



NanoCarb hyperspectral sensor: on performance optimization and analysis for greenhouse gas monitoring from a constellation of small satellites

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Abstract

NanoCarb is an innovative Fourier-transform imaging spectrometer dedicated to the measurement of CO₂ and CH₄. Both its unusual optical principle and sampling strategy allow to reach a compact design, ideal for small satellite constellation as investigated by the European project SCARBO. The NanoCarb performance assessment as well as a proof of concept is required in this framework. We have developed a design strategy to optimize the performances. We demonstrate the potential of the concept through an estimation of the sensitivity, compliant with the space mission target. We also present a preliminary mitigation of the bias induced by water on CO₂ and CH₄ retrieval, illustrating the efficiency and the flexibility of the NanoCarb partial interferogram sampling technique. The presented design reaches a sub-ppm random error for CO₂ and sub-10 ppb random error for CH₄, considering 128 km swath and 2 by 2 km² ground resolution. Design optimization and more systematic performances are discussed.

Keywords Hyperspectral sensor · Anthropogenic GHG emissions · Fourier-transform spectroscopy · Near-IR passive remote sensing

1 Introduction

One of the issues of studying climate change is to reduce uncertainties in estimating emissions of the main greenhouse gasses (GHG)—CO₂ and CH₄ [1]—leading to a better distinction between anthropogenic and natural sources. The first challenge for dedicated space missions is to monitor the atmospheric total column and/or vertical profiles with both a drastically improved statistical error and smaller systematic

bias. The second one is to provide an improved revisit frequency and spatial coverage of the monitoring.

LIDAR active technology (MERLIN European project [2] or ASCENDS project in the US [3]) achieves sensitive measurements independently to solar illumination conditions, in like manner dedicated spaceborne Thermal InfraRed (TIR) passive spectrometers (IASI [4]). Nevertheless, LIDAR are inherently limited to unidirectional sounding, while TIR spectrometers are less sensitive to the lowest atmospheric layers where anthropogenic activities are significant. By the opposite, Short Wave Infrared (SWIR or near IR) spectrometry only allows daytime monitoring, but with improved sensitivity to the lowest parts of the atmosphere. GOSAT [5] and GOSAT-2,¹ OCO-2 (see footnote 1) or TANSAT (see footnote 1), are current space missions dedicated to CO₂ and/or CH₄ monitoring in this domain. Geo-CARB (see

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footnote 1) in the US, or MicroCarb [6], CO2M (CarbonSat) [7], and TROPOMI (see footnote 1) in Europe are the future planned missions or spaceborne instruments.

Nevertheless, not all these missions can achieve sufficient spatial and temporal coverage, with a typical revisit frequency of 15 days for the majority of them, from a sun-synchronous orbit. In addition, individual costs of these missions are a real hurdle to put into orbit a constellation.

A way forward to increase the revisit frequency and spatial coverage is to complete these current reference missions with a constellation of low cost, small satellites, and calling for breakthroughs in compact hyperspectral sensing. The challenge is to adapt kilogram-class payload for typical 50 kg-class platforms, maintaining their performances compatible with current reference missions and the main science issues: kilometric ground resolution, sensitivity below 1 part per million (ppm) over the averaged dry column of CO₂, and below 10 parts per billion (ppb) for CH₄. Moreover, it is important to reach a 100 km swath to increase the spatial coverage and to contextualize some hotspots according to the background (e.g., plume).

In this framework, the Horizon 2020 project Space Carbon Observatory (SCARBO) [8] aims at assessing the feasibility of a low-cost constellation of small satellites onboarding dedicated GHG sensor. The SPEX [9] dedicated aerosols sensor is also provided in the platform. The project relies on small satellites to monitor CO₂ and CH₄ emissions, complementing the low-revisit high-performance satellites.

Several orbital configurations are investigated in this scenario. One of the simplest consists to put 40 small satellites into the same sun-synchronous orbit at 600 km height. With a 128 km swath and a small recovering between consecutive footprints at ground for cross-calibration purposes, it is possible to ensure a daily revisit over emerged lands at mid-latitudes.

The core miniature GHG sensor of this constellation is the NanoCarb concept, a static Fourier-transform imaging spectrometer, first introduced in sections III.B and III.C of [10]. In this concept, both the use of a low finesse Fabry–Perot array and a partial interferometric sampling strategy allow to achieve a large swath at high spectral resolution as well as to obtain an optimal use of the available pixels for a high sensitivity in a snapshot acquisition mode.

However, this uncommon instrumental concept enforces a shift of paradigm in atmospheric sounding domain. Indeed, the acquired data are quite different from classic dispersive or Fourier-based spaceborne systems: the information content of the data is very low (as defined in [11]), compared with a conventional radiance spectrum. Consequently, a careful upstream design of the instrument is required. The main issue is to maximize the sensitivity to the average column of CO₂ and CH₄, while minimizing biases due to other geophysical parameters.

In this paper, we introduce a forward approach to design NanoCarb, and then we analyze its performances for CO₂ and CH₄ atmospheric total column measurement. We base this approach on an analytical model of the NanoCarb radiometric sensitivity coupled with considerations over information present in the Fourier domain. We present a preliminary design of the two main components of the instrument which are the narrowband filter and the interferometer, and we investigate some promising features of the concept.

In Sect. 2, we present the NanoCarb principle and sampling strategy. Then in Sect. 3, we describe our radiometric model (direct or forward model), which is used in Sect. 4 to select the optimal spectral band, and in Sect. 5 to select the optical path differences (OPD). We derive and discuss some considerations about performance capabilities of the concept.

2 Nanocarb concept principle

In this section, we describe the principle of the NanoCarb spectrometer: optical principle and sampling strategy. A description of the expected data products is also presented, to introduce some potential calibration or data processing issues, even though the purpose of this paper is only focused on theoretical description and preliminary design.

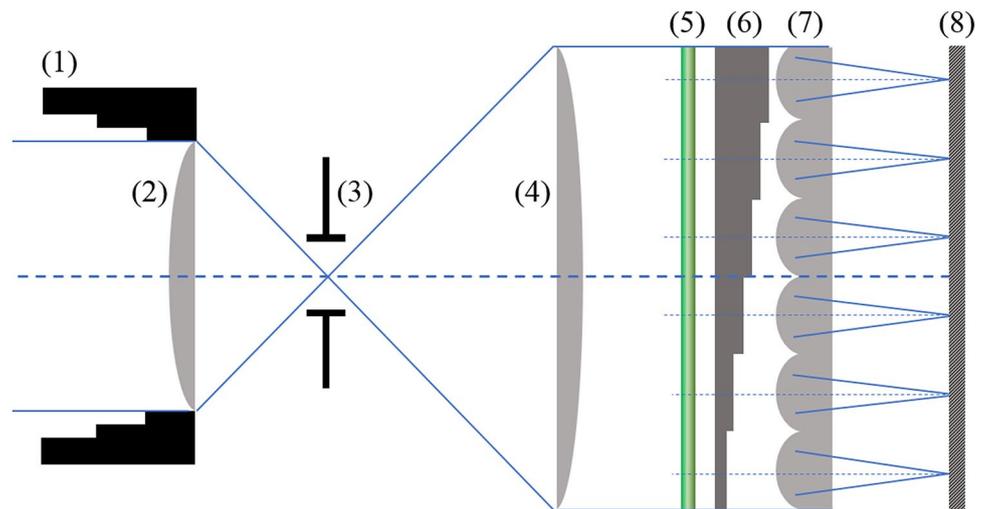
2.1 Optical principle

NanoCarb is a static Fourier-transform imaging spectrometer. We can distinguish two optical sub-systems: the front optics and the interferometric core, as it can be seen in Fig. 1.

The front optics consist of an afocal system—(2) and (4)—with a field stop (3), allowing to adapt the Field-of-View (FoV) independently from the interferometric core (5–7). A baffle (1) stays in front of the afocal system. An interference filter (5) selects the spectral band. Four independent spectral bands have been currently identified for SCARBO as reported in Table 1. In the scope of this paper, we consider only the B2 and B3 bands to optimize CO₂ and CH₄ sensitivity, assuming a clear sky and a perfect measurement of the atmospheric pressure, thanks to band B1.

The interferometric core is formed by the association of an interferometer array (6) with a microlens array (μlens—7). Each interferometer has a specific thickness. The focal plane array (FPA—8) is placed in the focal plane of the μlens array, to obtain a set of thumbnails, each one being associated to one thickness of the interferometric plate. In this configuration, the image formed in each thumbnail is a replication of the same FoV, modulated by the associated interferometer.

Each interferometer is a low-finesse Fabry–Perot (FP), which modulates the spectrum of the incoming light in accordance with the interferometer optical path difference (OPD),

Fig. 1 Optical principle of NanoCarb**Table 1** Current NanoCarb bands

Band	Region	Measurement
B1–O ₂	760 nm	Surface pressure, aerosols
B2–CO ₂	1.6 μm	CO ₂
B3–CH ₄	1.66 μm	CH ₄
B4–strong CO ₂	2.06 μm	Aerosols

which depends mainly on the Fabry–Perot thickness and slightly on the angle of incidence (this latter dependency creating the rings which modulate the image on each thumbnail, see Fig. 4). The currently investigated device is based on silicon, offering a natural reflection at the air–Si interface about 55%, which leads to two-wave-dominated interferences. This configuration is optimal to manage at the same time the compactness of the device and large FoV, as well as sensitivity as explained in [12] with more details. The device can be replicated, each one filling a part of the same FPA or using a dedicated FPA, to monitor synchronously several spectral bands.

Because interferential filter lies in the pupil plane in front of μFP array, a spectral band shift occurs in the field of view of each thumbnail. We integrate this shift in the NanoCarb numerical model and consider it in the design of the instrument.

This concept is compact and fully static. In the following, we will explain the partial interferometric sampling strategy of the NanoCarb concept, enabling snapshot acquisitions at high spectral resolution.

2.2 Partial sampling of the interferogram

The strategy we adopt to reach a high spectral resolution while keeping a large swath in snapshot mode is to target

only the useful information in the Fourier space. Thus, the NanoCarb spectrometer acquires only partial interferograms.

