Research Article

Observation of CCNR-type electrical switching in $Zn_{0.3}Mn_{0.7+x}Si_xFe_{2-2x}O_4$ spinel ferrite series



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Abstract

A nonvolatile memory effect exhibited by electric field-induced resistance switching has been the topic of intense research not only due to its applications as resistive random access memory but also from the basic physics point of view. Among several binary and ternary mixed oxide compounds, the ones which possess magnetic ions have shown a great promise. Spinel ferrite system $Zn_{0.3}Mn_{0.7+x}Si_xFe_{2-2x}O_4$ with varying *x* is investigated for its novel electrical switching properties. Both temperature and applied voltage dependence of current-controlled negative resistance-type electrical switching showed better than 200% of resistive switching ratios. Bulk polycrystalline samples showed composition *x* dependence of resistive switching. The current-voltage characteristics are modeled for low and high applied field regime, and the presence of space-charge-limiting current is confirmed. Thin films of the ferrite system grown by pulsed laser deposition showed almost nonexistence of resistive switching, suggesting that the bulk composition of the compound has a major role to play against the film–electrode interface.

Keywords Spinel ferrite · Electric switching · Space-charge-limiting current · Thin film

1 Introduction

In search of novel memory devices in the modern semiconductor industry, the resistive random access memory (RRAM) has shown a great promise. Despite several complex systems exhibiting promising RRAM behavior, the governing mechanism of electrical switching observed in materials, either the bulk or films, is still under debate. An in-depth analysis of the current–voltage (*I–V*) characteristics of materials is key in discussing the mechanism of RRAM. Owing to the existence of charge carriers produced thermally, the *I–V* characteristics of most materials in a small applied electric field are Ohmic (linear) in nature. With the increasing electric field, *I–V* behavior begins to be non-Ohmic in the majority of nonmetallic substances. A rapid increase in *I* with sufficiently high *V* makes the characteristics nonlinear or rather an exponential dependence near to the electronic breakdown of the material. In certain compounds, the *I* increases in a linear manner up to a specific applied field and beyond that huge current flows with *V* abruptly decrease. The mechanism in which materials demonstrate a sudden changeover from high conduction state to low conduction state is referred to as resistive switching (RS). There are three types of RS, namely (a) negative resistance controlled by a voltage (VCNR) or N-type, (b) negative resistance controlled by current (CCNR) or S-type and (c) bi-stable or memory switching.

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Very recent (2019–2020) review articles describe all the aspects, starting from fundamentals of resistive switching (RS), mechanisms responsible for switching action, fabrication processes, materials properties, etc., of resistive switching devices for nonvolatile RRAM and computing applications [1-9]. Many pure oxide thin film heterostructures, perovskite oxides, metal halide perovskite, manganites, single-crystalline materials, and thin films of chalcogenide semiconductors, organic and hybrid materials including graphene oxide have shown to exhibit resistive switching [1–9]. Mixed metal oxide systems such as ferrites (spinels, garnets, orthoferrites, etc.,) exhibiting memory and switching characteristics are of prime significance in the field of digital electronics. In the last couple of years (2018–2020), a very few research reports are available on RS exhibited by spinel ferrites in nanocrystalline [10–12], thin films [13-20], and nanocomposite [21] forms. Interestingly and surprisingly, no work has been reported on RS action demonstrated by bulk polycrystalline ferrite materials. Of course, it is seldom observed phenomenon in such a class of materials. In this sense, the present work is important. On the other hand, quadruple perovskite, calciumcopper-titanate (CaCu₃Ti₄O₁₂), and their isomorphous systems, pristine and substituted with different cations in various forms are commonly exhibited room-temperature and high-temperature nonlinear current versus voltage characteristics suitable for varistor applications [22].

