

Research Article

Influence of chamber design on the gas sensing performance of graphene field-effect-transistor



Lorenzo Lopez Jr. 1,2,3 · Vernalyn Copa 1,2,3 · Takeshi Hayasaka 3 · Maria Angela Faustino-Lopez 1,2 · Yichuan Wu 3 · Huiliang Liu 3 · Yumeng Liu 3 · Elmer Estacio 1,2 · Armando Somintac 1,2 · Liwei Lin 3 · Arnel Salvador 1,2

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Abstract

We report on the influence of chamber design on the gas sensing performance of a graphene field-effect-transistor (GFET). A conventional chamber (V = 400 ml) and a cap chamber (V = 1 ml), were used to perform dynamic measurements on a GFET. To gain a-priori knowledge on the gas flow in the chambers, Naiver–Stokes and convection-diffusion equations were numerically-solved using COMSOL Multiphysics. We numerically and experimentally observed two main factors that can affect the GFET performance: (1) the gas flow direction through the chamber and (2) the chamber volume. At 5-min exposure time, at least 200% higher GFET sensitivity was calculated from the cap chamber, which is expected since the conventional chamber is 400 times larger. Interestingly, even when the conventional chamber is fully saturated (at 90-min exposure time), the GFET sensitivity in the cap chamber is still better by 28.57%. We attributed this behavior to the swirling vapor flow in the cap chamber brought about by the U-shaped path. This effect causes multiple interaction of H_2O molecules with the GFET, resulting to higher computed sensitivity. However, at higher relative humidity, the GFET becomes populated, reducing the number of H_2O molecules that can re-interact with the sensor. In terms of GFET transient characteristics, a 154% and 86.9% faster response and recovery, respectively, were observed in the cap-design. This was due to its smaller volume that minimized poorly purged region in the chamber. But if the chambers have the same volumes, we may infer a faster GFET response and recovery from the conventional chamber where the gas flow is unperturbed. These results could contribute in designing a time efficient and cost-effective gas sensing system.

Keywords Gas sensor · Chamber · COMSOL MultiPhysics · Graphene · Field effect transistor

1 Introduction

Research on gas sensing technology had been widely explored for the past decades. One of the major parameters in the development of gas sensing applications is the design of the measurement system. Normally, sensors are tested using the following techniques [1, 2]: (1) static system and (2) dynamic system. The former typically consists of an airtight chamber, with vacuum levels around 1.333×10^{-5} mbar, that has a gas inlet for injection of the test gas

[3-8]; while the latter uses continuous gas flow through the chamber [9-14].

Materials modifications [15–19], sensors design [20–24], and testing schemes [25–29], were used to improve the performance of gas sensor devices; but only a few [30–32] reports on its dependence with the chamber design. This factor is important to efficiently investigate the characteristics of the sensor being tested. With an optimized design, turbulence and back flow are minimized, resulting to a more uniform gas distribution inside the chamber. In

Lorenzo Lopez Jr., Iplopez@msep.upd.edu.ph | ¹Materials Science and Engineering Program, University of the Philippine Diliman, 1101 Quezon City, Philippines. ²National Institute of Physics, University of the Philippines Diliman, 1101 Quezon City, Philippines. ³Berkeley Sensor and Actuator Center & Mechanical Engineering Department, University of California Berkeley, Berkeley City, CA 94720, USA.



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effect, sensor characteristics (i.e. sensitivity, response, and recovery) can be accurately investigated.

Recently, Niyat et al. [33] experimentally and numerically studied the gas sensing performance of tin oxide/reduced graphene oxide (SnO₂/rGO) inside a cylindrical chamber. They concluded that the pressure distribution in such a system is higher on the edges than at the center. Moreover, Endres et al. [31], compared the gas sensor array sensing performance in linear and circular measurement chambers. They experimentally showed that the former provides a shorter purging time (25 s for 500 sccm), while the latter allows for a more uniform gas flow.