The acquisition of partial interferograms was already developed in the 1970s by Kyle for temperature measurement in CO₂ lines [13], then by Fortunato who applied this method to the measurement of SO₂ concentration [14]. More recently, some studies can be found for the SIFTI instrument [15], and for IASI data processing [16], showing both a better geophysical bias mitigation, such as surface temperature, and an improvement of the signal-to-noise ratio (SNR).

In the case of NanoCarb, the spectral band is purposely selected to optimize the sensitivity over targeted regions of the interferogram, leading to an optical filtering of the useless information (interferants) both in the spectral and in the Fourier domain. An interferential filter as introduced in the previous sub-section is used to select the spectral band (spectral filtering), in front of the μ-interferometer array (interferometric filtering). This principle is well suited for CO₂ and CH₄, thanks to a periodic spectral pattern over specific bands, for example on the 1.6 μm band, or on the 1.66 μm band for CH₄ as visible in Fig. 2. In these conditions, the signature of these two species in the Fourier domain is well concentrated as visible in Fig. 3.

Hence, the aim of the partial interferogram sampling is:

- To obtain a measure of the *X*-specie concentration, as independently of the interferants as possible (for example, water).
- To obtain a measure of the interferants where appropriate, as independently as possible. The main atmospheric interferants are the aerosols and the water vapor. Temperature profile, pressure at ground, and albedo are also crucial in the retrieval process, and consequently require dedicated spectral or interferometric channels.

Figure 3 shows some partial derivative interferograms for a variation of concentration of CO₂, CH₄, and water, for the

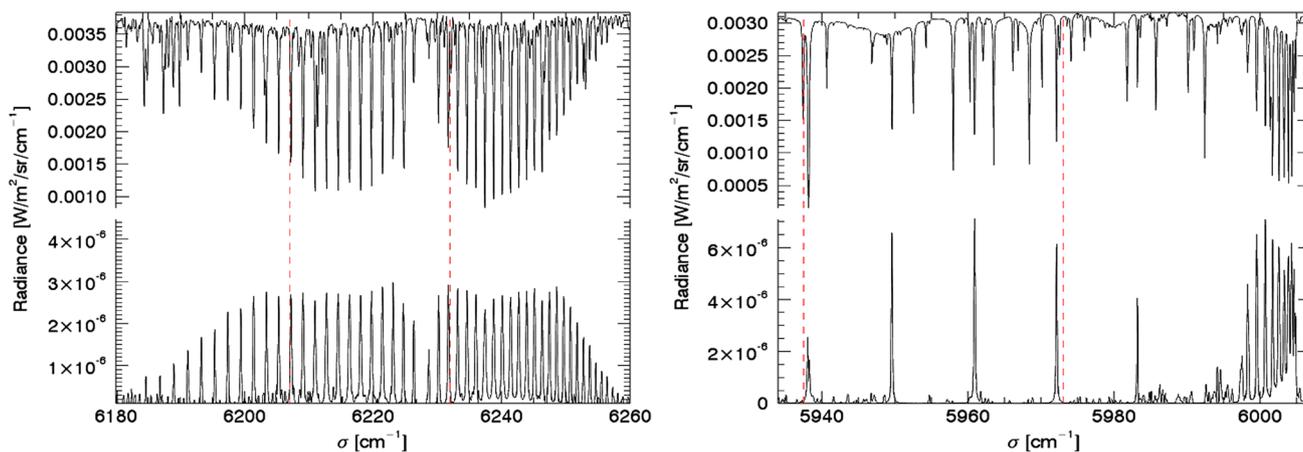


Fig. 2 Simulated spectral radiance (top curve) and partial derivative radiance (bottom curve). Left: on 1.6 μm CO₂ band, with a variation of 1 ppm of CO₂ over the total column for the derivative radiance. Right: on 1.66 μm CH₄ band, with a variation of 10 ppb of CH₄. The

spectral regions between the red dashed lines allow to optimize the sensitivity of NanoCarb and are targeted by the relevant interferometric filter

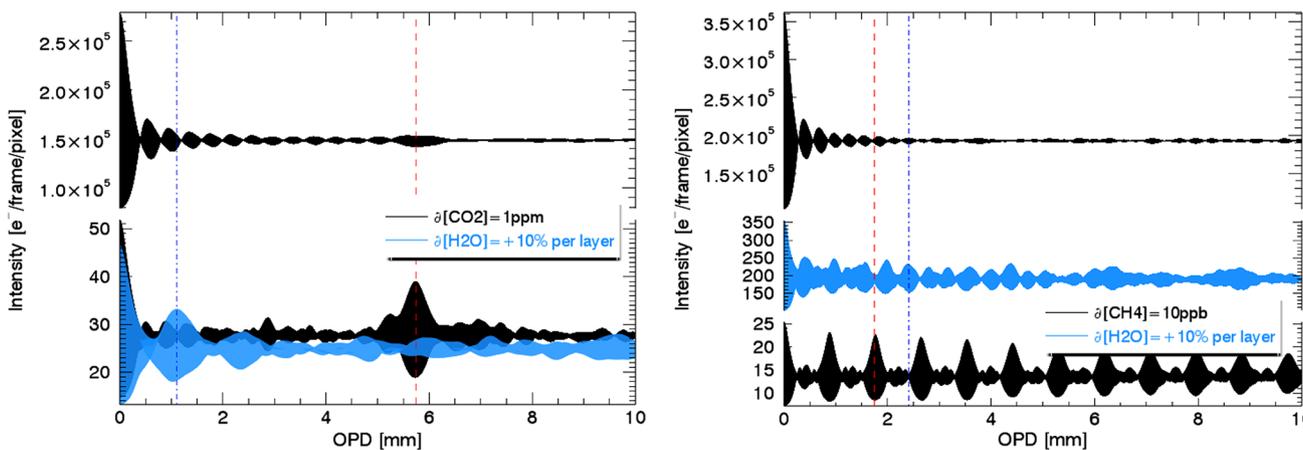


Fig. 3 Interferogram (top curve) and partial derivative interferograms (bottom curve). Left: for the CO₂ band, with a variation of 1 ppm of CO₂ for the black-plotted derivative. Right: for the CH₄ band, with a variation of 10 ppb of CH₄ for the black-plotted derivative; in blue,

the partial derivative for a variation of 10% of the total column of water for each band. The red dashed lines show the OPD where the sensitivity is optimal. The blue dotted dashed lines show a maximum of sensitivity to water. Albedo=0.2, solar angle=20°, iFoV=2 km

spectral band chosen as an example. The red dashed lines highlight the regions where the sensitivity of CO₂ or CH₄ is optimal, but biased by water, while the blue ones spot regions where the measurement of water is optimal with respect to CO₂ or CH₄.

Figure 4 shows how with a controlled distribution of thickness over the FP array we can sample the interferogram at targeted OPD, potentially through disjoint interferometric regions. For a given region, the thicknesses are chosen to achieve a continuous $\lambda/4$ -sampling. First, this allows to obtain a good sampling of the interferogram on a single acquisition, and thus enables snapshot acquisition mode; second it is more convenient for fringe contrast estimation based on generalized ABCD techniques [17].

A particular attention must be paid to the choice of spectral bands and sampled OPDs when designing NanoCarb. These points will be detailed, respectively, in Sects. 4 and 5.

2.3 NanoCarb data products

The L0 data product of NanoCarb is the snapshot focal plane intensity acquired by the detector. For demonstration purpose, a noiseless simulation can be seen in Fig. 4—left, in the unrealistic case of a spatially and spectrally uniform scene.

The colored points spot an individual field of view (iFoV) imaged in all the thumbnails. The extraction of the intensity

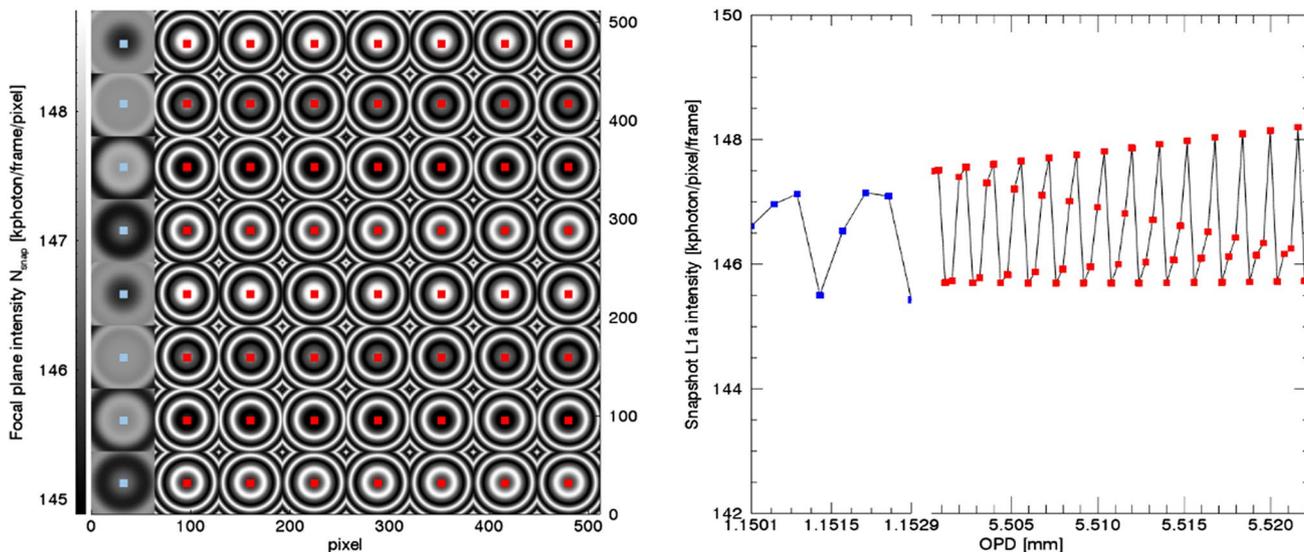
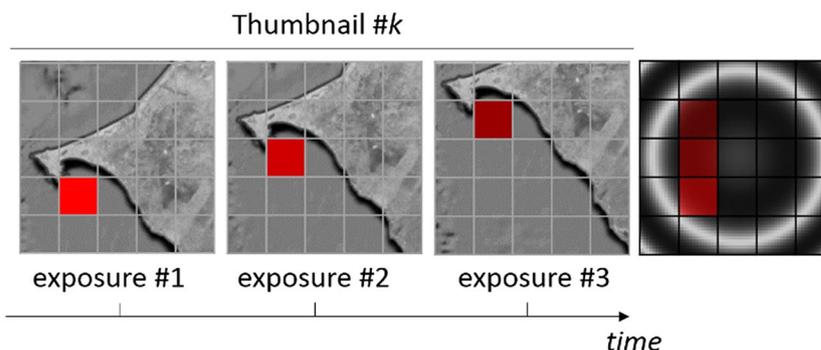


Fig. 4 Left: simulated focal plane intensity of NanoCarb in the 1.6 μm CO₂ band, equivalent to L0 data product. Right: extracted snapshot interferogram for the spotted iFoV, equivalent to L1a data product. In this configuration, we target two disjoint interferometric

regions. The left thumbnail column is dedicated to the sampling of particularly water-sensitive fringes, while the rest of the thumbnails are dedicated to CO₂

Fig. 5 iFoV shifting across the thumbnail as a function of time. The right image illustrates how the intensity is modulated over each exposure with the Fabry–Perot rings crossing. In that way, we can measure several interferometric states for this iFoV



on a single exposure for this iFoV allows to retrieve the associated partial interferogram (Fig. 4-right), assuming a lab calibration of the corresponding OPD (see [18] and [19]). This snapshot interferogram is the **L1a data product** of NanoCarb. For a given iFoV at this level, n_{FP} , the number of thumbnails—or FP—per spectral band, is the number of interferometric channels.