In the present investigation, the composition, $Zn_{0.3}Mn_{0.7}Fe_{2}O_{4}$, is selected as parent material and Si⁴⁺ is chosen for the substitution based on the following facts: (1) from the high-frequency applications' point of view, this composition possesses excellent properties, for instance high initial magnetic permeability, a large value of saturation magnetization, high electric resistivity and low power loss, (2) the substitution of a tetravalent cation such as Si⁴⁺ lowered eddy current losses and increased the temperature stability, (3) Si⁴⁺ ions have a tendency to segregate at the grain boundaries and create a low conductivity layer that strongly affects structural, microstructural, magnetic, electric and dielectric properties [23-28]. The recent advancements and emergent curiosity in the RS demonstrated by mixed oxides prompted us to re-examine the classic ferrite system to comprehend the high electric field instability in the multicationic spinel ferrite series, $Zn_{0.3}Mn_{0.7+x}Si_{x}Fe_{2-2x}O_{4}$.

2 Experimental details

Comprehensive experimental technicalities relating to the synthesis of spinel ferrite series, $Zn_{0.3}Mn_{0.7+x}Si_xFe_{2-2x}O_4$ with x = 0.0, 0.1, 0.2 and 0.3, by usual two-step sintering ceramic route, crystalline phase identification as well as

structural parameters determination by analyzing X-ray diffraction patterns are given in previous reports [23, 24].

The scanning electron microscope, model XL-30 ESEM with EDAX (make: Philips, Netherlands) operated at 30 kV, was employed to register energy-dispersive analysis of X-ray spectra. Current (*I*) against voltage (*V*) (ranging from 0 to 400 V) measurements were taken using Aplab made (model 7332) high-voltage dc regulated power supply at selected temperatures in the range of 300–673 K. The cylindrically pelletized samples having a diameter of 1.0 cm and thickness of the order of 0.3–0.4 cm were used for *I–V* characteristic measurements. The faces of the disc sample were polished by rubbing with zero-grade sand-paper and washed in mild hydrochloric acid (HCI) and propanone. Eventually, graphite was applied on both the plain faces of the pellets on which aluminum foil was also put up for perfect electrical contacts.

The high-quality ceramic target (having a 1.8-cm diameter) of Zn_{0.3}Mn_{0.8}Si_{0.1}Fe_{1.8}O₄ (ZMSFO) (x=0.1) composition was intended for pulsed laser deposition (PLD) using Q-switched Nd: YAG laser (Ekspla Co., Model HT-303) having $\lambda = 355$ nm. The ferrite thin film was grown on Pt/SiO₂/ Si substrates with dimension of 15×15 mm². The laser pulse repetition frequency was kept at 10 Hz, and laser powers were fixed at 1.7 J for the PLD experiment [29]. During the film deposition process, temperature of the substrate was maintained at 973 K and oxygen partial pressure of 1mTorr was controlled. The ZMSFO film was deposited on the substrate with the aid of a metal shadow mask on an area of 10×15 mm², and film thickness was monitored to be ~ 260 nm. In order to create the top electrode (area $400 \times 300 \,\mu\text{m}^2$), highly pure Ag was evaporated thermally on the active layer of ferrite film through a metal shadow mask. The heterostructure Ag/ZMSFO/Pt/SiO₂/Si was characterized at 300 K by means of grazing incidence X-ray diffraction (GIXRD, PANalytical), atomic force microscopy (AFM) (Nanosurf AG), stylus surface profiler (DekTac. 3.2, Veeco) and I-V characteristics (Keithley 4200 semiconductor characterization system).

3 Results and discussion

The elemental analyses of synthesized compounds of the series, $Zn_{0.3}Mn_{0.7+x}Si_xFe_{2-2x}O_4$ (x=0.0-0.3), were conducted, and in Fig. 1 the representative EDAX pattern for the composition with x=0.2 is displayed. The peaks are well assigned in accordance with the standard positions. The weight percentage (wt%) of various elements determined from the chemical formulae and that from EDAX analysis correspond well mutually. The results summarized in Table 1 confirm the expected stoichiometry without the loss of any ingredient. The peak-to-background (*P/B*) ratio was found to be large



Fig. 1 Typical EDAX pattern for x=0.2 composition of the system $Zn_{0.3}Mn_{0.7+x}Si_xFe_{2-2x}O_4$

