

# Diffusion coefficient of toluene in methanol

## 3 Diffusion in Liquid Mixtures

### 3.1. Data

#### 3.1.1. Diffusion in Binary Mixtures

C <sub>7</sub> H <sub>8</sub>	(1)	toluene	108-88-3
C H <sub>4</sub> O	(2)	methanol	67-56-1
Mutual Diffusion Coefficient: $D_{12}(x_i)$ ; $T = 301 \pm 0.5$ K; Method: DIA			Ref.: [1991R2]
$x_1$	$p$ [kPa]	$D \cdot 10^9$ [m <sup>2</sup> /s]	
0.0000	101.32	1.374	
0.0955	101.32	1.264	
0.2036	101.32	1.185	
0.3744	101.32	1.170	
0.6180	101.32	1.364	
Intradiffusion Coefficient: $D_{1T}(T)$ ; Method: TAYLOR			Ref.: [1987S14]
$T$ [K]	$\rho_2$ [kg/m <sup>3</sup> ]	$D \cdot 10^9$ [m <sup>2</sup> /s]	
313.2	774	3.05 ± 2%	
373.2	712	6.72 ± 2%	
423.2	651	11.1 ± 2%	
473.4	559	18.3 ± 2%	
Comment: tracer diffusivities along vapor–liquid coexistence curve, given liquid densities of methanol			
Intradiffusion Coefficient: $D_{2T}(T)$ ; Method: DIA			Ref.: [1984E7]
$T$ [K]	$p$ [kPa]	$D \cdot 10^9$ [m <sup>2</sup> /s]	
298.15	101.32	3.99 ± 1%	
Comment: values at trace concentration of <sup>14</sup> C labelled component (2)			

## Symbols and Abbreviations

Short Form	Full Form
$D$	diffusion coefficient
$p$	pressure
$T$	temperature
TAYLOR	Taylor dispersion technique
DIA	diaphragm cell
$x_i$	mole fraction

## References

- [1984E7] Eastal, A. J., Woolf, L. A.: J. Chem. Soc., Faraday Trans. I **80** (1984) 1287–1295.  
 [1987S14] Sun, C. K. J., Chen, S. H.: Ind. Eng. Chem. Res. **26** (1987) 815–819.  
 [1991R2] Ramprasad, G., Das, T. R., Mukherjee, A. K.: J. Chem. Eng. Jpn. **24** (1991) 389–391.