The L1b data product is the final vector of measures to be inverted. It is the concatenation of all the consecutive snapshot exposures during the scan of the scene as illustrated in Fig. 5. Between each exposure, the interferometric state changes (Fabry–Perot rings), as well as the spectral response of the pixel. This latter is due to the spectral shift of the interferential filter in the field. For this reason, a representation of a L1b interferogram makes no sense on the OPD scale used in Fig. 4. This co-registration of snapshot acquisitions is crucial to reach a sufficient SNR as we will see later in this paper. For a given iFoV at this

level, $n_{FP} \times N_s$ is the maximum number of interferometric channels per spectral band, with N_s the number of pixels by side of thumbnail (assuming that the satellite has moved one iFoV forward between two frames).

The L2 data product is the concentration value for a single iFoV, retrieved from this L1b concatenated partial interferogram. In the case of a multi-band design, the measurement vector to be inverted is the concatenation of all the L1b concatenated partial interferograms from each one of these bands.

3 Analytical radiometric model of performances

We propose here a solution to estimate the sensitivity of NanoCarb as a function of the instrumental parameters. Model of intensity over the FPA at each level of data

is introduced as well as derived analytical performance expression. In this model, the scene is assumed to be homogeneous at the scale of the spatial resolution.

3.1 NanoCarb instrumental model

We use a simple analytical model of the NanoCarb intensity over the focal plane, derived from the approximate expression of the transmission of a Fabry–Perot assuming Fresnel reflections at the interfaces. This normalized transmission in intensity for a monochromatic radiation is:

$$T_{\text{FP}} \approx \frac{1}{1 + M \sin^2\left(\frac{\varphi}{2}\right)}, \quad (1)$$

$M = 4R/(1 - R)^2$ is the finesse term depending on the reflectivity at the FP interfaces R and φ the phase shift between two consecutive transmitted rays:

$$\varphi = 2\pi\sigma\delta, \quad (2)$$

where σ is the wavenumber and δ is the OPD.

A weighted integration of Eq. (1) over the considered spectral band $\Delta\sigma$ allows to derive the number of electrons per pixel on a single frame on the detector N_{snap} :

$$N_{\text{snap}} = \eta \int_{\Delta\sigma} \frac{\left(\frac{1}{hc\sigma}\right) T_{\sigma} L_{\sigma}}{1 + M_{\sigma} \sin^2\left(\frac{\varphi(\sigma)}{2}\right)} d\sigma, \quad [e/\text{frame/pixel}], \quad (3)$$

where L_{σ} is the spectral terrestrial radiance (see next subsection), T_{σ} the normalized spectral transmission of the instrument including, for instance, detector quantum efficiency and interferential filter spectral shape, and η is the σ -independent radiometric term, gathering exposure time t_{exp} , single pixel optical etendue ΩS , and panchromatic optical transmission of the instrument T_{opt} :

$$\eta = \Omega S t_{\text{exp}} T_{\text{opt}}, \quad [\text{strd.m}^2 \cdot \text{s}]. \quad (4)$$

Equations (1–4) allow to simulate 1D-interferograms with NanoCarb radiometric properties on an arbitrary scale of OPD as presented in Fig. 3.

Such a kind of interferogram is very convenient to explore information content and sensitivity in the Fourier domain, by deriving fringe visibility V . The visibility trend with the OPD, $V(\delta)$, paints the interferometric shape or envelope of the interferogram. This latter is estimated as follows:

$$V(\delta) = F(\delta)/\bar{N}, \quad (5)$$

where $F(\delta)$ is the coherent flux at δ in electron per frame and per pixel and \bar{N} is the mean level of the interferogram in the same unit. Band optimization and interferometric region

targeting described in the following sections are based on this envelope estimation.

3.1.1 On NanoCarb data product models

We can easily simulate 2D-images of the NanoCarb L0 data product (e.g. Fig. 4) from Eq. (3) by implementing NanoCarb geometry, including FP thickness distribution and angular dependency in the field of view of each thumbnail. The main feature is an OPD variation in the FoV responsible for FP ring patterns, explicitly stated in the following expression:

$$\delta(\sigma, i, \theta, T) = 2n(\sigma, T)\varepsilon_i \cos\theta_r, \quad (6)$$

where T is the temperature, ε_i the mechanical thickness of the i th FP, n the optical refractive index of the FP material, and θ and θ_r , respectively, are the incident angle over the FP and the related refracted angle in the FP. A θ -dependent shift of the filter spectral band can be integrated at this level in T_{σ} .

We overlook, in this paper, the point spread function (PSF) of the microlens array, as it can be shown that at least 90% of the fringe visibility can be saved with a Nyquist sampling of the PSF. A convolution of the 2D NanoCarb image with the effective PSF could simulate this effect with a quite good accuracy.

Since we only consider homogeneous scene, this model can also be used to simulate NanoCarb L1a product, that is the partial 1D-interferogram at a given iFoV (given θ).

Noise is added on this interferogram, taking into account photon noise (Poisson distribution) and additive read out noise (RON), with normal distribution.

3.1.2 On thermal issues

The T dependency in Eq. (6) induced by temperature dependency of the optical index of silicon enables a thermal analysis of the system based on the model described above. This more advanced study is out of the scope of this paper, which focuses on the sensitivity of the concept for GHG measurement. However, preliminary investigations have shown that the interferometric envelope variation induced by a 0.1 K delta- T is one magnitude smaller than the one induced by a 1 ppm variation of CO_2 . Such a thermal regulation over a 0.1 s exposure is fully achievable with current cooling devices.

3.2 Atmospheric model

Terrestrial radiance L_{σ} is simulated using the Standard US Model [20] for temperature and pressure profiles as well as atmospheric composition and concentration profiles. This model is implemented in the radiative transfer code LBLRTM [21], using the spectroscopic database HITRAN

[22]. We simulate nadir terrestrial radiances with LBLRTM in the 1.6 μm and 1.66 μm band, taking into account fine spectroscopic effects such as line blending.

For this preliminary work, we assume a clear sky. CO₂ and CH₄ retrieval with input measurements of a joint aerosol-dedicated instrument (SPEX [9]) will be implemented later in the framework of SCARBO. By focusing on the 1.6 μm and 1.66 μm band, we assume also that the atmospheric pressure is perfectly measured in the NanoCarb dedicated band.

3.3 Analytical noise model

We aim here at deriving an expression of the radiometric noise in ppm or ppb of the considered specie *X* over the detector, to deliver a preliminary design of the instrument. A solution is to compare the required sensitivity for a targeted concentration variation, to the effective SNR over the fringe visibility measurement.

The required sensitivity $S_{\partial[X]}$ to measure a given concentration variation $\partial[X]$ of the specie *X* can be expressed as the ratio of variation of the coherent flux F_X in the targeted interferometric region:

$$S_{\partial[X]} = \frac{\partial F_X}{\partial[X]} / F_X. \tag{7}$$

Replacing $V_X = F_X / \bar{N}_{ph}$, the mean fringes contrast in the targeted Fourier region and \bar{N}_{ph} , the total mean number of photo-electron contributing to the coherent flux estimation, we obtain:

$$S_{\partial[X]} = \frac{\partial V_X}{\partial[X]} / V_X + \frac{\partial \bar{N}_{ph}}{\partial[X]} / \bar{N}_{ph}. \tag{8}$$

We now introduce $S_{ph,X}$, the effective sensitivity over the coherent flux, expressed as the inverse of the SNR over the coherent flux estimation \hat{F}_X :

$$S_{ph,X} = 1 / \text{SNR}(\hat{F}_X), \tag{9}$$

given the mean interferometric intensities shown in Fig. 3, it is consistent to assume a photon-noise dominated regime. Under this condition, $\text{SNR}(\hat{F}_X)$ is derived from [23]:

$$\text{SNR}(\hat{F}_X) \approx V_X \sqrt{\bar{N}_{ph}}. \tag{10}$$

The ratio $S_{ph,X} / S_{\partial[X]}$ gives a good appreciation of the NanoCarb radiometric sensitivity for the given concentration variation $\partial[X]$. Indeed, $(S_{ph,X} / S_{\partial[X]}) \partial[X]$ is directly a noise over the concentration estimate in the same unit as $\partial[X]$. For example, a ratio of 0.5 for $\partial[\text{CO}_2] = 1$ ppm means a noise over L1b data product equivalent to 0.5 ppm.