Table 1 The results of the elemental analysis of $Zn_{0.3}Mn_{0.9}Si_{0.2}Fe_{1.6}O_4$ (x=0.2) composition

Element	EDAX (wt%) ±2%	Expected (wt%)	(<i>P/B</i>) Ratio		
Fe	40.26	39.19	81.18		
Mn	21.63	21.68	46.08		
Zn	08.70	08.59	15.61		
Si	02.45	02.46	05.66		
0	26.96	28.07	54.37		

for the different peaks, so background fitting was free from any inaccuracy.

Precise knowledge of the cation distribution over crystallographic sites in magnetic oxides is highly essential for understanding their physical properties. In order to determine the distribution of metallic cations among the crystallographic interstitial sites, tetrahedral (A-) and octahedral (B-) sites, X-ray diffraction line intensity calculations were performed using the powder – x software [30] and indigenous software [31] based on the formula suggested by Buerger [32]

$$I_{\rm hkl} = \left|F_{\rm hkl}\right|^2 \cdot P \cdot L_{\rm p} \tag{1}$$

where I_{hkl} is the relative integrated intensity, F_{hkl} is the structure factor, P is the multiplicity factor and L_{p} is the Lorentz polarization factor = $[(1 + \cos^2 2\theta)/\sin^2 \theta \cdot \cos \theta]$. According to Ohnishi and Teranishi [33], the intensity ratios of planes I(220)/I(440), I(400)/I(422), and I(220)/I(400) are considered to be sensitive to any change in cation distribution. The intensities of (220) and (422) planes are mostly sensitive to cations on the tetrahedral sites, while the intensity of (400) plane depends on cations on both the sites. There is a good contrast in the atomic scattering factor of Si⁴⁺ and Zn²⁺ to that of Fe³⁺ and Mn²⁺, but the scattering factor of Mn^{2+} is close to that of Fe³⁺ [34]. We had estimated the amount of ferric (Fe³⁺) ions on the A-site and B-site through low-temperature (T = 80 K) ⁵⁷Fe Mossbauer spectral intensity calculations by considering the integrated areas under the Lorentzians corresponding to the A- and B-sites, which were taken as proportional to the amount of Fe³⁺ ion on these sites. The details are given elsewhere [35, 36]. The values of the Seebeck coefficient and absolute ferric ion concentration on the B-sites are further used to calculate ferrous ion (Fe²⁺) concentration as discussed in [35]. Any alteration in the distribution of cations causes a significant change in the theoretical values of the X-ray diffraction line intensity ratio. Therefore, in the process of arriving at the final cation distribution, the site occupancy of all the cations was varied for many combinations and those that agree well with the experimental intensity ratios, the fitting of the magnetization data at 80 K [26] and Mossbauer data analysis [36] were taken into consideration. The resultant cation distribution and comparison of intensity ratios of planes are shown in Table 2. The percentage accuracy in cationic occupancies is of the order of $\pm 5\%$.

Figure 2 depicts the current (*I*) against the electric field (*E*) plots for all compositions registered at various temperatures. It is found that for the pristine composition (x=0.0) switching action was recorded at $T_s \ge 373$ K, while for the compositions, x=0.1, 0.2, and 0.3, switching was noted, respectively, at $T_s \ge 323$ K, $T_s \ge 473$ K, $T_s \ge 523$ K. In

