fluid of lung allotransplants, in which chlorotyrosine was also present indicating an interaction with neutrophil myeloperoxidase [68]. Nitrated proteins were also detected in other respiratory pathologies such as asbestosis, idiopathic pulmonary fibrosis, asthma, etc. (Table 1) [14, 60, 66, 69–78].

In the lung, ONOO⁻ is reported to be responsible for the alterations of surfactant of type II alveolar cells [79]. A recent study demonstrated an increase of nitrated proteins in bronchoalveolar lavage fluid (BAL) of acute respiratory distress syndrome (ARDS) patients treated with inhaled NO, as a consequence of ONOO⁻ formation or myeloperoxidase activity [78]. These observations raise the question of a possible noxious role of NO administration in ARDS patients [50]. We found elevat-

ed concentrations of nitrated proteins in BAL fluids of patients with ventilator-associated bronchopneumonia and/or ARDS [56]. These nitrated protein concentrations correlated to the number of neutrophils and the presence of active neutrophil enzymes (myeloperoxidase and elastase). It is plausible that these nitrated proteins are the footprint of ROS and more specifically the markers of peroxy-nitrite production as the result of *in situ* NO and superoxide anion production by the activities of NOS and NADPH-oxidase. These nitrated proteins could also originate from the interactions of nitrites (derived from NO) with the activity of myeloperoxidase and its by-product HOCl.

There is thus considerable evidence of peroxy-nitrite-induced nitrations and alterations *in vivo*, but are nitrated proteins simply markers of oxidative processes that previously occurred or can these altered proteins be cytotoxic for the neighboring cells and tissues, particularly in the lungs? This question is far from being solved, but it is now clear that tyrosine nitration can perturb the signal transduction in the cells, that nitration of guanine may result in DNA mutations, and that nitration of enzymes leads to dysfunctions. In the lung, nitration of surfactant protein A causes a loss of the ability of this protein to aggregate lipids or to bind to mannose receptors, which could compromise the antibacterial functions of surfactant protein A. In our studies on nitrated proteins in BAL fluids, we also found that these BAL fluids were cytotoxic for healthy alveolar cells (type II alveolar cells, A549) in culture in correlation with the concentrations of myeloperoxidase, that we explained by an interaction of myeloperoxidase activity with H₂O₂ or NO-derived products released by living alveolar cells.

However, more *in vivo* studies identifying the target proteins for nitration are needed to assure that nitrated proteins are more than biomarkers of oxidative activity. The recent description of a 'nitrotyrosine dinitrase' activity in rat tissue homogenates suggest that regulation or defense mechanisms against tyrosine nitration exist *in vivo* [80]. It is thus rather plausible that excessive nitrations would have toxic consequences.

The Effects of Oxidative Stress on Signal Transduction *in vivo*

The *in vivo* effects of oxidative stress on signal transduction are less documented. Data obtained with cell cultures demonstrated the role of ROS in the signal transduction pathways activated by the proinflammatory cytokines IL-1 β and TNF- α . These two cytokines regulate the redox-sensitive transcription factor NF- κ B, which participates in a variety of immune, inflammatory, and acute-phase responses. As this nuclear factor is more and more recognized as an anti-apoptotic factor, sustained NF- κ B activation in the lung of ARDS patients may increase the survival of proinflammatory neutrophils, perpetuating the process of lung inflammation. In the lungs of patients with ARDS, increased numbers of neutrophils activated to produce ROS and inflammatory cytokines are present, and have a decreased rate of apoptosis [81, 82]. In sepsis, an increased activation of NF- κ B in peripheral blood monocytes correlated with poor outcome [83]. Alveolar macrophages from patients with ARDS showed increased NF- κ B activation compared with those of ventilated patients without ARDS, in relation to the acute inflammation response in the lung [84].

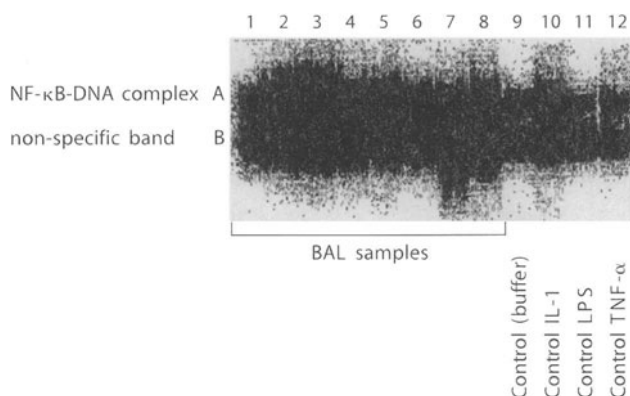


Fig. 4. Detection of NF-κB activation in alveolar cell in culture incubated with bronchoalveolar lavage fluids of intensive care patients (bronchopneumonia and/or ARDS). Comparison with NF-κB activation obtained with lipopolysaccharide (LPS), interleukin (IL)-1β and tumor necrosis factor (TNF)-α (well-known activators of this transduction factor)

Recently, we observed that BAL fluids of patients with ventilator-associated bronchopneumonia (VAP) and/or ARDS increased the activity of NF-κB in normal alveolar cells, and that this increase of NF-κB activity was significantly different in VAP and ARDS groups compared to a group of ventilated patients without these pathologies (Fig. 4). As expected, this activation was correlated with the BAL concentrations of IL-1, but also with the concentrations of IL-8, active myeloperoxidase and the number of neutrophils in BAL. For a better understanding of these observations, we tested the effects of specific ROS on the activation of NF-κB in alveolar cells by using isolated myeloperoxidase and the NO-derived species nitrites and peroxy-nitrite. Concentrations of nitrites and peroxy-nitrite ranging from 10^{-7} to 10^{-5} M were found to activate NF-κB to the same level as 100 U TNF-α, 10^{-5} M H_2O_2 , and 10^{-5} M HOCl, compounds which were previously reported as NF-κB activators. We also found that 5 μg of human myeloperoxidase was as active as TNF-α, and attributed this activity to the *in situ* reaction of myeloperoxidase with H_2O_2 or NO-derived species that were produced by the alveolar cells [15, 16, 58].

Conclusion

The importance of RNOS is no longer contested in normal and pathological pathways, and NO-derived species, especially peroxy-nitrite, are considered as responsible for previously reported toxic effects of NO. Some trends are also clear in the subject of ROS and oxidative stress on signal transduction and apoptosis. But the fields of RNOS research changes rapidly and most of our knowledge is derived from laboratory studies on animal models or isolated cells. These studies, albeit necessary and remarkable, can bring simplified concepts, giving the idea of a clear sequence of events that starts with stimuli and ends with oxidative stress leading to cell damage or gene activation. However, *in vivo* studies do not fit these simple schemes for the

role of oxidative stress and ROS in humans. In inflammatory situations, an oxidative stress is expected arising from an excessive stimulation of neutrophils of which the apoptosis appears to be delayed. Interventions that inhibit RNOS production and signal transduction alterations could have a beneficial anti-inflammatory response *in vivo*, but could also create immunosuppression. To secure anti-RNOS therapies, the role of specific RNOS remains to be more clearly delineated, especially by studies using 'physiological' concentrations of ROS.

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