Considering (8) and (10), we obtain:

$$S_{ph,X} / S_{\partial[X]} = \frac{1}{\sqrt{\bar{N}_{ph}} \frac{\partial V_X}{\partial[X]} + \frac{V_X}{\sqrt{\bar{N}_{ph}}} \frac{\partial \bar{N}_{ph}}{\partial[X]}}, \tag{11}$$

we can observe two terms at the denominator of the expression above:

- The left term shows as expected a statistical error improvement with the total number of photoelectrons as a function of the photon noise. Moreover, this term highlights the evolution of the NanoCarb sensitivity as a function of the contrast variation over the targeted interferometric regions. It requires a maximization of the interferometric Jacobian contrast, e.g., in the 5.6 mm OPD region for the CO₂ (Fig. 3). Only the spectral filtering performed by the interferential filter permits such an optimization.
- The right term is a second order ($\sim 10^{-3}$ compared to $\sim 10^1$ for the left term in the worst case). Nevertheless, it illustrates a potential downgrading of the sensitivity when an optimization of the Jacobian contrast decreases the total number of electrons in the targeted region. Thus, a joint trade-off must be carefully achieved to not degrade the fringe visibility when optimizing the spectral bandwidth.

On the maximum sensitivity interferometric area of the CO₂ band, the expected $\partial V_{XCO_2} / (\partial[\text{CO}_2] = 1 \text{ ppm})$ is ranged around 10^{-4} , calling for $\sim 10^8$ electron per iFoV for a target random error of 1 ppm in a snapshot acquisition.

3.3.1 Assumption over intensity level

\bar{N}_{ph} in expression Eq. (11) can state for the number of photoelectron contributing to the signal at L1a (snapshot frame) or L1b (final co-registered interferogram).

- **On L1a snapshot partial interferogram:** $\bar{N}_{ph} = \bar{N}_{snap} n_{FP}$ is the mean total level of photoelectron contributing to the signal for a given iFoV in a single snapshot frame, with n_{FP} the number of Fabry–Perot for the considered spectral band.
- **On L1b concatenated partial interferogram:** As explained previously, the L1b data product is the co-registration of snapshot acquisitions while the iFoV shifts across the thumbnail during the orbit. Given n_{exp} the number of co-registered exposures for a given iFoV, we can express the total number of photoelectrons in L1b data products as:

$$\bar{N}_{ph} = \bar{N}_{snap} n_{FP} n_{exp}; [e/fov], \tag{12}$$

Table 2 FPA characteristics used in NanoCarb radiometric study, based on NGP features [24]

FPA format	1024 × 1024
QE	0.9
Sensitivity range	0.5–2.5 μm
Pixel pitch	15 μm
Readout noise	170 e^-
Operating temperature	170 K
Saturation level	590 ke^-

n_{exp} depends on both the number of pixels per thumbnail side N_S and on acquisition frame rate. To permit signal co-registration, the minimum frame rate must be set to achieve an orbital brooming of one pixel between two exposures. Thus, the maximum number of co-registered exposures for one single iFoV is:

$$\max\{n_{\text{exp}}\} = N_S \times \text{broom}_{\text{pix}}, \quad (13)$$

where $\text{broom}_{\text{pix}}$ is the orbital brooming expressed in pixel.

4 Optimization of the spectral bands

This section aims to illustrate the design strategy of the NanoCarb spectral bands, for SNR purposes. We expect to demonstrate the radiometric performances of NanoCarb and its potential for the SCARBO mission, expressed in terms of

statistical error over the total column of CO_2 or CH_4 measurement, with our forward model.

4.1 Investigated NanoCarb design

We based this preliminary design over the state-of-the-art MCT SWIR FPA 1 k × 1 k NGP [24], which has already been chosen for the MicroCarb mission [25] for instance. Relevant FPA characteristics are presented in Table 2. We nominally allocate one FPA per spectral band (1024 by 1024 pixels), as illustrated in Fig. 6.

We target a pixel sampling at ground of 1 km, enabling a ground resolution of $2 \times 2 \text{ km}^2$ with a Nyquist PSF sampling. A choice of 128×128 pixels per thumbnail provides a good trade-off between swath and number of interferometric samples, for a $1920 \mu\text{m}$ FP and microlens pitch. Hence, we achieve a 128 km swath for an 8 by 8 Fabry–Perot interferometer array. The targeted effective acquisition frame rate is chosen for one pixel brooming per frame. We derive the maximum enabled exposure time from this frame rate. Table 3 summarizes these elements of design we consider for the radiometric study.

4.2 Parametric optimization of the SNR

With the basic design above, we present here an optimization of the NanoCarb CO_2 and CH_4 spectral bands, using synthetic interferograms, coupled with our radiometric performance model, and a model of 4-cavities narrowband filter. The optical transmission of the device is in range about 80%. We assume a linear evolution of the absorbed power

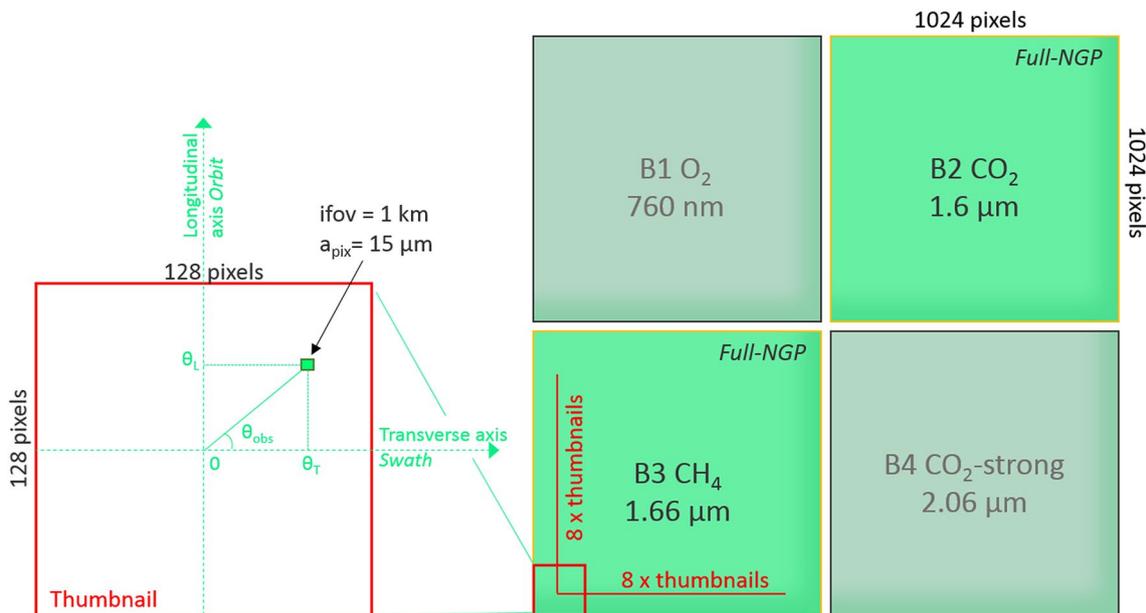
**Fig. 6** NanoCarb allocated spectral bands and nominal investigated configuration of spatial sampling

Table 3 Nominal NanoCarb band configuration for performance estimation

Altitude	600 km
Frame rate	1-pixel brooming
Dedicated FPA size	1024 × 1024
Pixel per thumbnail side N_S	128
Number of Fabry-Perot and thumbnails n_{FP}	64
Pixel sampling at ground (iFoV)	1 km
Swath (fov)	128 km
Maximum exposure time	144.46 ms

with the concentration. This last point is well suited on the unsaturated 1.6 μm and 1.66 μm bands we consider here.

We vary the filter central wavenumber σ_0 over the spectral regions of interest, as well as its full width at half maximum (FWHM) from 1 to 100 cm^{-1} . For each couple $\{\sigma_0; \text{FWHM}\}$, we compute the single-iFoV snapshot radiometric sensitivity from a differential interferometric envelope $\partial V/\partial X$, with ∂X a finite variation of concentration for the specie X . On this envelope, we search where the fringe visibility is maximum in the range of OPD corresponding to the Fourier signature of absorption line mean periodicity over the related spectral band.

Figure 7 shows the obtained map of sensitivity, expressed in ppm^{-1} and ppb^{-1} per iFoV and per frame, respectively, for CO_2 and CH_4 .

- The red squares in the maps spot the optimal filter in terms of SNR.
- The contours highlight regions as a percentage of the optimal sensitivity.
- The partial derivatives in radiance for $\partial[\text{CO}_2] = 1 \text{ ppm}$ and $\partial[\text{CH}_4] = 10 \text{ ppb}$ are superimposed to the σ_0 scale, illustrating the evolution of the sensitivity with the periodic spectroscopic pattern.
- Some regions in the maps are aberrant, and set to zero. On the CO_2 map: some parts of the region around $\sigma_0 = 6280 \text{ cm}^{-1}$ enclosed by the 25% contour; on the CH_4 map: the region around $\sigma_0 = 6000 \text{ cm}^{-1}$. On these regions, the algorithm fails to find a signature on the targeted interferometric range. In the case of CO_2 , there are no CO_2 absorption lines, and the interferogram is dominated by residual continuum shape. Concerning CH_4 , the band is dominated by saturated line group around 6000 cm^{-1} .

CO_2 : the line periodicity is small ($\sim 2 \text{ cm}^{-1}$) and presents some quite important jitters along the σ_0 -axis. As a consequence, the maximal sensitivity evolves rapidly with a shift of spectral band, which is responsible for unstabilities of the relative $\text{CO}_2/\text{H}_2\text{O}$ sensitivity in the FoV for the optimal red point $\{\sigma_0 = 6216 \text{ cm}^{-1}; \text{FWHM} = 17 \text{ cm}^{-1}\}$. The symmetric

optimal filter $\{\sigma_0 = 6336.3 \text{ cm}^{-1}; \text{FWHM} = 17 \text{ cm}^{-1}\}$ is also rejected due to a huge amount of water on this region. The finally chosen filter $\{\sigma_0 = 6213 \text{ cm}^{-1}; \text{FWHM} = 24 \text{ cm}^{-1}\}$ is a trade-off between $\text{CO}_2/\text{H}_2\text{O}$ sensitivity, SNR, and stability in the FoV.

CH_4 : the periodicity of the CH_4 absorption line is large ($\sim 12 \text{ cm}^{-1}$). Consequently, the optimal filter $\{\sigma_0 = 6078 \text{ cm}^{-1}; \text{FWHM} = 69 \text{ cm}^{-1}\}$ is wide, and the sensitivity slowly revolves around this optimal point, which is convenient to design the narrowband filter. Hence, despite of the large amount of water on this band, some interesting interferometric water-free regions, stable in the FoV, can be found as we will see in next section. Thus, the red spotted filter parameters are chosen for B3.

Table 4 summarizes the selected B2 and B3 bands.