Mn-Si	<i>a</i> (Å) ±0.002 Å	Actual cation distribution	<u>/(400)</u> /(422)		<u>/(220)</u> /(400)		Oxygen deficiency	$(\Delta d/d) \times 10^{-3}$
(<i>x</i>)			Obs	Cal	Obs	Cal	(δ)	
0.0	8.465	$(Zn^{+2}_{0.25}Mn^{+2}_{0.40}Fe^{+3}_{0.35})$ $[Zn^{+2}_{0.05}Mn^{+2}_{0.30}Fe^{+3}_{1.02}Fe^{+2}_{0.63}]O_{3.68}$	1.924	2.103	1.122	1.264	0.32	-
0.1	8.488	$\begin{array}{l}(Zn^{+2}_{0.25}Si^{+4}_{0.10}Mn^{+2}_{0.43}Fe^{+3}_{0.22})[Zn^{+2}_{0.05}Mn^{+2}_{0.37}\\Fe^{+3}_{1.49}Fe^{+2}_{0.09}]O_{3.95}\end{array}$	2.107	2.257	0.992	1.052	0.05	1.3515
0.2	8.497	$\begin{array}{l}(Zn^{+2}_{0.25}Si^{+4}_{0.15}Mn^{+2}_{0.45}Fe^{+3}_{0.15})\left[Zn^{+2}_{0.05}Si^{+4}_{0.05}Mn^{+2}_{0.45}\right.\\ Fe^{+3}_{1.36}Fe^{+2}_{0.09}\right]O_{3.95}\end{array}$	2.071	2.191	1.294	1.206	0.05	1.8805
0.3	8.505	$\begin{array}{l}(Zn^{+2}_{0.25}Si^{+4}_{0.21}Mn^{+2}_{0.48}Fe^{+3}_{0.06})[Zn^{+2}_{0.05}Si^{+4}_{0.09}Mn^{+2}_{0.52}\\Fe^{+3}_{1.24}Fe^{+2}_{0.10}]O_{3.95}\end{array}$	1.800	1.951	1.066	1.189	0.05	2.3705

Table 2 Lattice constant (a), distribution of cations, oxygen deficiency (δ) and microstrain ($\Delta d/d$) for Zn_{0.3}Mn_{0.7+x}Si_xFe_{2-2x}O₄ system



Fig. 2 Current (I) against electric field (E) characteristics for x = 0.0–0.3 compositions registered at selected temperatures

accordance with T_{s} , electric field strength (E_s) required for switching action to takes place is found to be 1416 V/ cm for x = 0.0 composition, while for x = 0.1, 0.2 and 0.3 compositions E_s is found to be 1206 V/cm, 1276 V/cm and 1321 V/cm, respectively. In short, T_s and E_s increase with Mn–Si content (x) for $x \ge 0.1$. According to Vaingankar et al. [37], electrical switching is influenced by structural and microstructural parameters. The observed increase in the lattice parameter with x (Table 2) implies that charge carriers require more amount of energy to transfer from one cationic site to another. Thus, the observed increase in T_s and E_s may be correlated with a subsequent increase in the unit cell dimension. The compositional variation of lattice constant has clearly implied that in the system greater proportion of Si⁴⁺ ions residing at the grain boundaries, while a small quantity did indeed enter into the crystal lattice [24, 26]. Earlier, it has been shown in the case of W-type barium ferrite ($BaCo_{1.3}Zn_{0.7}Fe_{16}O_{27}$) that grain boundary occupancy of Si⁴⁺ ions is 5–10 times greater than Si⁴⁺ ions residing inside the grains [38]. Our results are consistent with these findings. The presence of silicon ions at the grain boundaries forms a low-conductivity layer as addressed above. Thus, on increasing the concentration (*x*) intragranular flow of charge carriers get restricted, which turns out in observed increase in E_s and T_s with content (*x*). Earlier, in the case of the $Cd_xCo_{1-x}Fe_{2-y}Cr_yO_4$ spinel ferrite system, it was reported that the switching field (E_s) decreases with a decrease in grain size (*D*) [14]. In the present case, an average grain size reduces from 12 µm (x = 0.0) to 6 µm (x = 0.3) [28], but at the same time E_s is found to increase with (x). This contradicts the previous results.

A careful examination of Fig. 2 revealed the nature of the switching performance. The deviation from the Ohm's law becomes more prominent with an increase in *V* across the sample when the *E* is just beyond the value designated by 'A.' During the first cycle, the breakdown abruptly takes place and the field drops to the value marked by B, specified as the first breakdown field; hereafter, the *I* increases monotonously, from B to C. The aggregate resistance

switching ratio (R_{high}/R_{low}) approximated for the different compositions is better than 200% at the respective T_s .