In this work, we compared the performance of a graphene field-effect-transistor (GFET) in conventional (V = 400 ml) and cap (V = 1 ml) chambers in a dynamic system. Our experimental results, along with simulations using Naiver-Stokes and convection-diffusion equations, show that (i) the gas flow direction in the chamber significantly affects the GFET sensitivity; and (ii) a smaller chamber volume minimizes poorly purged regions resulting to a faster GFET response and recovery. This study may be useful in optimizing a chamber design to efficiently investigate the device under test.

Fig. 1 Fabrication process of GFET. (1) Metal electrodes deposition of Au/Pd (50 nm/25 nm) using electron beam deposition, (2)–(3) Graphene etching, (4) wire bonding of GFET on test chip

2 Experimental details

2.1 GFET fabrication

The GFET was fabricated via a shadow-masking technique. Three laser-cut masks were used to form the patterns of the metal electrodes and the graphene channel. First, electrodes (50 nm Au/ 25 nm Pd) were deposited on the graphene substrate (Graphenea [34]: 10 mm x 10 mm of monolayer graphene on SiO_2/Si) using electron beam deposition (schematic diagram shown in Fig. 1). Then the graphene channel was defined via a two-step etching procedure using O_2 plasma. Lastly, the metal electrodes were connected to the ceramic packaging using a wire bonder.

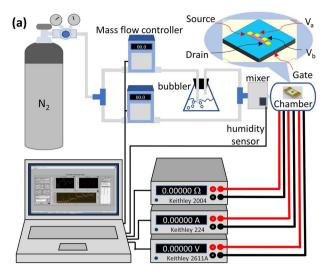
2.2 Graphene characterization and GFET testing

To investigate the quality of the graphene after the fabrication process, Raman spectroscopy and scanning electron microscopy (SEM) were performed. The Raman signal of graphene was obtained using a 514 nm wavelength excitation source, with a power of 5 mW and a beam size of \sim 1 μ m. The micrograph of the device was captured using a Schottky field emission SEM (ZEISS-SEM).

Prior to the gas sensing experiment, the gate voltage (V_g) was modulated, to ensure that the GFET is working properly. The V_g and source-drain current (I_{sd}) were supplied using a high accuracy DC power supply, while a 7.5-digit precision digital multimeter was used to measure the potential drop across V_a and V_b (Fig. 2a). 1 μ A and -3 V were used as the I_{sd} and V_g , respectively, to comply with the condition for wearable devices which requires ultra-low power consumption.

To properly control the gas flow through the chamber, mass flow controllers (Omega FMA3706) were incorporated to the gas sensing system. Target vapor concentrations of 20%, 35% and 50% were produced by bubbling $\rm H_2O$ using high purity (99.98%) $\rm N_2$ gas and was then fed through the chamber. The total flow rate of the injected gas was set to 200 sccm.

We used two chamber designs: (i) conventional (V = 400 ml) and (ii) cap (V = 1 ml) chambers. Our cap chamber has a U-shape path for the gas flow direction, and linear for the conventional chamber. The cap chamber is made from delrin installed with BNC cable for electrical connections, while the conventional chamber is composed of stainless steel equipped with screw terminal





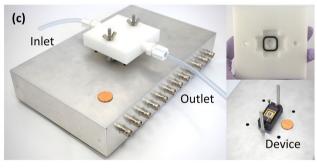


Fig. 2 a Schematic diagram of gas sensor testing system showing the gas line and electrical connections. Actual image of **b** conventional chamber and **c** cap chamber with the mounted GFET device

wires, as shown in Fig. 2b and c, respectively. For the cap chamber, the sensor is mounted on a fixed integrated circuit socket, and on a breadboard for the conventional chamber. A commercial humidity sensor (HIH-4000-001) was also integrated to the system to monitor the relative humidity (R.H.). The entire gas sensing system was automated for ease of data collection.