4.3 Extrapolation of the noise over CO_2 and CH_4 measurements on L1b data product

The last sub-section illustrated the strategy to design the spectral bands of NanoCarb. Given the two retained B2 and B3 bands, we aim at extrapolating here the intrinsic radiometric sensitivity of the instrument at L1b level.

We plot in Fig. 8 the statistical error on CO_2 or CH_4 concentration measurement as expressed in Eq. (11) as a function of the percentage of maximum co-registered snapshot acquisitions $\max\{n_{\text{exp}}\}$. We assume here that all the interferometric samples of NanoCarb are located on the maximum CO_2 or CH_4 sensitivity area of each band derived from the previous study. We investigate four observation scenarios: 1 and 1.5 km of ground pixel sampling (ground resolution respectively 2×2 and $3 \times 3 \text{ km}^2$), and a terrestrial albedo of 0.05 and 0.2 with a solar zenithal angle of 55° within clear sky.

4.3.1 Statistical error on CO_2 measurement

The exploitation of all the available snapshot acquisitions largely fulfills sub-ppm noise on CO_2 measurement, in all the considered configurations, and observation conditions. The performances are even below 0.5 ppm in the more favorable observation conditions (albedo > 0.2). Hence, the use of only 50% of the available snapshot acquisitions allows again to reach a sub-ppm noise. This is a benefit to mitigate interpolation issues between the different frames caused, for example, by platform dis-pointing or jitter, as well as cloud effects.

4.3.2 Statistical error on CH_4 measurement

The 10-ppb sensitivity target is reached too in all observation conditions, with the co-registration of at least 50% of

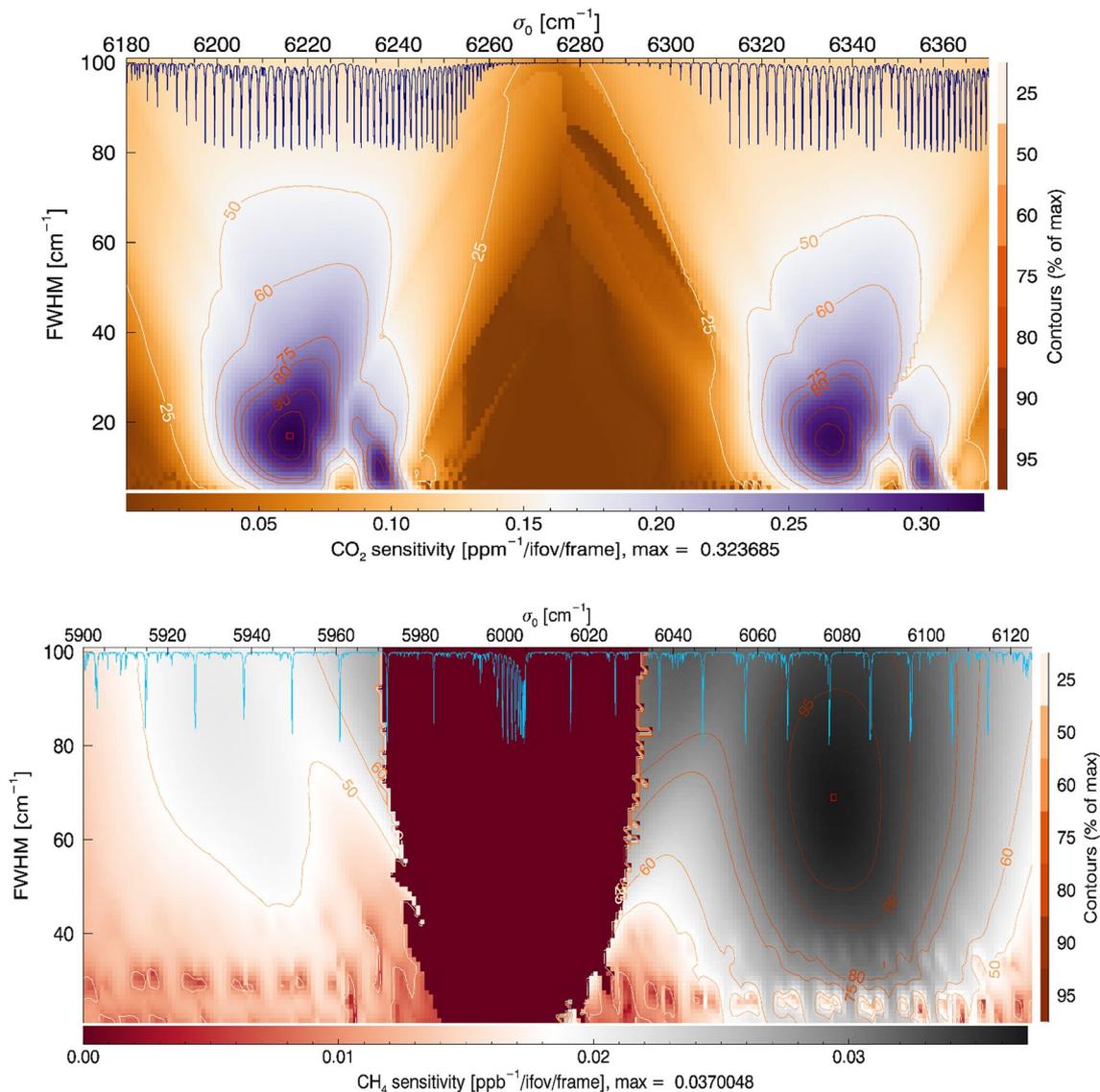


Fig. 7 Sensitivity evolutions for CO₂ (top) and CH₄ (bottom) at maximum OPD as a function of the interferential filter parameters: central wavenumber σ_0 (x-axis) and FWHM (y-axis). To the top of each map is superimposed the differential spectrum, respectively, for a finite

variation of 1 ppm of CO₂ and 10 ppb of CH₄. The contours highlight evolutions of the sensitivity as a percentage of the optimal filter (red square)

Table 4 Selected filter for B2 and B3 bands and relevant snapshot sensitivity, given a solar zenithal angle of 55° and an albedo of 0.2

Band	Filter		Max. snapshot sensitivity/iFoV
	σ_0 (cm ⁻¹)	FWHM (cm ⁻¹)	
B2—CO ₂ —1.60 μ m	6213	24	3.10 ppm
B3—CH ₄ —1.66 μ m	6079	69	27.02 ppb

the available frames. We can expect a noise around 5 ppb in the worst observation conditions (albedo=0.05), by co-registering all the available frames.

In conclusion, within optimal instrumental conditions (no instrumental defects, photon noise only, perfect knowledge of the other geophysical parameters):

- 0.2–1 ppm target for CO₂ noise is reached with a nominal allocation of one NGP for the 1.6 μ m band.
- 1–5 ppb for CH₄ noise is reached in the same conditions.
- These good intrinsic performances enable (1) a decrease of the exposure time up to a factor 4 to mitigate pointing blurring effects, (2) the allocation of samples to other dedicated areas of the interferogram, for example, to jointly measure water.

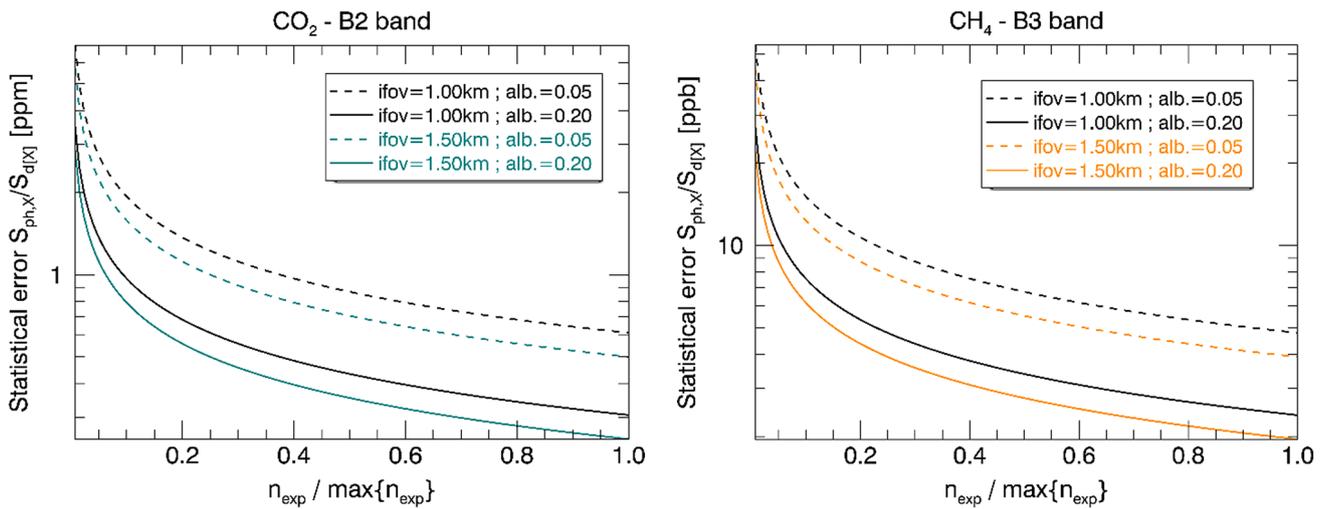


Fig. 8 Evolution of the NanoCarb statistical error on CO₂ and CH₄ measurements as a function of the number of co-registered exposures. On *x*-axis, one means the use of the total number of available frames when the iFoV shifts across the thumbnail, while zero is the snapshot mode