The aggregate ON-state current is noted to be ~ 179 mA at the distinct temperatures for all the compositions. It is interesting to note that the switching performance of the samples remains unchanged even when the cycle was performed again after 2 weeks indicating the absence of the 'aging effect.' On the other hand, when the ferrites were under the second switching cycle that was running later than the first cycle, it was observed that the lower value of the electric field is needed for breakdown to take place. This leads to the occurrence of Joule self-heating.

As discussed above, the large class of oxide systems exhibits this interesting phenomenon. Thus, it is difficult to pinpoint any precise reason as being responsible for electrical switching and it may be different for the different systems. Commonly, oxide series containing Jahn–Teller (JT) ions (Mn³⁺ and Cu²⁺) are the most probable candidates expected to demonstrate switching actions [39].

The series $Zn^{+2}_{0.3}Mn^{+2}_{0.7+x}Si^{+4}_{x}Fe^{+3}_{2-2x}O^{-2}_{4}$ is fundamentally free from JT ions. Thus, the observed switching action cannot be ascribed to JT ions as suggested. The structural transformation observed [37] likewise cannot be admitted as the origin of this action, as neither of these ferrites demonstrated such transformation in structure as a function of temperature, which otherwise would have been observed in thermal variation of two-probe dc conductivity measurements [26]. The oxygen deficiency (δ), which is customary in aforesaid ferrites, should be allowed for the possibility, but this is most unlikely as verified by EDAX pattern analysis. Further, thermoelectric power measurements have shown that $\delta = 0.05$ remains constant for x = 0.1 - 0.3 compositions [35] (Table 2). Thus, oxygen deficiency cannot be responsible for observed switching action and corresponding changes in T_s and E_s values. Similar observations of the trivial role of oxygen vacancy on RS have been reported for NiO and perovskite manganite $La_{0.67}Sr_{0.33}MnO_3$ [40, 41]. The formation of JT Mn^{+3} ion is presumably in such systems, and concurrently, the formation of a comparable amount of JT ferrous ions (Fe^{+2}) from ferric ions (Fe⁺³) is anticipated such that the charge neutrality retains. If switching in the system is accorded to the presence of Mn^{+3} ions in the series, then requisite E_s and T_s showed a decrease with Mn–Si content (x), but the findings show an opposite trend.

In addition, thermoelectric power measurements have also demonstrated that the ferrites under study are n-type semiconducting materials. Consequently, switching owing to the cooperative Jahn–Teller distortions does not appear to be tenable. On examining the cation distribution (Table 2), it is evident that on Mn–Si substitution (x), Fe³⁺ ion concentration decreases from the B-site as well as from the system itself. This restricts the exchange of electron