3 Results and discussion

3.1 Properties of graphene channel

The effective area of graphene channel was measured to be 2.4 mm by 0.4 mm, as shown in Fig. 3a. Since shadow-masking technique was utilized to pattern the electrodes, it is expected that the metal will diffuse during the deposition process which appears as the dark regions around the electrodes.

Figure 3b shows the Raman spectrum of monolayer graphene before and after fabrication process. The prominent peaks at 1340.1 cm⁻¹, 1592.9 cm⁻¹, and 2680.3 cm⁻¹ correspond to the D-peak, G-peak and 2D-peak respectively [35]. The enhancement of the D-peak signifies that defects were introduced to graphene after fabrication process [36]. The intensity ratio of the 2D-peak and the 2G-peak (I_{2D}/I_G) of graphene before and after fabrication are 1.4 and 1.0, respectively; signifying the presence of a monolayer graphene even after the fabrication process [35]. Due to instrument limitations, the neutrality point from the I_{sd}-V_a curve (Fig. 3c) cannot be observed. A GFET using an intrinsic graphene is expected to have a Dirac point close to 0 V_a [37]. But since the substrate that we used is SiO₂, when exposed to ambient environment, it induces an involuntary p-type doping [38, 39]. This shifts the Dirac point to the positive side in terms of V_a , in this case, above +80 V, beyond our sweeping voltage range. Nonetheless, the fabricated GFET was successfully modulated from -80 V to +80 V, indicating that the device is working properly.

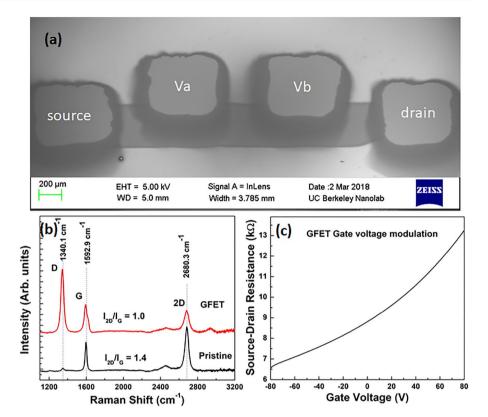
3.2 COMSOL multiphysics simulation of gas flow

The gas flow through the chamber is modeled using laminar flow by solving Navier-Stokes equation [40] which is given by,

$$\rho \frac{D\overline{U}}{Dt} = -\nabla P + \rho g + \eta \nabla^2 U \tag{1}$$

where ρ , U, P, g, and η are the mass density, velocity field, gas pressure, acceleration due to gravity, and viscosity, respectively. The expression on the left side corresponds to the inertial forces which is the superposition of pressure forces (first term), buoyancy forces (second term), and viscous forces (third term). Equation 1 can also be expressed into Cartesian coordinates as follows:

Fig. 3 a SEM image of the GFET showing the graphene channel and patterned electrodes; **b** Raman signal shows the presence of graphene even after the fabrication process; and **c** Gate voltage modulation of GFET signifying that the device is working properly



$$X : \rho \left(\frac{\partial U}{\partial t} + U \frac{\partial U}{\partial x} + V \frac{\partial U}{\partial y} + W \frac{\partial U}{\partial z} \right)$$

$$= -\frac{\partial P}{\partial x} + \rho g_x + \mu \left(\frac{\partial^2 U}{\partial x^2} + \frac{\partial^2 U}{\partial y^2} + \frac{\partial^2 U}{\partial z^2} \right)$$
(2)

$$Y: \rho \left(\frac{\partial V}{\partial t} + U \frac{\partial V}{\partial x} + V \frac{\partial V}{\partial y} + W \frac{\partial V}{\partial z} \right)$$

$$= -\frac{\partial P}{\partial y} + \rho g_y + \mu \left(\frac{\partial^2 V}{\partial x^2} + \frac{\partial^2 V}{\partial y^2} + \frac{\partial^2 V}{\partial z^2} \right)$$
(3)

$$Z: \rho \left(\frac{\partial W}{\partial t} + U \frac{\partial W}{\partial x} + V \frac{\partial W}{\partial y} + W \frac{\partial W}{\partial z} \right)$$

$$= -\frac{\partial P}{\partial z} + \rho g_z + \mu \left(\frac{\partial^2 W}{\partial x^2} + \frac{\partial^2 W}{\partial y^2} + \frac{\partial^2 W}{\partial z^2} \right)$$
(4)