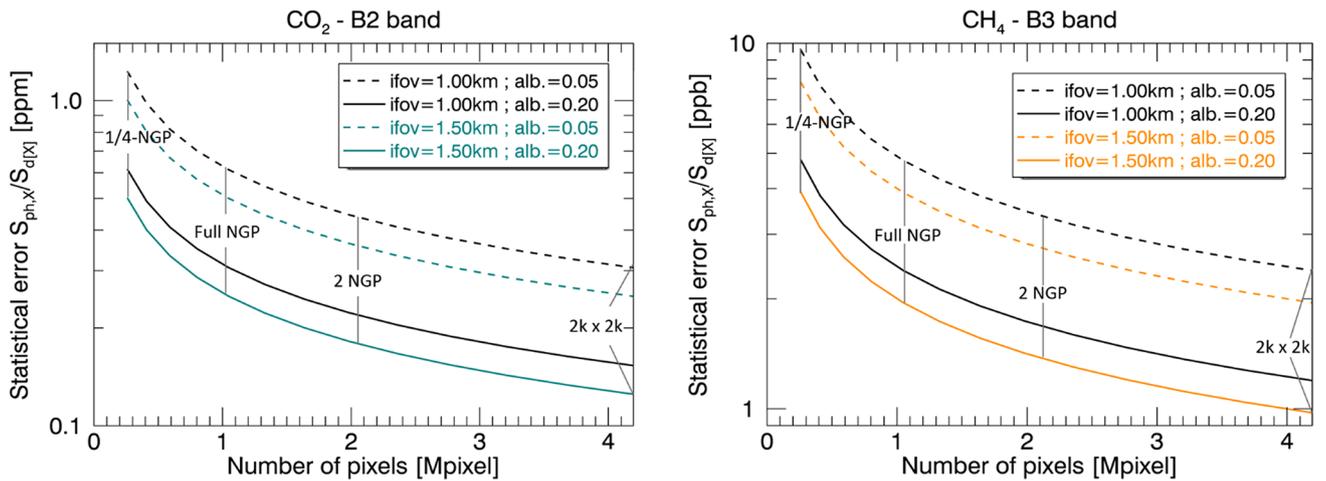


Fig. 9 Evolution of the noise on L1b as a function of the total number of pixels over the FPA. The statistical error is given for a co-registration of all the available frames $\max\{n_{exp}\}$. The number of pixels per thumbnail and the pixel pitch are unchanged. Thus, the number of

co-registered exposures is kept constant, as well as the exposure time. The *x*-axis is equivalent to an increase in the number of thumbnails, and thus, to the number of interferometric samples

To go further, an interesting feature of the NanoCarb concept is the dependence of the noise on L1b data to the number of pixels used on the considered spectral band: the number of thumbnails (or FP), n_{FP} , can be easily increased by enlarging the FPA area dedicated to the spectral band, any other parameters fixed. Consequently, \bar{N}_{ph} will increase linearly with the size of the dedicated FPA area. Figure 9 shows the noise trends as a function of the number of dedicated pixels, in the case we co-register all the available frames.

This optimization of the NanoCarb radiometric performances is done with a very faint evolution of the instrumental complexity. As the number of pixels increases for

the totality of the spectrometer, we are confident to a linear increase of the volume of the optical part of NanoCarb (from the objective to the FPA), driven by the size of the sensitive area on the detector. As an example, if we evaluate a volume of $\sim 250 \text{ cm}^3$ with one NGP, the volume with a next generation 2 k 2 k NGP-like [26] will be approximately increased by a factor of eight. Weight and full thermal regulation must be treated as such, which is a real advantage compared to dispersive spectrometry techniques and scanning Fourier-transform spectrometers.

We demonstrated here the intrinsic radiometric capabilities of the NanoCarb concept for CO₂ and CH₄, in a

Table 5 Geophysical variables to be retrieved from NanoCarb measurement, and impact over the different bands. In green—the dedicated spectral bands when occurred

Geophysical variable	Sensitivity			
	B1	B2	B3	B4
CO ₂ profile	×	Yes	×	Yes
CH ₄ profile	×	×	Yes	×
H ₂ O profile	×	Yes	Yes	Yes
Surface pressure	Yes	Yes	Yes	Yes
Albedo	Yes	Yes	Yes	Yes
Clouds and aerosols optical depth	Yes	Yes	Yes	Yes

forward approach. Nevertheless, this study does not study any impact of the instrumental or geophysical biases, and does not allow consequently to accurately choose the interferometric samples.

5 Design of the Fabry–Perot micro-interferometer array

Previous section focused on band optimization for SNR purposes. We develop and illustrate here a proper strategy to mitigate bias impacts over CO₂ or CH₄ measurement, by designing the micro-FP interferometer array and choosing the OPDs to sample. We mainly present water-related issues as justified in a first sub-section. A second sub-section presents a forward exploration of the information content of the interferogram, as the last one introduces an inverse approach to estimate the performances and generalize this optimization.

5.1 Biases over B2 and B3 bands and geophysical variables

The main geophysical variables to be retrieved, which affect the NanoCarb measurement over the previously considered spectral bands, are summarized in Table 5.

These geophysical parameters potentially bias the measurement of averaged CO₂ and CH₄ dry column, and must be addressed in the design of the micro-interferometer array. Nevertheless, at this preliminary level of design, we focus in this paper only on B2 and B3 bands:

- Clouds and aerosols issues, monitored on B1 and B4 bands, are consequently not considered despite their huge impact over CO₂ and CH₄ measurement. We assume a perfect joint measurement by SPEX, similar to clear sky assumption. An upcoming study will be led to assess the un-bias capability of the SPEX instrument over Nano-

Carb measurement, with a realistic model of performances.

- We assume a perfect joint estimation of surface pressure from B1 band.
- The albedo estimation accuracy relies on the radiometric calibration capability of the device, and has not been yet studied. Works based on experimental characterization are in progress to evaluate FPA mis-calibration impact over our signal knowledge. A discussion about it is proposed in paragraph 5.4.
- Water is one of the main interferants affecting B2 and B3 bands, and is consequently treated in this preliminary design of micro-interferometer array.

We present below a simple study of the impact of a mis-estimation of the total column of H₂O over the averaged column of CO₂ retrieval. The goal is to derive an estimation of the required accuracy we have to reach, for illustrative purposes. The retrieval algorithm is a simple least mean square method based on a Levenberg–Marquardt algorithm. Its specificity is to work with partial concatenated L1b interferograms, without intermediate radiance spectrum retrieval.

We target the B2 5.6 mm area of the interferogram where the sensitivity to CO₂ is optimal. All the 64 FP are dedicated to the sampling of this region. Apart from the bias on H₂O, the retrieval L1b model is perfect, and the simulated data are noiseless. The CO₂ initial guess for the retrieval is about 95% of the data value (400 ppm). Figure 10 presents the evolution of the absolute accuracy over CO₂ estimate with respect to the bias on water. We notice a strong impact of a water misestimation, because 10% error on water induces a bias greater than 1 ppm over the CO₂, while 1% leads to a bias greater than 0.1 ppm.

This observation was well expected if we consider the CO₂ interferogram as in Fig. 3—left: the partial derivative for 1 ppm of CO₂ is comparable in flux to the one for 10% of water in the 5.6 mm region. Therefore, both H₂O and CO₂ have a significant and comparable impact over the fringe intensity. With similar considerations, the impact of water over CH₄ band is greater by a factor 10 at least.

This result justifies the need to refine the NanoCarb design to mitigate the water impact on B2 and B3 bands. Especially, we will consider in the next sub-section the selection of several dedicated interferometric regions in a forward approach.

5.2 Design for water mitigation with forward model

The forward model of NanoCarb is implemented here to search for interferometric regions in B2 and B3 bands where sensitivity to CO₂ and CH₄ is optimal related to water and instrumental issues, especially the spectral band

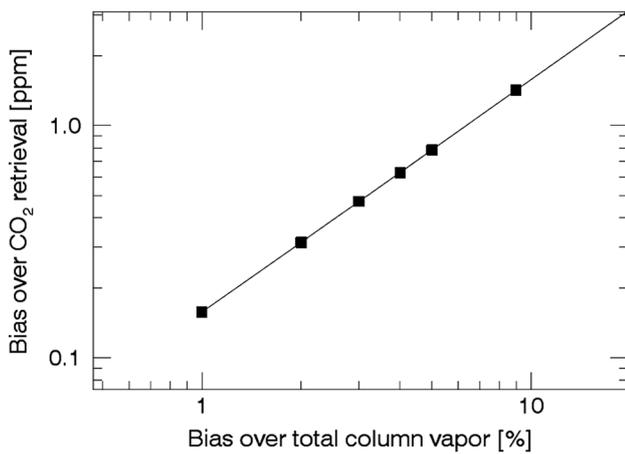


Fig. 10 Evolution of the bias over total column of CO₂ retrieval as a function of a misestimation of the total column of H₂O. The maximum CO₂ sensitivity region of the interferogram for the chosen filter is targeted

shift in the FoV of the filter, responsible for interferometric signature variabilities and instabilities.

Figure 11 shows central wavenumber shift in the FoV of a thumbnail. For the investigated design, the spectral shift is quite similar in B2 and B3 bands, with a maximum of +9.5 cm⁻¹ on the corners, and +5 cm⁻¹ at maximum on axis.

Full interferograms are computed over a relevant range of OPD. Then the envelope variations $\partial V/\partial X$ are derived, with X , respectively: a 1 ppm CO₂ finite variation, a 1 ppb CH₄ variation, and a 10% H₂O variation. These envelopes are plotted on Figs. 12 and 13, for each pixel of a half thumbnail along the longitudinal axis (direction of signal

co-registration) and for four iFoV on the swath (transverse axis) as represented by multi-colored arrows in Fig. 11.

On Figs. 12 and 13, the superimposed blue curves show the evolution of the water interferometric signature along the longitudinal axis, while the brown–white–green curves state for CO₂ on B2 on Fig. 12 and CH₄ on B3 on Fig. 13. Each graph (a–d) on both Figs. 12 and 13 represents one among the four selected iFoV on the swath.

CO₂ on B2 Fig. 12: The maximum of sensitivity for CO₂ at 5.6 mm seems stable in the FoV even for the most extreme iFoV of the swath. Nevertheless, this interferometric region is also sensitive to water in quite important proportions (typically 50% of CO₂ visibility), calling for dedicated H₂O disjoint region. The 1.2 mm water region is rejected due to huge sensitivity variations in the FoV. On the contrary, the 2.4 mm water-dominated region is stable in the FoV while the sensitivity to CO₂ is almost null. As a conclusion, the CO₂ water-biased 5.6 mm region and the water CO₂-free 2.4 mm region are selected to guarantee an unbiased measurement of CO₂ concentration.

CO₂ on B2 has a harmonic signature at 11.2 mm of OPD that we did not consider here. Indeed, such a high OPD induces high frequency FP ring pattern and consequently potential pixel sampling issue. This consideration is related to the interferometric transposition of the Jacquinot criterion about spectral resolution and sampling.