among Fe³⁺-Fe²⁺ ions, and as a consequence, the values of E_s and T_s required for switching action increase. When highly magnetic Fe³⁺ ions are replaced by non-magnetic Si⁴⁺ ions that do not take an active part in the conduction mechanism, that curtails the conduction via the B-site. Additionally, the space-charge-limiting current (SCLC) may also be responsible for observed switching action. As stated in the band theory of insulators, supplemental charge carriers are injected through a metallic electrode via an insulator-metal interface. In Fig. 3, it can be seen that for low-voltage region, up to 2.3 V, the current increases almost linearly, suggesting the slope value of log/versus log/ curves to be approximately 1 for all the compositions. However, beyond 2.3 V, a sharp increase in the current deviates from the linearity and yields a slope of the curves between 1.8 and 2.8 (Fig. 3), which implies that in the high resistance state the conduction mechanism is static induction current (SIC) or SCLC-type. Similar findings and detailed mechanisms are reported in refs. [29] and [42] for binary and ternary oxide compounds, respectively. In SCLC over the insulating layers, increase in large current takes place when trapping sites in the insulating bulk are fully occupied at a threshold voltage (V_{T}) . By and large, $V_{\rm T}$ is farther than 200 V in the primary action of RS. Beyond the V_{T} , I swiftly increases (Fig. 3). The drastic rise in I may be induced by trap charge—SCLC. Thus, when the thermally produced free charge carrier density within the system is higher as compared to the density of charge carrier injected, Ohmic (IaV) characteristic is noticed. In the manner now being exemplified from the log/ against logV plots (Fig. 3) which exhibit that at low voltage, I follows Ohm's law, and at breakdown voltage, the slope of the curve increases hinting the presence of SCL current which could be referring to as the possible cause of electric switching [43]. This is also an essential feature of CCNR type of electrical switching. Eventually, the *I–E* characteristics show gradual switching that can be thought of due to CCNR-type switching. On the other hand, fast switching action may be expected on account of VCNR-type switching. Further, I–E characteristics (Fig. 2) exhibited by all the compositions regardless of ferrimagnetic or paramagnetic in nature and are akin to the ON-OFF characteristics of a silicon-controlled rectifier. At long last, an effort has been made to correlate microstructural parameters [grain size (D) and density of defects (d_D)], similar to that of structural parameters (the lattice constant (a) and cation distribution), with electric parameters, E_s and nonlinearity coefficient (α). The values of α and E_s are deployed to characterize the functioning of a device. It is advisable to have a large value of α to protect the device against any electric surges. The a value determined from standard method [22], for example, increases from $\alpha = 2.0$ (x = 0.0) to $\alpha = 2.8$ (x = 0.3) at T = 523 K, consistent with the slope



Fig. 3 Temperature dependence of log/ versus log/ plots for the $Zn_{0.3}Mn_{0.7+x}Si_xFe_{2-2x}O_4$ spinel ferrite series

values determined from log I versus log V plots (Fig. 3), as expected. It has been reported that $d_D \propto 1/D^2 \propto E_s^2$ [22]. On decrease in *D* with content (*x*), d_{D} is found to increase from 6.94×10^9 /m² for x = 0.0 composition to 27.8 × 10⁹/m² for x = 0.3 composition, suggesting an increase in microstrain due to the enhancement in tensile force acting on the planes. The variation in lattice microstrain, corresponding to a fractional change in plane spacing, $\Delta d/d$, has been computed for the substituted compositions by taking into account 'd' spacing value of the most intense (311) peak and is presented in Table 2. This value of $\Delta d/d$, however, comprises both tensile strain and compressive strain and must be divided by a factor of two to get the maximum tensile strain alone, if these two strains are presumed to be the same in strength. Table 1 reveals that strain gradually increases with increasing Mn–Si content (x) for x = 0.1-0.3compositions. A single-peak analysis method [strain determination by the value of $\Delta d/d$ from only (311) reflection] is based on the assumption that strain broadening is Gaussian, while size broadening is Lorentzian. Other methods analyze the full diffraction pattern, trying to extract the information about the crystallite size and strain based on the different angular dependence of their respective broadening effects. A relatively simple and thus still widely used method is the Williamson–Hall (W–H) plot. This method actually considers the broadening (β) of peaks as a function of a diffraction angle (2θ), which is assumed to be the combined effect of size-induced broadening and strain-induced broadening. In this method, both size and strain broadening are assumed to be Lorentzian, and the integral breadth of both components is additive.

From the W–H plots (Fig. 4), the linear slope gives the value of the lattice strain. It is found that the system $Mn_{0.7+x}Zn_{0.3}Si_xFe_{2-2x}O_4(x=0.0-0.3)$ comprises positive and negative values of strain. The positive slope for x=0.0 composition suggests the presence of tensile strain



Fig. 4 Williamson and Hall plots for strain determination for all the compositions of the spinel ferrite series $Zn_{0.3}Mn_{0.7+x}Si_{y}Fe_{2-2x}O_{4}$