To obtain the field velocity of the gas flowing through the chamber, Navier-Stokes equation must be simultaneously solved with the continuity equation, which is given by

$$\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mu) = 0 \tag{5}$$

For incompressible gas flow, this equation yields:

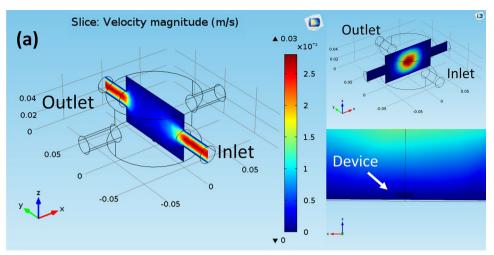
$$\nabla \cdot \mu = 0 \tag{6}$$

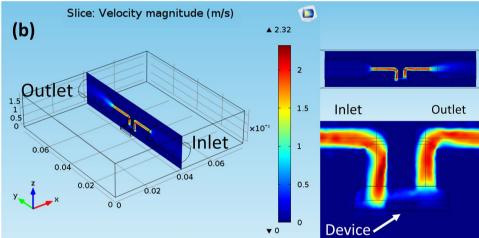
No-slip and adiabatic boundary conditions were set all over the chamber except for the inlet and outlet.

The parameters for the dimensions of the chambers were initially defined to easily generate their geometry, wherein conventional and U-shape chambers have 400 ml and 1 ml volume, respectively. The meshing was built using a free tetrahedral. A time dependent study, with time profile ranging from 1 s to 3000 s, was implemented to generate the velocity profile of $\rm H_2O$ vapor through the chambers, and the $\rm H_2O$ concentration profile at the device location.

The velocity profile of gas through the conventional and cap chambers are depicted in Fig. 4. The inlet flow rate was set to 200 sccm. As expected, the gas flow follows the linear path and U-path directions for the conventional and cap chambers, respectively. Due to the

Fig. 4 Laminar flow of gas through the **a** conventional chamber and **b** cap chamber showing the x-z plane coordinate and device location. A linear gas flow direction is observed from the conventional chamber, while a U-path direction for the cap chamber. The gas velocity magnitude in cap chamber is 100 times faster than that in the conventional chamber, which is due to dimension differences





smaller volume of the cap chamber, the velocity magnitude of gas through it is 100 times faster. The gas velocity distribution in the conventional chamber significantly changes depending on the location of the device, due to its large volume. In our simulation, we positioned it at the center, to accurately compare the gas flow in both chambers.

Furthermore, convection-diffusion equation [41] was used to model the gas transport through the chamber, which is given by,

$$\frac{\partial c_i}{\partial t} + \nabla \cdot (-D_i \nabla c_i + c_i u) - R_i = 0 \tag{7}$$

where *c*, *D*, *u*, and *R* are the concentration, diffusivity, velocity, and production or consumption term, respectively.

The cap chamber reached the target set point at 35% H_2O vapor concentration, for an exposure time (t_{exp}) of 5

mins. At this t_{exp} , the conventional chamber reach only 12% H_2O vapor concentration (shown in Fig. 5). This is expected since it has a larger volume. Based from our simulation, the time required for the conventional chamber to reach the target set point is 90 mins. At saturation levels, we can assume that the number of H_2O molecules interacting with the device in both chambers are approximately equal. We can therefore abate the effect of volume differences and effectively compare gas flow characteristics. These simulation results were also investigated experimentally in the following section.