CH₄ on B3 Fig. 13: The CH₄ signature presents several harmonic regions every 1.05 mm of OPD, and thus as many sensitive areas. Nevertheless, only the first two have fringes in a favorable proportion of sensitivity CH₄/H₂O, and we limit the study to the 2.5 mm-wide plotted regions. CH₄ signature is very stable in the FoV, as well as several

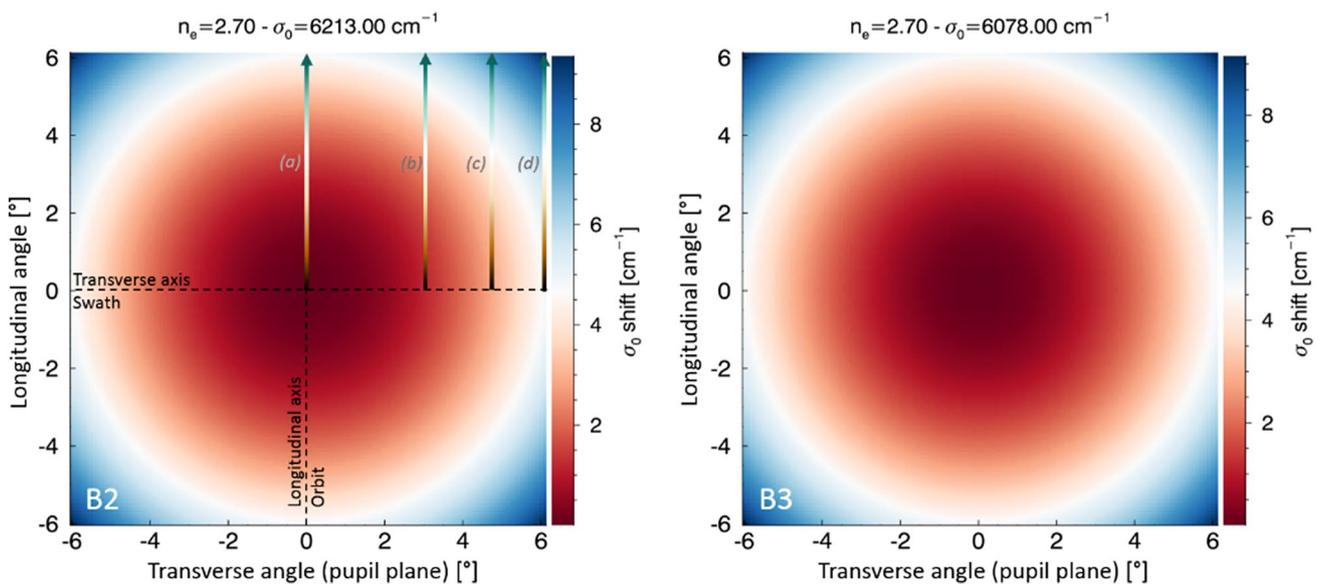


Fig. 11 Filter central wavenumber shift in the FoV of a thumbnail for B2 (left) and B3 (right). n_e is the effective index of the interferential filter

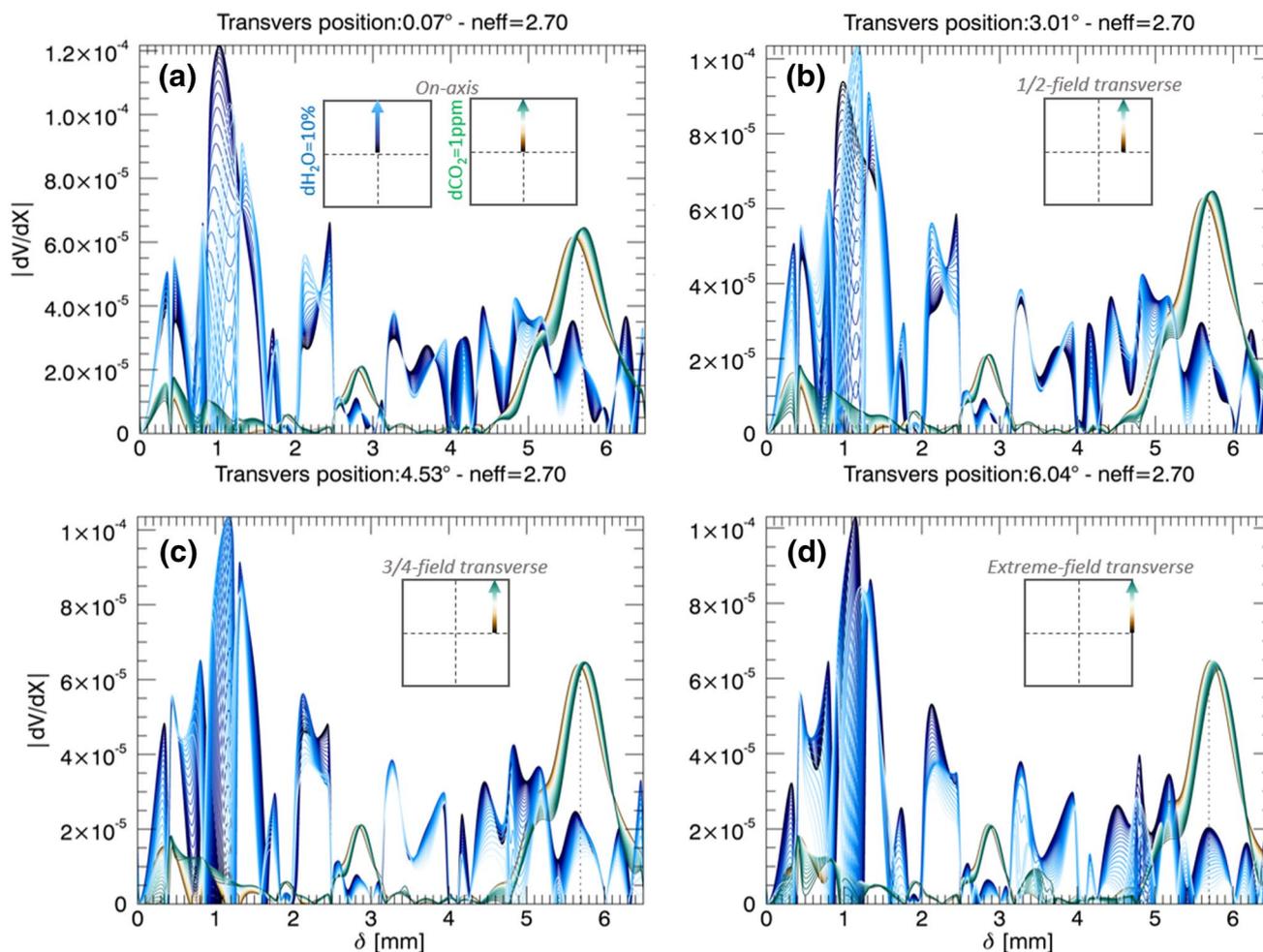


Fig. 12 Evolution of the CO₂ interferometric signature—envelope—along the longitudinal axis (brown–white–green superimposed curves), and of the water (blue superimposed curves). **a** iFoV at the

center of the swath; **b** iFoV at the half-swath; **c** iFoV at the $\frac{3}{4}$ of the total swath; **d** extreme iFoV on the swath

water areas such as at 0.9 mm, 1.25 mm, and 2.35 mm. The maximum sensitivity of CH₄ at 1.05 mm seems to be free of water. Additional samples placed on the second CH₄ harmonic could provide also a measurement of a more varied kind of information, for instance about vertical profile.

5.3 Illustration of CO₂ and water retrieval with two OPD areas

To illustrate the bias reduction provided by the choice of specific OPD areas in the same spectral band, we present in Fig. 14 a preliminary retrieval result from a L1b B2 measurement vector. This vector is composed by eight water-dedicated samples around 2.4 mm of OPD, and 54 CO₂-dedicated samples around 5.6 mm of OPD. The NanoCarb configuration is the one defined in Table 3, and the filter is the B2 filter defined in Table 4. For this retrieval run, we consider a perfect inverse model, noiseless data,

and a guess concentration of 90% of the real value for the total column of CO₂ and H₂O. The retrieval algorithm is the same as used in sub-Sect. 5.1.

Despite a slow convergence of the water estimation, this very interesting result shows a quick convergence of the CO₂ column retrieval toward a numerically limited fraction of the real value. It seems that the chosen partial sampling of the interferogram permits in this case a robust degeneracy removing of water over the CO₂ retrieval.

The next sub-section introduces some leads to improve this design as well as assess its performances.

5.4 Discussion about design optimization and performances assessment

We illustrated previously the forward strategy to design the NanoCarb μ FP array considering one of the main interferant over B2 and B3 bands, allowing to establish a preliminary