(strain = $+0.865 \times 10^{-4}$), whereas the negative slope for x = 0.1, 0.2 and 0.3 compositions exhibits compressive strain [44] (strain = -0.606×10^{-4} , -1.59×10^{-4} and -1.93 \times 10⁻⁴, respectively). All the plots exhibit a scatter point known as a 'deviation pattern,' which is typical of strained cubic materials [45]. The co-substitution of Mn–Si causes significant changes in the structural properties and eventually affects other physical properties. The compressive strain also indicates that microstrains cannot be a dominant source of broadening [46]. On the other hand, in coarse-grained materials, crystallite/grain size has a negligible effect on X-ray diffraction line broadening. The compressive strain increases with increasing content (x) in the system. The variation of strain with x is similar to that of strain determined from single (311) reflection for x > 0.0compositions.

The tensile strain increases owing to the increasing defect density (d_D) as discussed above. Earlier, a detailed role of defects in governing the RS has been reviewed for several oxide thin film systems [47]. Thus, one can expect better nonlinear behavior (higher value of α), as observed. At the same time, as a result, an increase in E_s

with Mn–Si concentration (x) is anticipated as noticed for x = 0.1-0.3 compositions. These experimental outcomes are of great significance as they bring out the advancement of electric field-induced RS devices.

Figure 5 displays the GIXRD pattern and AFM image of the ZMSFO thin film. The diffraction line analogous to the single face-centered cubic (fcc) phase is clearly seen. The lattice parameter anticipated based on the GIXRD analysis was 8.43 Å, which is smaller as compared to the bulk counterpart (a = 8.488 Å) (Table 2). One or more of the following factors causes the smaller value of unit cell parameter for nanostructured thin film: (1) low degree of crystallinity in nanostructured thin film, (2) the alteration in the site occupancy of cations and (3) the existence of lattice defects that affect the surface of the nanoparticles. Surface energy and tension of the nanoparticles are significant that lead to shrinkage in the lattice, which elicits a reduction in lattice constant [48]. AFM image shows a homogeneous distribution of grains, and the aggregate particle size approximated is varying between 50 and 65 nm. The AFM analysis has shown RMS



Fig. 5 GIXRD pattern of ZMSFO film grown on Pt/SiO₂/Si substrate. A single cubic phase can be seen (**b**). Bi-dimensional AFM surface topography of ZMSFO/Pt/SiO₂/Si heterostructure



Fig. 6 *I-V* characteristics of Ag/ZMSFO/Pt/SiO₂/Si heterostructure. The inset shows the schematic of measurement geometry and arrows indicate typical voltage sweeping cycles

roughness of the order of ~ 6.8 nm that is equitably modest alluding to even-textured grain morphology.

Figure 6 illustrates *I–V* characteristics of Ag/ZMSFO/Pt/ SiO₂/Si heterostructure examine by employing the bi-terminal method. The inlay shows a schematic of a cross section of the Ag/ZMSFO/Pt/SiO₂/Si sandwich configuration intended for the *I–V* measurements. Figure 6 shows linear *I–V* behavior for every cycle of voltage sweeping implying the absence of RS characteristics. These outcomes point out that the electrical transport associated with the interface is most unlikely for the RS phenomenon in the ferrite thin film.

4 Conclusions

In conclusion, spinel ferrite system $Zn_{0.3}Mn_{0.7+x}Fe_{2-2x}O_4$ (x = 0.0-0.3) in its bulk and thin film forms have been studied for its current-controlled negative resistance-type

SN Applied Sciences A Springer Nature journat electrical switching properties. Elevated temperatures and high electric fields are essential parameters along with the Mn–Si concentration (x) for the observed enhancement in the electrical switching in the ferrite oxide system. Moreover, the expansion in lattice parameter, abatement of octahedral site occupancy of Fe³⁺ ion in spinel lattice, and partial grain boundaries occupancy of Si⁴⁺ ions may have an important role for the switching characteristics. The modeling of the I-V data confirmed the presence of space-charge-limiting current, which could be a responsible source of observed switching. The PLD grown ferrite thin film shows linear behavior in current against voltage characteristic without any signature of resistive switching, suggesting that the bulk composition of the compound has a major role to play against the film-electrode interface.

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Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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