3.3 Device testing of GFET

Figure 6 shows the GFET response to different H_2O vapor concentrations, using the conventional and cap chambers. The inset shows the R.H. measured by the commercial sensor. For the cap chamber, the target set points of

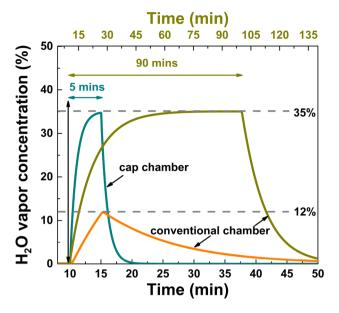


Fig. 5 The gas transport of H_2O vapor through the conventional chamber with $t_{exp}=5$ mins (orange line) and $t_{exp}=90$ mins (green line), and cap chamber with $t_{exp}=5$ mins (cyan line). The cap chamber reach the target set point at 35% at $t_{exp}=5$ mins, while the conventional chamber requires $t_{exp}=90$ mins. Double x-axis was used for comparison purposes

20%, 35% and 50% R.H. were reached at $t_{exp} = 5$ mins. This agrees well with the simulation results presented in the previous section. Using the same exposure time, the sensor response in the conventional chamber barely reach half of the target set points. Additionally, after the sensor's recovery, a slight increase in its response can be noticed. This can be attributed to the non-uniformity of H_2O vapor distribution, which may be an effect of turbulence and back flow inside the large-volume conventional chamber. This effect was minimized when the GFET is exposed for 90 mins, which indicates that the chamber is fully saturated and the H_2O molecules interacting with the GFET is more uniform.

The GFET sensitivity (S) can be written as,

$$S(\%) = \left| \frac{\Delta R}{R_0} \right| \times 100 \tag{8}$$

where ΔR , R_0 are the change of GFET resistance, and the GFET resistance in the carrier gas, respectively.

The calculated sensitivities of the GFET in conventional and cap chambers for different R.H. levels, are tabulated in

Table 1. At $t_{exp} = 5$ mins, we can observe at least 200% GFET sensitivity improvement from the cap chamber. This is expected since the conventional chamber volume is 400 times larger than that of the cap chamber. But even when the conventional chamber is fully saturated (at $t_{exp} = 90$ mins), the GFET sensitivity is still better by 28.57% (for R.H. = 20%). At 50%, however, the GFET sensitivities in the two chambers become comparable. The sensitivity ratios ($\frac{\Delta R}{R_0}$ | $t_{conventional}$) are shown in Fig. 7. Generally, at higher R.H., the ratio decreases and approaches the simulation values. The ratios fit better with the simulation when both the chambers are fully saturated (Fig. 7b).

The differences in the sensitivity is attributed to the different gas flow directions for each chamber (illustrated in Fig. 8). The U-path in the cap design may have induced a swirling vapor flow in the chamber, introducing multiple interaction of $\rm H_2O$ molecules with the device, which resulted to the higher calculated sensitivity at lower R.H. (20% and 35%). However, at higher R.H. (50%), the GFET becomes more populated, reducing the number of $\rm H_2O$ molecules that can re-interact with the device in the cap chamber. This means that, at higher R.H., the number of $\rm H_2O$ molecules interacting with the GFET in both the chambers will be comparable, resulting to approximately equal sensitivities. The swirling vapor effect was also observed by Zhang et al. [42], but for a different application, i.e. in a U-shaped chemical vapor deposition chamber.

The response and recovery time of the GFET were obtained using Eq. 9 and Eq. 10 respectively.

$$t_{res} = t_{90} \tag{9}$$

$$t_{rec} = t_{10} - t_{max} (10)$$

where t_{max} , t_{90} and t_{10} are the time at maximum sensor response, time at the sensor response is 90% of the maximum, and time at the sensor response is 10% of the maximum, respectively.