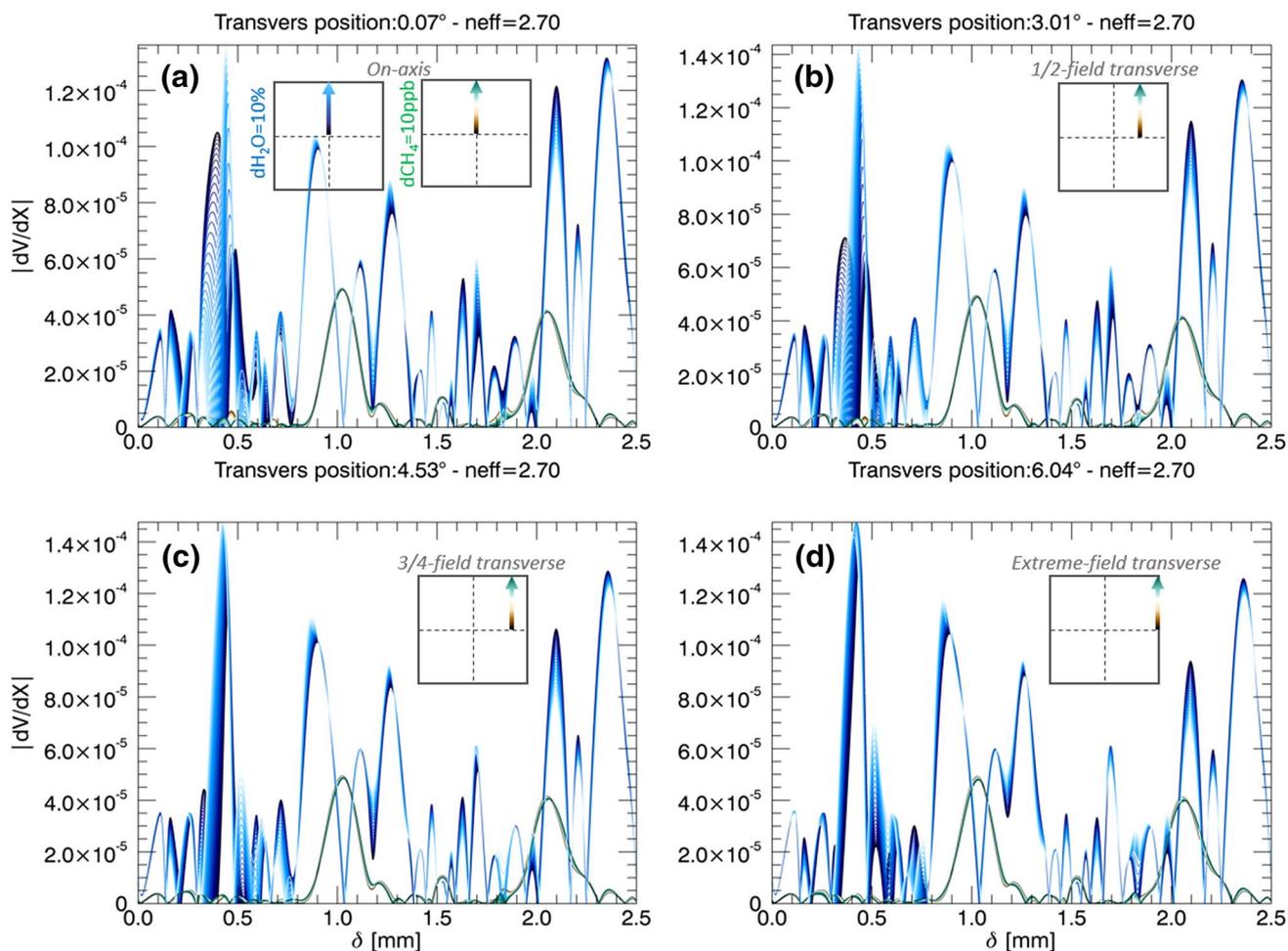


Fig. 13 Evolution of the CH₄ interferometric signature—envelope—along the longitudinal axis (brown–white–green superimposed curves), and of the water (blue superimposed curves). **a** iFoV at the

center of the swath; **b** iFoV at the half-swath; **c** iFoV at the 3/4 of the total swath; **d** extreme iFoV on the swath

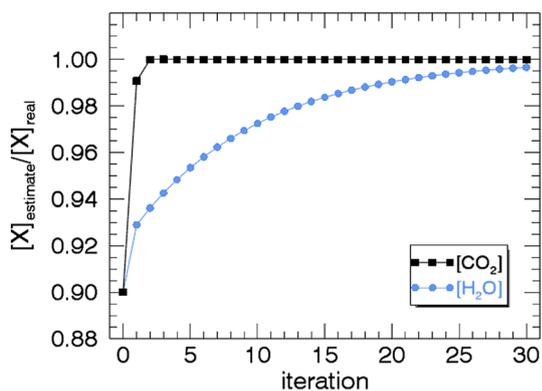


Fig. 14 Retrieval convergence of a joint estimation of CO₂ and H₂O columns on the 1.6 μm band, by sampling two dedicated disjoint regions on the interferogram, one for water and the other one for CO₂

design of the instrument. However, some questions remain about OPD choice, which cannot be assessed with this model:

- Generalization of the approach to all the geophysical variables described in Table 5, as well as instrumental biases.
- Number of allocated FPs (samples) per identified region.
- Sampling strategy: (1) multi-point sampling of the interferometric envelope, calling for multiple disjoint regions, or (2) concentration of all the samples on some regions illustrated in the previous sub-section. While the concentration of samples at maximum sensitivity areas ensures a maximum SNR, the variety of measured information is very poor, potentially responsible for important sensitivity to other geophysical biases or vertical-induced profile effects.

Table 6 Main considered instrumental issues for a NanoCarb-based payload, related impact over the performances, and assumed criticality for the final data quality considering knowing solutions

	Impact	Solution	Criticality
Spectral response calibration	High	OCT scan or spectral scan of each narrowband	Low
Radiometric calibration	High	To be established	High
Detector PRNU	High	Inflight calibration protocol has to be addressed	Low/high
Parasitic/stray light	High	High criticality known for many space instruments with a similar performance aim but a different concept (e.g., dispersive spectrometer).	High
Filter thermal stability	High	~0.1–1 K thermal regulation	Low
Interferometer thermal stability			
Platform stability	Low	Specification to be determined	Low
Component behavior in space	Unknown	Radiation resistance of silicon-based components, multi-layer processed surfaces, detector aging, etc., has to be addressed	Unknown

- Validation of the chosen interferometric samples in terms of reached absolute accuracy on L2 data product (systematic biases).

5.4.1 Generalized bias mitigation strategy for OPD choice

A simple way to take into account all the geophysical variables in the design of NanoCarb interferometers could be based on a retrieval model coupled with a Monte Carlo exploration of instrumental parameter space (position and number of samples, number of regions, etc.). A stray forward would be:

1. Fix the NanoCarb key features, for instance number of pixels, number of Fabry–Perot, FoV, etc.
2. Fix the spectral bands by SNR forward analysis as presented in the previous section.
3. Forward analysis of sensitivity of the interference domain as presented in sub-Sect. 5.2. The aim is to identify consistent interferometric regions and consequently restrain OPD range of analysis, for both all bands and geophysical parameters. A Bayesian approach of the information could be also considered as described in Sect. 2.5 of [11].
4. Multi-variable retrieval on L1b data for n sets of n_{FP} samples chosen by Monte Carlo in the restrained parameter space.

This method potentially enables to derive performance trends as a function of the number of allocated samples per identified region as well as the distribution of regions along the interferometric envelope.

5.4.2 Instrumental issues

Instrumental issues are just starting to be addressed in the SCARBO framework. Table 6 gathers some of them identified for NanoCarb, for discussion purposes.

Both spectral response and radiometric calibration of the instrument are critical to retrieve radiometrically calibrated L1a, and then L1b interferograms from L0 raw images. Even if the spectral response characterization of the instrument does not constitute a stumbling block by using either interferometric techniques such as optical coherence tomography [18, 19] or a spectral scan, efforts have to be provided to ensure a radiometrically calibrated device, with an accuracy to be determined. This latter may have a critical impact to disentangle between albedo variations over the scene and aerosol-induced extinction. Moreover, questions remain about calibration stability in time and in-flight capabilities of re-calibration with the considered 50 kg-class platform.

The Photo-Response Non-Uniformity (PRNU) of the detector and more exactly the residual fixed-pattern noise (RFPN) is related to the radiometric issue and must be treated as relevant, and its impact may be very significant with a Fourier-transform spectrometer [27].

Also related to radiometric calibration, and with a potential huge impact, is the issue of parasitic and stray light. Indeed, it has been observed in another FP-based space instrument such as GHGsat-D “CLAIRE” [28] that stray light may be a critical point for the performances. A snapshot static Fourier-transform spectrometer such as NanoCarb is potentially more sensitive to temporal stray light variation compared with push broom or scanning spectrometers, because each interferometric channel and iFoV are associated with one different pixel of the detector.

On the other hand, as introduced previously in this paper (sub-Sect. 3.1), thermal issue does not seem to be a hurdle to the instrument operability, since a 0.1–1 K thermal stability enables to avoid impacts over the measurement.

Platform stability and component behavior in space will be addressed later, as the efforts are currently focused on the demonstration of the NanoCarb concept feasibility at a quite early level.

Laboratory prototypes are being integrated to address some of these issues, as well as an air-borne prototype is also planned for in-flight demonstration of the NanoCarb performances in a relevant environment.

5.4.3 Systematic performances assessment

The performance assessment of NanoCarb is an iterative work in progress in the H2020 SCARBO framework. A Bayesian optimal estimation of CO₂ and CH₄ averaged dry column is assessed from the presented model of NanoCarb for a complete set of atmospheres and many observation scenarios. The followed methodology has been applied for CO₂ M error parametrization [29]. This study, done with the Laboratoire de Météorologie Dynamique (LMD),² will provide both expected random error of NanoCarb for CO₂ and CH₄ at L2 (to be compared with the noise at L1b presented in this paper), and systematic error. The latter will enable to validate the presented OPD choice and geophysical bias mitigation, or to refine it if relevant.

6 Conclusion

NanoCarb is an original concept of imaging spectrometer combining the use of innovative interferometric components and an unusual partial interferogram sampling technique. These two combined features are investigated in the SCARBO project to assess the feasibility of a constellation based on such miniaturized payloads to monitor the anthropogenic GHG emissions with a daily revisit and a global coverage.

In a forward approach, we showed the radiometric capabilities of this concept to measure CO₂ and CH₄ in the near-infrared, through an optimal design of the narrowband filters. The expected theoretical and intrinsic performances, in terms of statistical error over measurement of the averaged column of CO₂ and CH₄, are comparable to other dedicated space missions. Then we addressed the design of the interferometer, to manage geophysical biases. The design strategy has been illustrated for water mitigation. For instance, a significant theoretical decrease in water-induced bias compared to CO₂ measurement can be expected using an 8 × 8 interferometer array with 8 water-dedicated and 54 CO₂-dedicated Fabry–Perot.

The study must now be completed to assess the performances of the concept in terms of systematic bias, taking into account all relevant geophysical interferants, and then validating or refining the interferometer design. Therefore, operability of the instrument also needs to be addressed,

by taking into account instrumental and platform-induced issues, as well as in-orbit calibration capabilities. An experimental proof of concept is planned in the framework of the SCARBO project.

In conclusion, it is interesting to note that the NanoCarb concept is very flexible at the design level, allowing to target various spectral signatures and therefore several different species in the atmosphere. In each case, the instrument is spectrally and interferometrically optimized for the latter, assuming a good knowledge of what is to be measured. This feature is well suited for trace gasses in the Earth's atmosphere. Thus, NanoCarb concept is more a CO₂- and CH₄-dedicated sensor than a classic spectrometer, which is unusual in earth observation, but a key feature of miniaturization.

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² <http://abct.lmd.polytechnique.fr/>.

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