Figure 9 shows the GFET response and recovery times in the conventional and cap chambers. It can be noted that at higher R.H., the GFET response does not vary very much, and its recovery is slower. The fast response and slow recovery in GFET can be explained by the interaction of H₂O molecules with the graphene/SiO₂ layers.

According to Smith et al. [43], this humidity sensing effect in graphene/SiO₂ layers is attributed to the interaction of graphene layer and impurity bands in SiO₂, with the H₂O electrostatic dipole moment. In effect, a

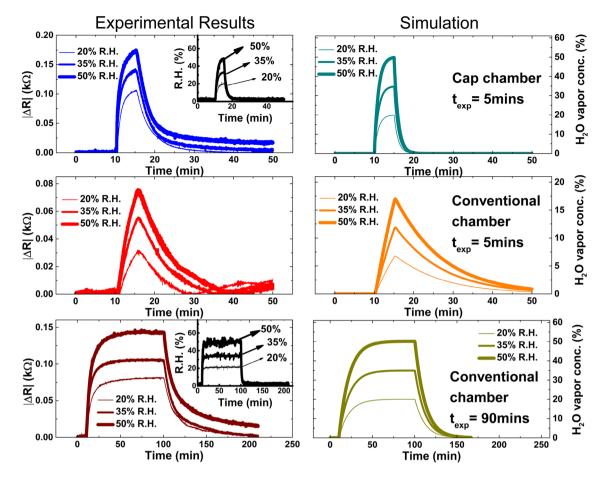


Fig. 6 The GFET response in a cap chamber and a conventional chamber (left) with its corresponding simulations (right). The GFET was exposed for 5 mins (top) in the cap chamber, and 5 mins (middle) and 90 mins (bottom) in the conventional chambers using 20%, 35% and 50% H₂O vapor concentrations. Both the experi-

mental and simulation results show that the GFET response reach the target set point in the cap chamber at $t_{exp}=5$ mins, while in the conventional chamber $t_{exp}=90$ mins is required. The relative humidity was monitored and shown in the inset

Table 1 Summary of the calculated GFET sensitivities in the cap and conventional chambers under the influence of different R.H. levels. At lower R.H. level, the GFET exhibits better sensitivity in cap chamber compared to that of in a conventional chamber

R.H. (%)	GFET sensitivity					
	Cap chamber ^a (%)	Conventional chamber ^a (%) (%)	Conventional chamber ^b (%) (%)			
20	1.2	0.4	0.9			
35	1.5	0.6	1.2			
50	2.0	0.9	1.9			

 $at_{exp} = 5 \text{ mins}$

graphene/SiO $_2$ -based sensor will have a faster response time and slower recovery time. Due to the smaller volume of the cap chamber, a faster GFET response (by at least 154%) and recovery (by at least 86.9%) are expected, as shown in Table 2. A smaller chamber volume minimizes the poorly purged region. But when volumes are equal, we may expect a faster GFET response and recovery in the conventional design, due to the unperturbed gas flow in the chamber's linear path direction.

On a related note, Annanouch et al. [44] also demonstrated the effect of chamber design, but for a SnO_2 chemiresistor. Their results showed that a boat-shaped chamber (V = 2.35 ml) have faster response and recovery

 $^{^{}b}t_{exp} = 90 \text{ mins}$

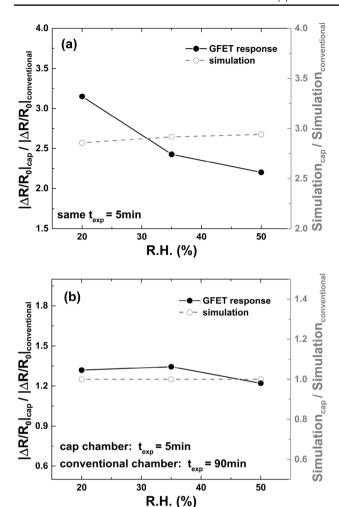
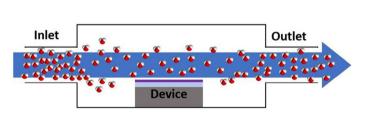


Fig. 7 The GFET sensitivity ratio in the conventional and cap chambers for the **a** same exposure time, and for **b** different exposure times along with its corresponding simulations. Generally, At higher R.H., the $\left|\frac{\Delta R}{R_0}\right|_{cap}/\left|\frac{\Delta R}{R_0}\right|_{conventional}$ decreases and approaches the simulation values

times compared to a cross-shaped chamber (V = 300 ml). They attributed this to the enhanced gas concentration homogeneity and reduced dead volume in the boat-shaped chamber. But in our case, aside from the effect of volume, we surmise that the improvement in the detected target gas is also based on the re-interaction of the gas molecules with the sensor, brought about by the U-shaped path in the cap chamber.

4 Conclusion

We successfully investigated the effect of chamber design on the gas sensing performance of a GFET, via comparing GFET performances in conventional (V = 400 ml) and cap (V = 1 ml) chambers. At saturation levels (abating volume differences), a higher GFET sensitivity was observed in the cap chamber for lower R.H. levels (20% and 35%). This was attributed to the swirling vapor effect in the U-path direction of the chamber, causing multiple interaction of H₂O molecules to the GFET. At high concentration (50% R.H.), however, this effect is minimized, resulting to comparable GFET sensitivities between the chambers. Furthermore, faster GFET response and recovery times were observed in the cap chamber. The smaller volume of the cap-design minimizes the poorly purged region in the chamber. But if the chambers have the same volumes, faster GFET response and recovery may be expected from the conventional chamber, where the gas flow direction is unperturbed. This study can contribute not only on the optimization of gas chamber design for efficient device testing; but



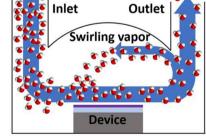


Fig. 8 Illustration of gas flow through the conventional chamber (top) and the cap chamber (bottom). The gas flow in the cap chamber follows a U-path direction that resulted to a swirling vapor,

while a linear-path in the conventional chamber, wherein the gas flow is unperturbed. In effect, a better sensitivity is observed in the cap chamber

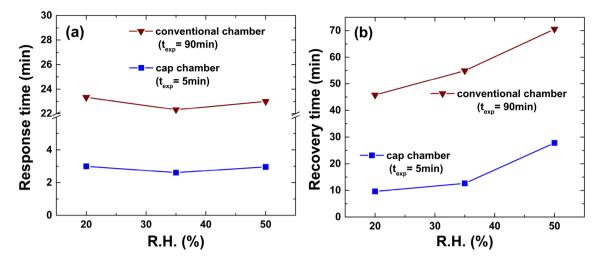


Fig. 9 The **a** response time and **b** recovery time of GFET in the conventional and cap chambers under the influence of different humidity levels. The GFET transient characteristic is better in the cap chamber due to its smaller volume compared to the conventional chamber

Table 2 Summary of the calculated GFET response and recovery in the cap and conventional chambers under the influence of different R.H. levels. The percent difference (% diff.) for the t_{res} and t_{rec} using the cap and conventional chambers were calculated

R.H. (%)	t _{res}	t _{res}			t _{rec}		
	Cap ^a	Con. ^b	% diff.	Cap ^a	Con. Con. ^b	% diff.	
20	3.0	23.3	154.4	9.6	45.7	130.6	
35	2.6	22.3	158.0	12.6	54.9	125.2	
50	2.9	23.0	154.3	27.8	70.6	86.9	

 $^{^{}a}t_{exp} = 5 \text{ mins}$

also on possible applications in microfluidics and drug delivery, wherein efficient fluid transport is required.

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Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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 $^{^{}b}t_{exp} = 90 \text{ mins}$

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